



# Future Proofing Corrosion Control

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# AGENDA

## Future Proofing Corrosion Control

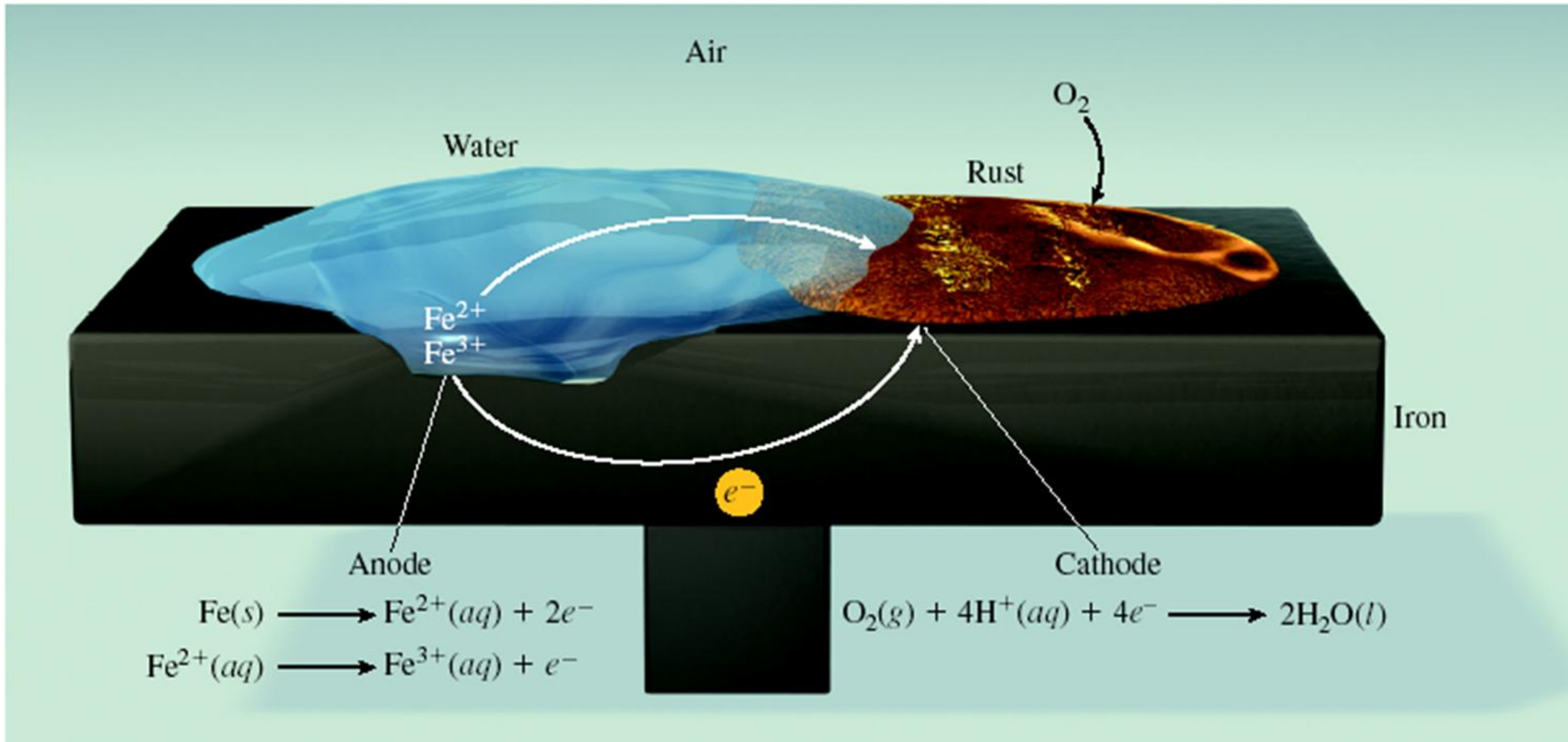
- 1 Corrosion Formation and Inhibition
- 2 Inorganic Corrosion Inhibitors
- 3 Organic Corrosion Inhibitors
- 4 **NEW** HIE Organic CI Application Data
- 5 **NEW** BOP Organic CI Application Data
- 6 Summary



# Corrosion Formation & Inhibition



# Corrosion Cell Diagram

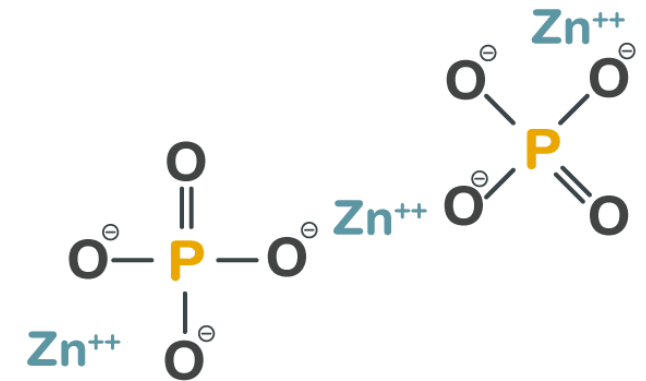
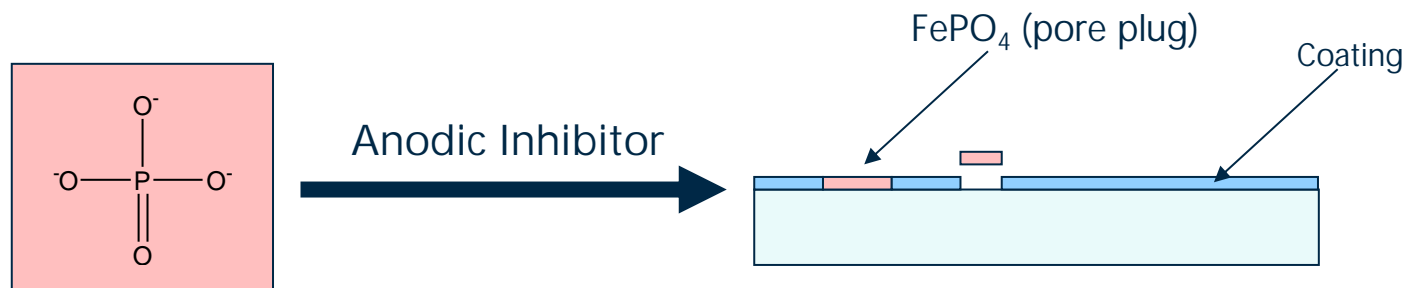


