

