# Traditional Corrosion Inhibition

## Inorganic Metallic Salts

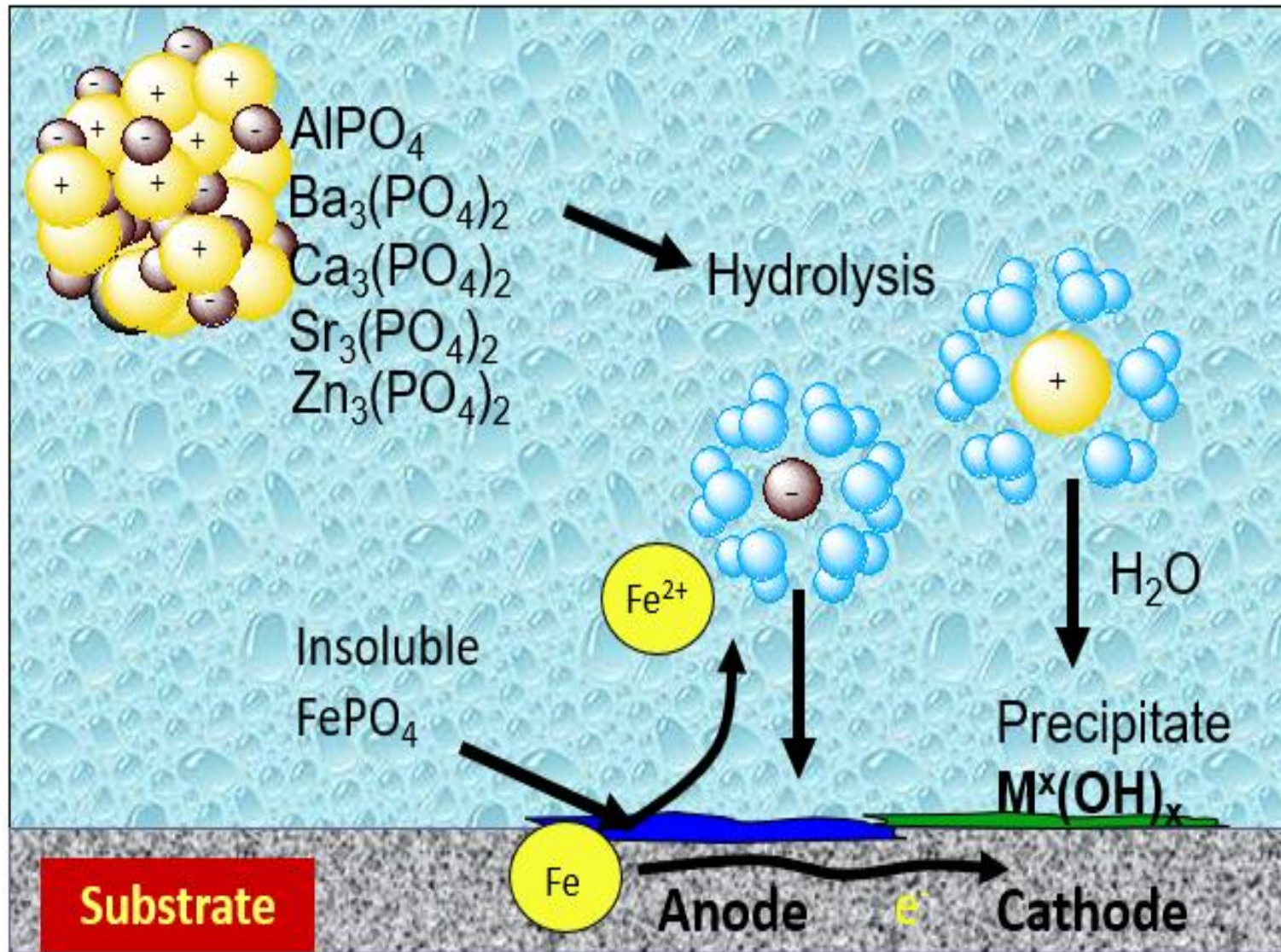
### Mechanisms:

- Anodic Passivation
  - Migrate and react at the anode to form salts that act as a protective barrier.
  - Adsorb at anodic sites, preventing dissolution of metal in acidic media.
- Precipitation
  - Protective film-forming compounds.
  - Block anodic and cathodic reactions by precipitating onto surface of metal





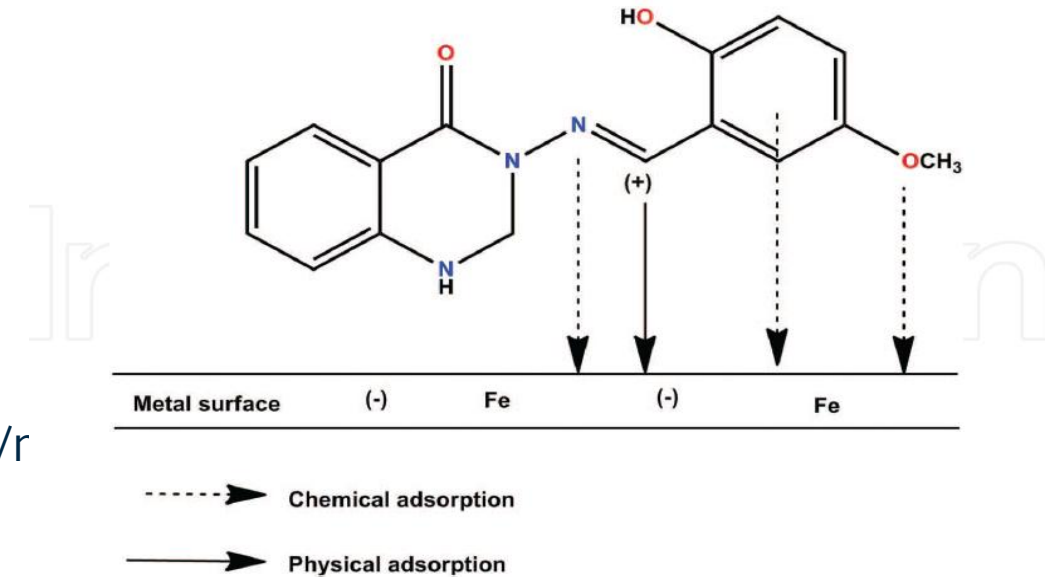
# Mixed Metal Cation Phosphates – Passivation Mechanism



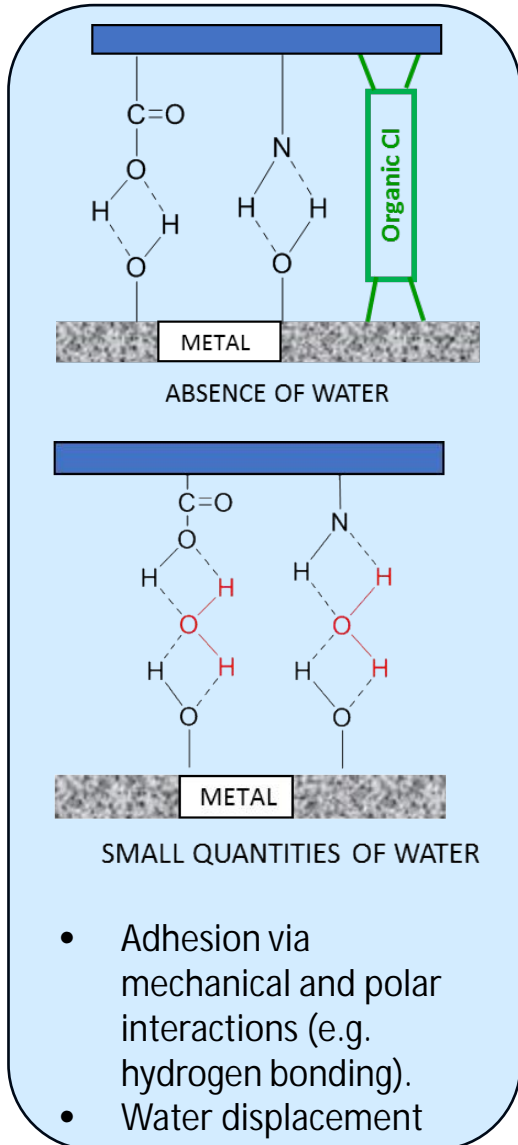
Phosphate	K <sub>sp</sub>
Li <sub>3</sub> PO <sub>4</sub>	2.4x10 <sup>-4</sup>
CaHPO <sub>4</sub>	1.0x10 <sup>-7</sup>
MgNH <sub>4</sub> PO <sub>4</sub>	2.5x10 <sup>-13</sup>
AlPO <sub>4</sub>	1.3x10 <sup>-20</sup>
FePO <sub>4</sub>	1.3x10 <sup>-22</sup>
Mg <sub>3</sub> PO <sub>4</sub>	6.3x10 <sup>-24</sup>
Ba <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	1.3x10 <sup>-29</sup>
Ca <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	2.0x10 <sup>-29</sup>
Sr <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	1.0x10 <sup>-31</sup>
Zn <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	9.0x10 <sup>-33</sup>
CePO <sub>4</sub>	2.9x10 <sup>-34</sup>

# Why Use Organic Corrosion Inhibitors?

- Provide multiple inhibitive mechanisms
- Good solubility
- Low particle size >> minimal gloss impact
- Liquid and powder options
- Low use levels >> maintain cost of use
- Adhesion promotion
- Adsorb on surfaces (physisorption or chemisorption)
  - **Physisorption** (electrostatic interaction b/n charged metal surface and charged inhibitor)
  - **Chemisorption** (transfer or share of unbounded electrons b/r molecule and metal surface)
- Act synergistically with inorganic corrosion inhibitors



# Mechanisms of Organic Corrosion Inhibitors



Interfacial Activity  
Improve coating wet adhesion

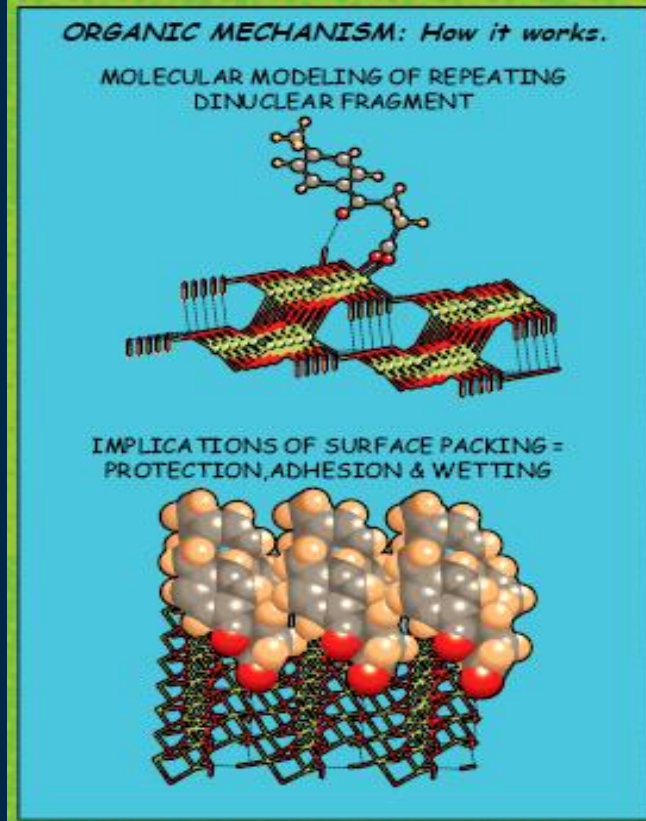
Barrier Properties  
Reduce porosity & permeability in coating

Anodic Activity  
Formation of insoluble complex salts at anodic defect sites

Adsorption Activity  
Protective layer formation

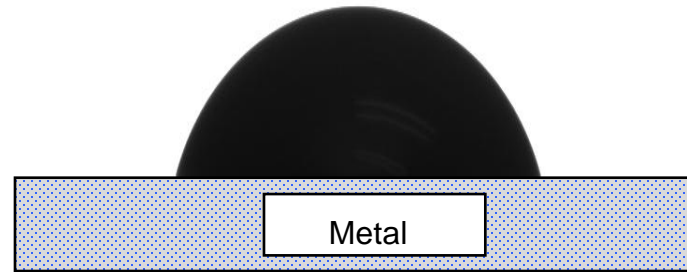


# Mechanisms of Organic Corrosion Inhibitors



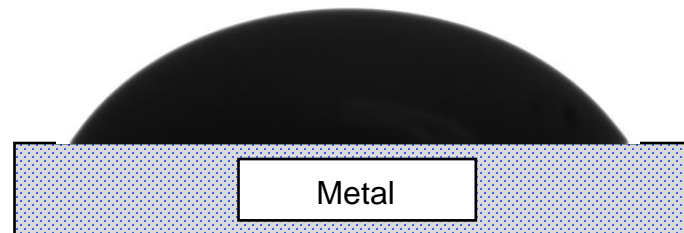
- Adhesion Promotion
- Pore Plugging
- Reduce Water Uptake
- Anodic Passivation
- Flash Rust Prevention

## Improve Substrate Wetting



Acrylic resin solution:  $77^\circ$

Higher contact angle = poorer wetting



Acrylic resin solution +  
3% organic Inhibitor:  $57^\circ$

Lower contact angle = better wetting

# Organic Inhibitor Benefits

Best practices for utilizing organic chemistries

## Performance enhancer

- Synergy with inorganic inhibitors.
- Provides unique mechanisms to enhance performance
- Provides adhesion + passivation.



## Improve product labeling

- Reduce dependency on heavy metals. (chromium, zinc, lead, barium)
- Replace VOC containing CI's.
- Lower use level than inorganics.



## Stand Alone Inhibitor

- Effective at low use levels in WB and SB formulations.
- Minimal impact on gloss.
- Can provide flash rust protection.



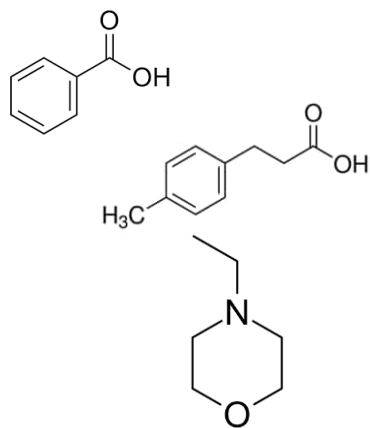
# Organic Compounds as Corrosion Inhibitors

## Structure and Compatibility

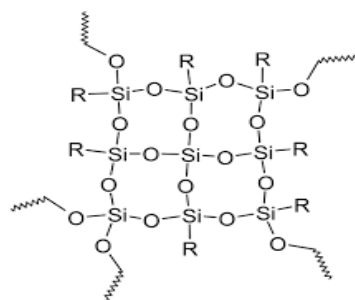
Compatibility



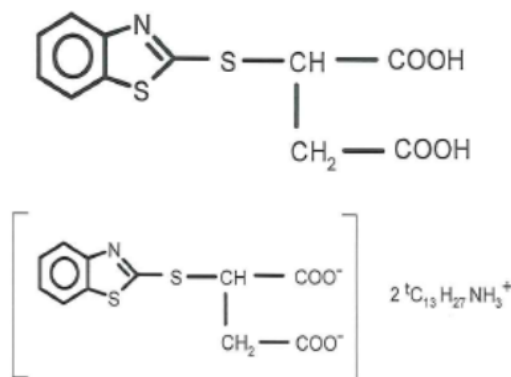
Amine-Acid Salts



Silane-Siloxane derivatives



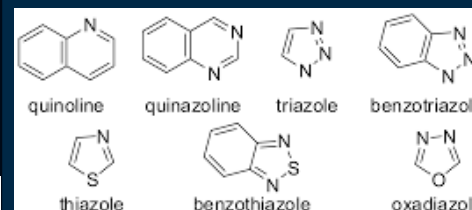
Di-Acids, Acid-Amines



NEW



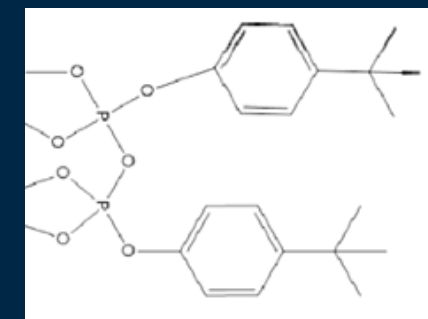
Heterocyclic Ion Exchange (HIE)



NEW



Bio-based Organophosphate (BOP)





# Organic Corrosion Inhibitor Case Studies

HIE - Heterocyclic Ion  
Exchange

BOP – Biobased  
Organophosphate



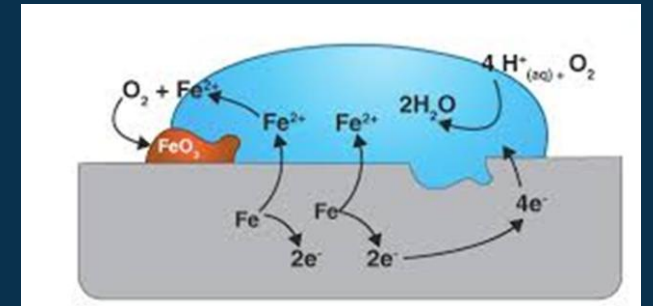
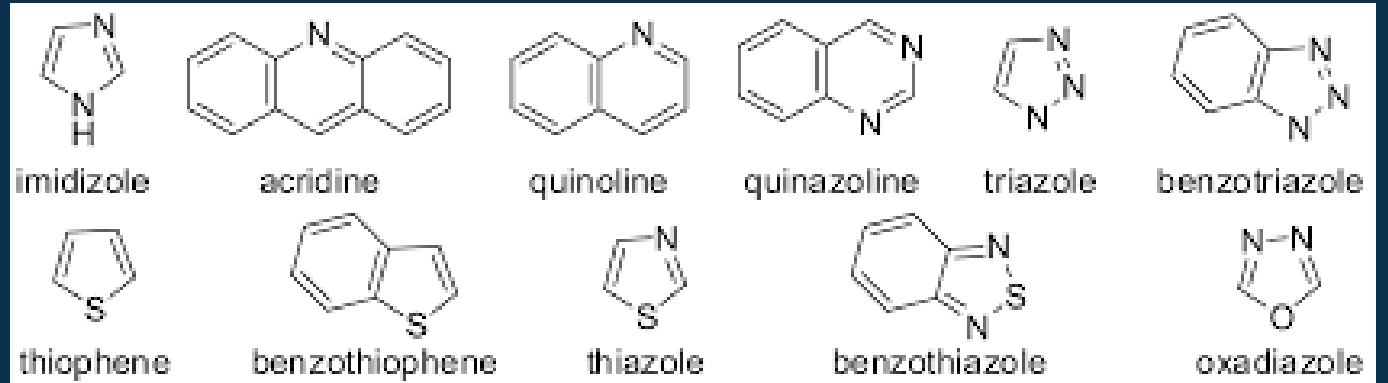


# HIE: Heterocyclic Ion Exchange

- Smart anti-corrosive pigment
- Releases a heterocyclic organic corrosion inhibitor from a micro-reservoir
- Release occurs via ion-exchange
- Released organic combats corrosion

## Key Features

- Fast reaction
- Long lasting within coating
- Multiple protective mechanisms
  - Forms a protective layer blocking anodic activity.
  - Neutralizes pH of corrosion front by donating electrons or receiving  $H^+$  protons.
  - Bind with substrate slowing down cathodic delamination.



# Solvent-based Medium Oil Alkyd Primer

HIE as stand alone corrosion inhibitor

CRS ~ ASTM B-117 ~ 336 hours ~ DFT 75µm

Control

1% Organic  
Comp A

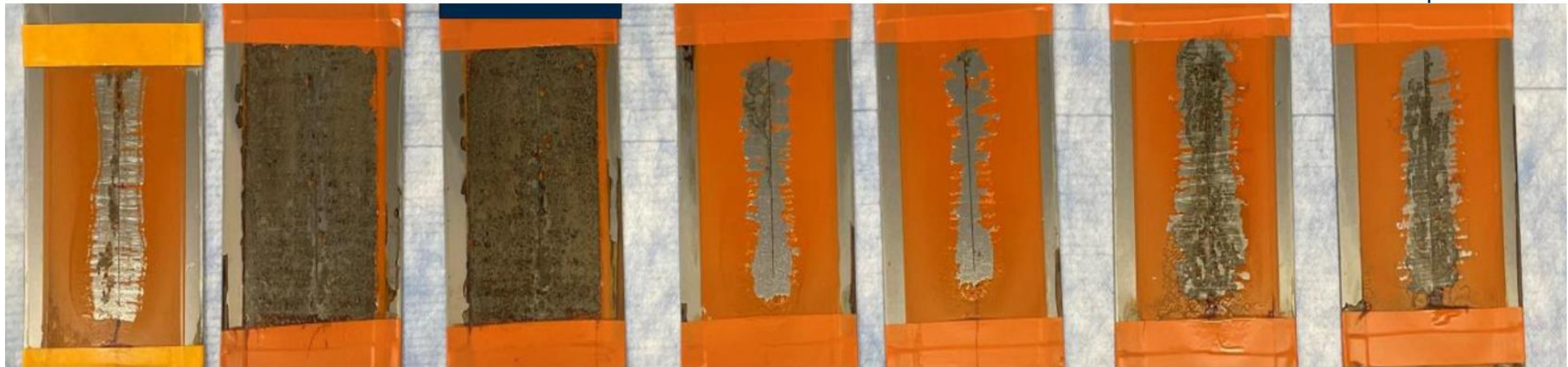
2% Organic  
Comp A

1% HIE

2% HIE

1% Organic  
Comp B

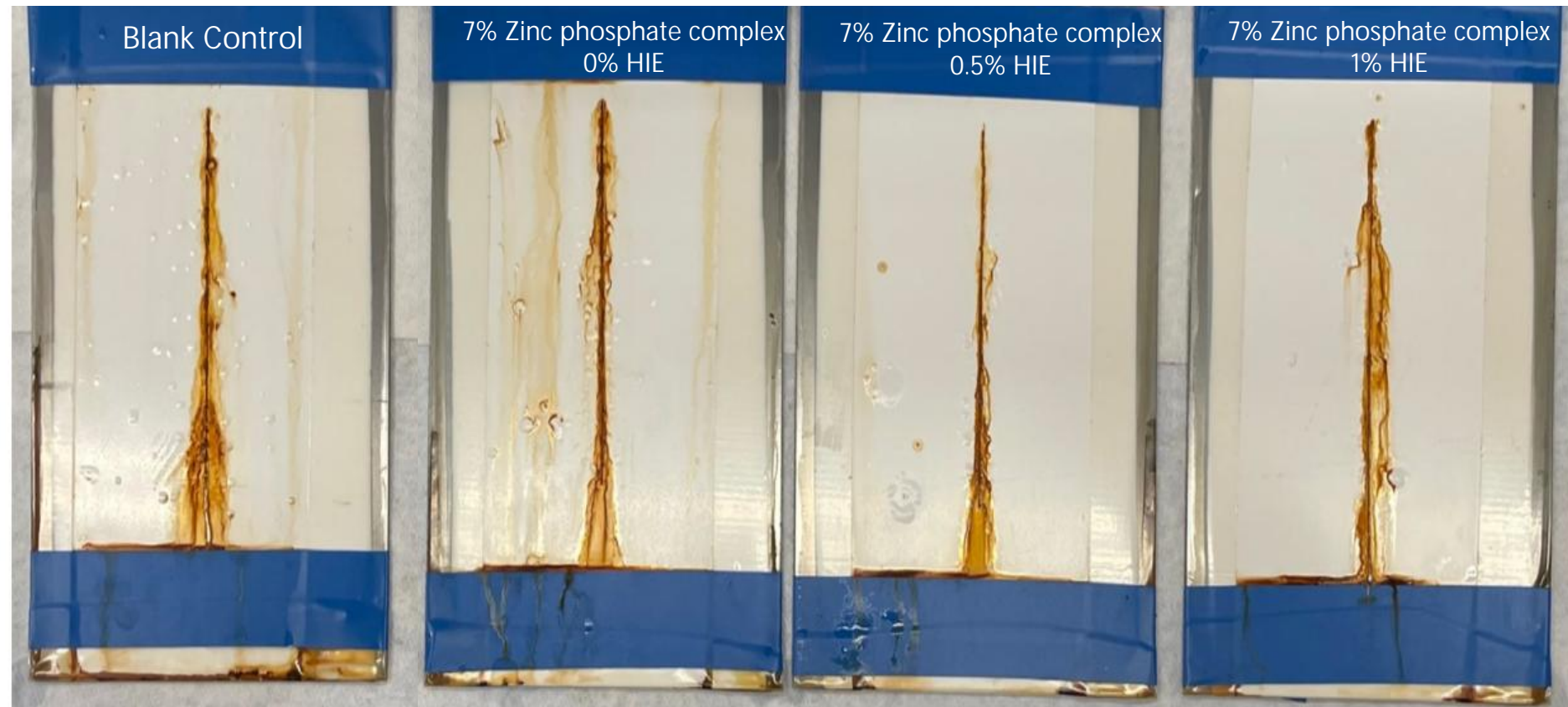
2% Org  
Comp B



# Water-based Acrylic DTM

HIE as performance enhancer

CRS ~ ASTM B-117 ~ 144 hours ~ DFT 70-80 $\mu$ m



# Co-blend for Synergy

HIE optimization of corrosion inhibitor synergy

Powder Performance - 1500 hours SST

5% Zn PHOSPHATE



5% Zn PHOSPHATE &  
1% HIE

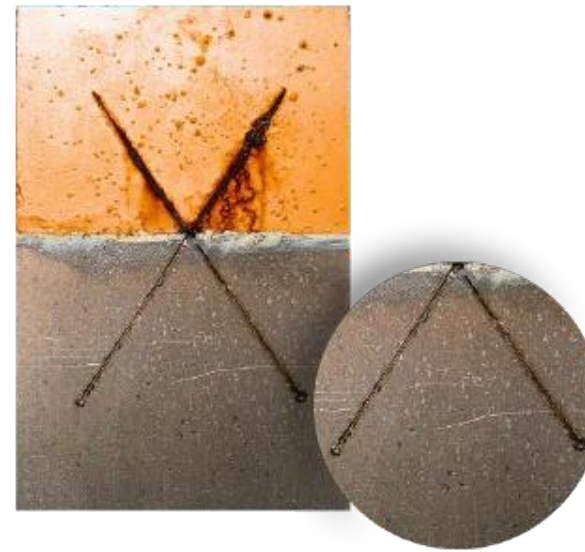


Epoxy-polyester powder performance

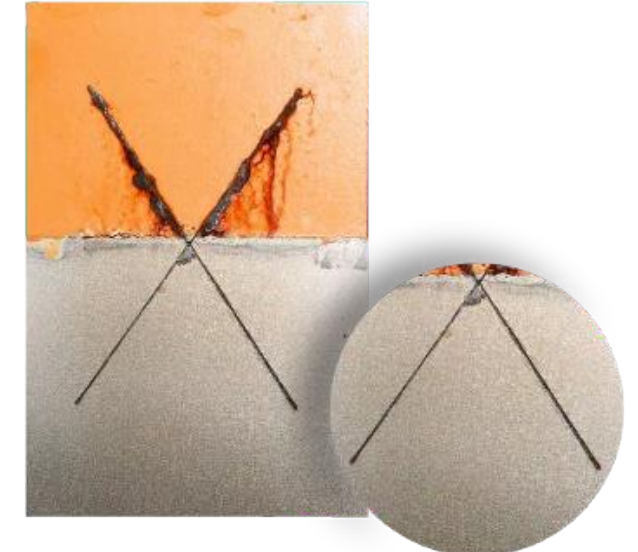
- Reduced corrosion at scribe according to ISO 12944
- Decreased coating delamination
- Substrate shot blasted steel Sa 2.5
- ISO 9227-Neutral Salt Spray DFT: 60-90  $\mu\text{m}$

WB Primer & Topcoat - 1000 hours SST

4% Zn PHOSPHATE



2% Zn PHOSPHATE &  
2% HIE



WB Acrylic with SB PU topcoat

- Reduced corrosion at scribe
- Improved surface corrosion protection
- Substrate Steel
- DFT: 40-50  $\mu\text{m}$  (Primer) / 70-80  $\mu\text{m}$  (Topcoat)

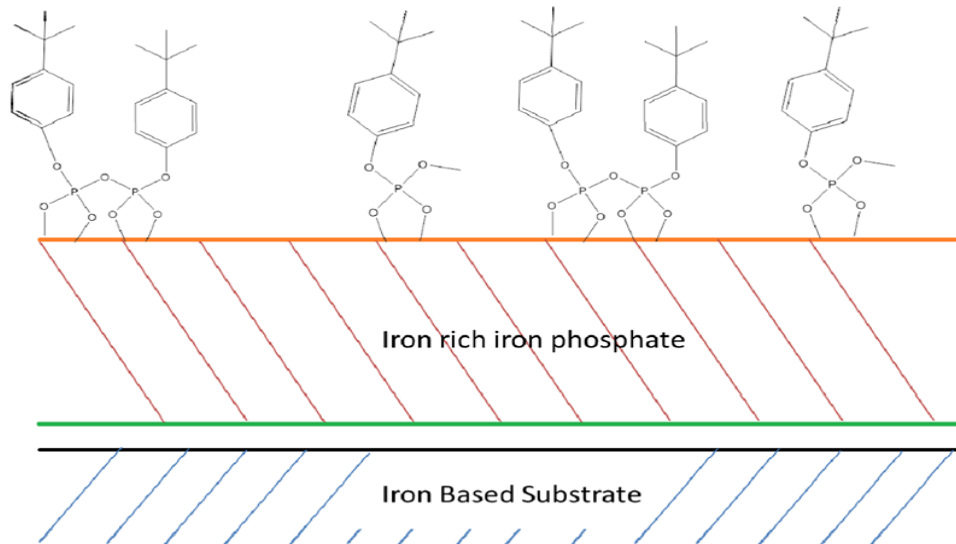


# BOP: Bio-based Organophosphate

- ✓ Unique **liquid** CI with **clean MSDS** (50% biobased)
- ✓ Suitable for both **solvent- & water-based coatings**
- ✓ Long term corrosion protection up to **C4**
- ✓ **Good adhesion** on difficult substrates
- ✓ **Good stability** – easy to use
- ✓ **Ecolabel** compliant: <0.5% VOC

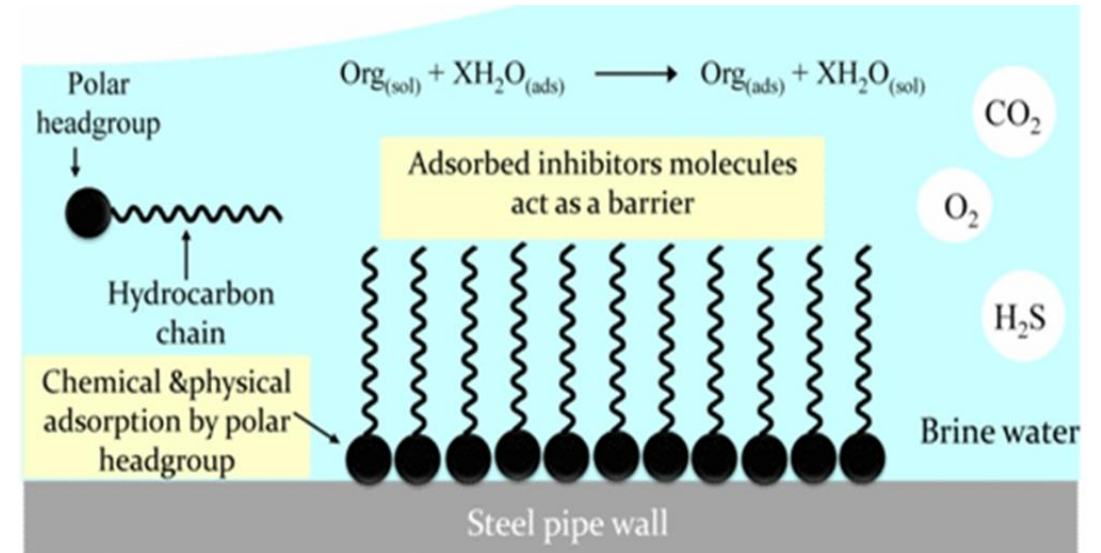
## PART I:

Organophosphate: adhesion benefits



## PART II:

Biobased solvent: hydrophobicity



# Water-based 2K Epoxy

BOP as performance enhancer

ASTM B117

DFT: 65 microns

Substrate: Cold Roll Steel



5% Ca phosphate



5% Ca phosphate  
+ 1% COMP



5% Ca phosphate  
+ 1% BOP



5% Ca phosphate  
+ 2% COMP



5% Ca phosphate  
+ 2% BOP

5% Calcium  
phosphate

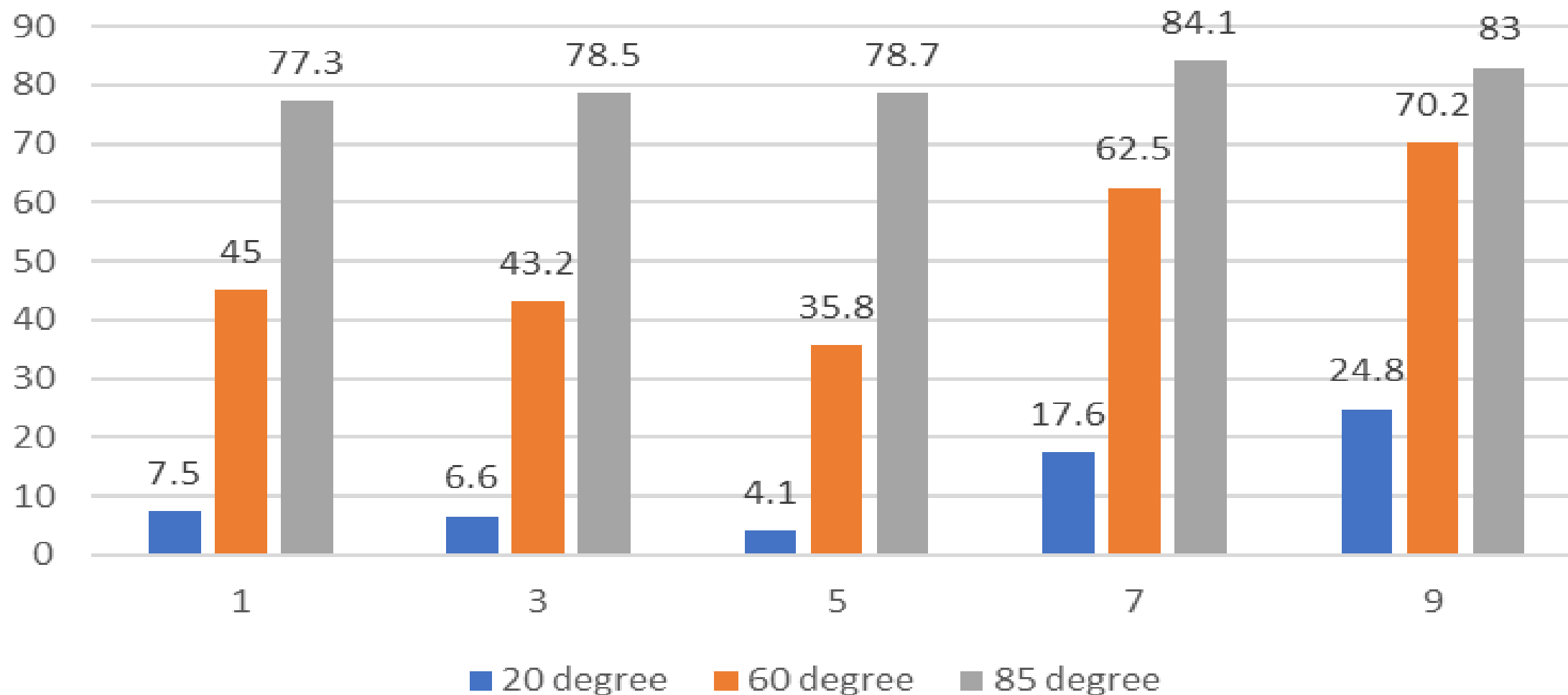
5% Calcium  
phosphate +  
1% BOP

5% Calcium  
phosphate +  
2% BOP

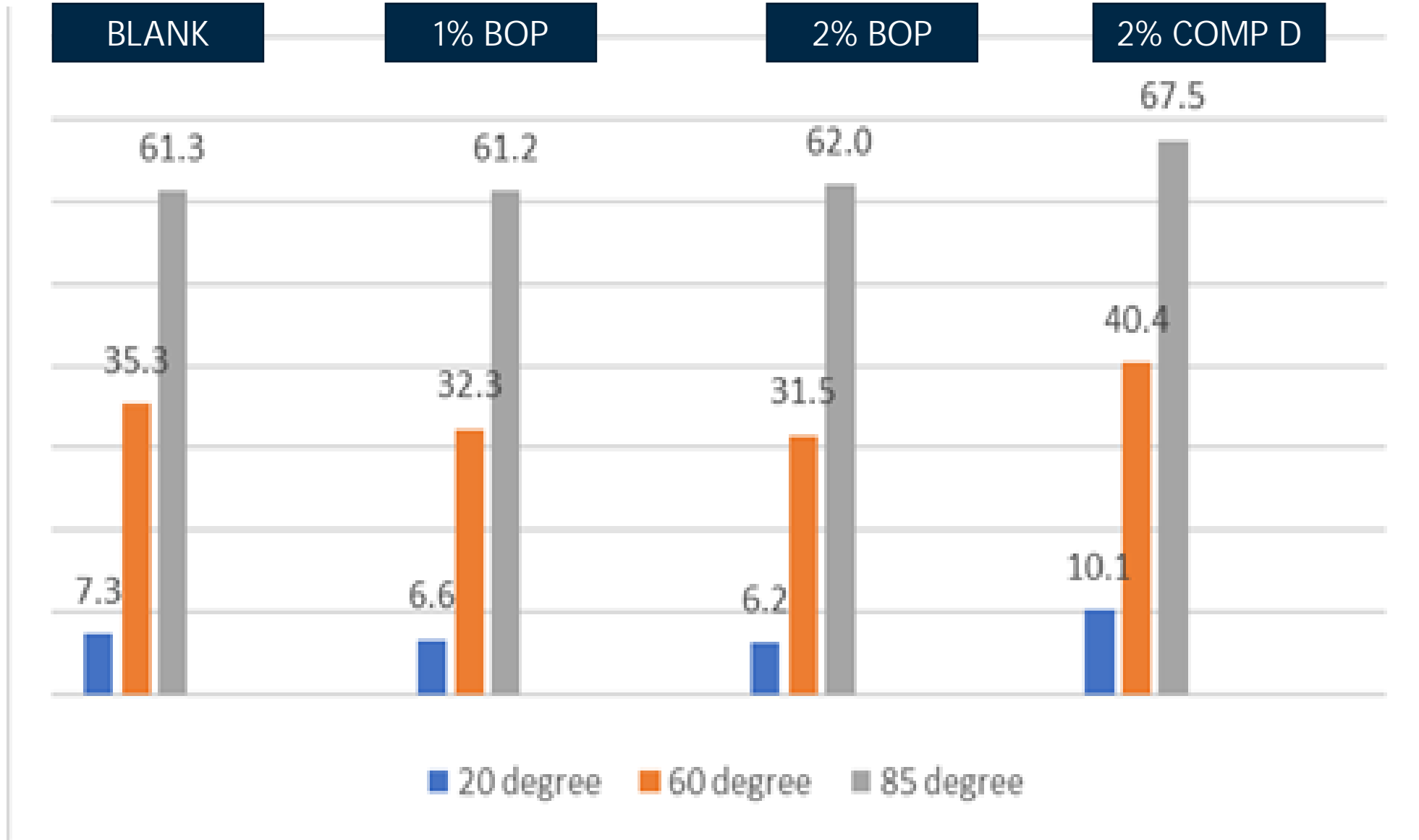
5% Calcium  
phosphate +  
1% COMP D

5% Calcium  
phosphate +  
2% COMP D

## 2K WB Epoxy Gloss



# WB Self-Crosslinking Acrylic





# WB Self-Crosslinking Acrylic

BOP as stand-alone corrosion inhibitor

ISO 9227: 120 hour NSS  
70-75 $\mu$ m

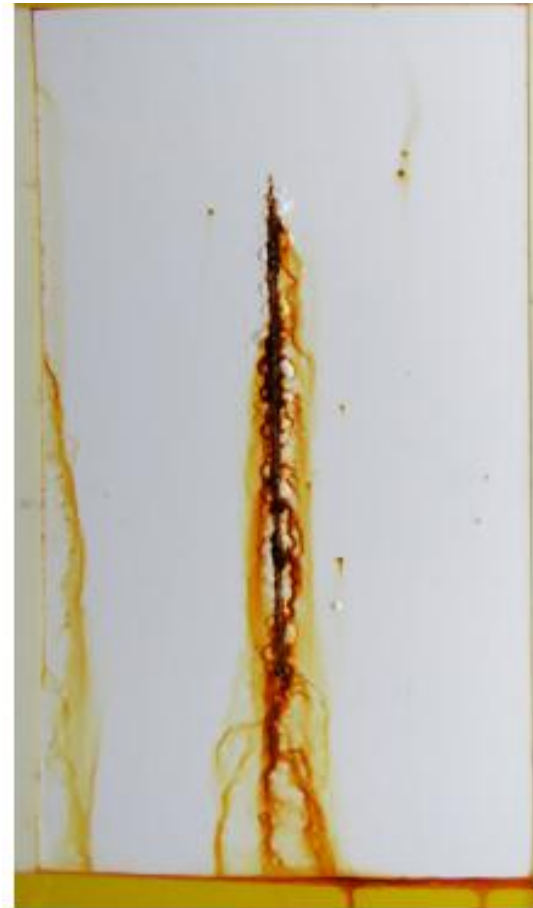
Single coat via drawdown  
Cold Rolled Steel

Before Spatula Scrape Test

CONTROL



2% COMP D



2% BOP



# WB Self-Crosslinking Acrylic

BOP as stand-alone corrosion inhibitor

ISO 9227: 120 hour NSS  
70-75µm  
Single coat via drawdown  
Cold Rolled Steel  
After Spatula Scrape Test

CONTROL



2% COMP D



2% BOP



# WB Self-Crosslinking Acrylic

BOP as stand-alone corrosion inhibitor, heavy metal replacement

ISO 9227: 1080 hour NSS  
120  $\mu\text{m}$  DFT; 2 coat application  
Cold Rolled Steel  
Scribed according to ISO 12944-6

CONTROL



2% COMP D



2% BOP



2% COMP C

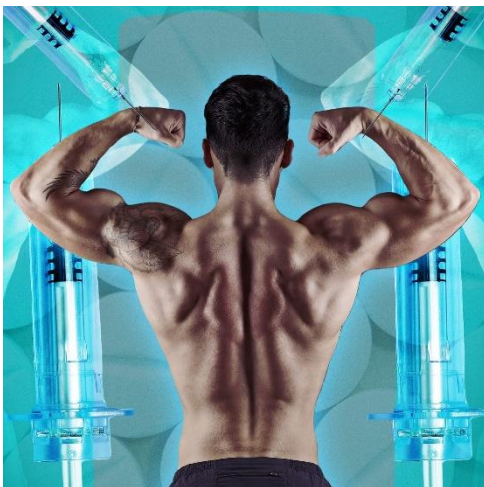


# Summary

Best practices for utilizing organic chemistries

## Performance enhancer

- Combining inorganic and organic corrosion inhibitors will take you to your next level of performance.



## Improve product labeling

- Organic inhibitors can be used to reduce or eliminate heavy-metal usage.
- VOC-free and bio-based products are economical and effective.



## Stand Alone Inhibitor

- Organic inhibitors can stand alone to...

“BUST THE RUST”







# Thank You

Future Proofing Corrosion Control

Visit us at Table 4!

Valuable contributions from the ICL P&C team:

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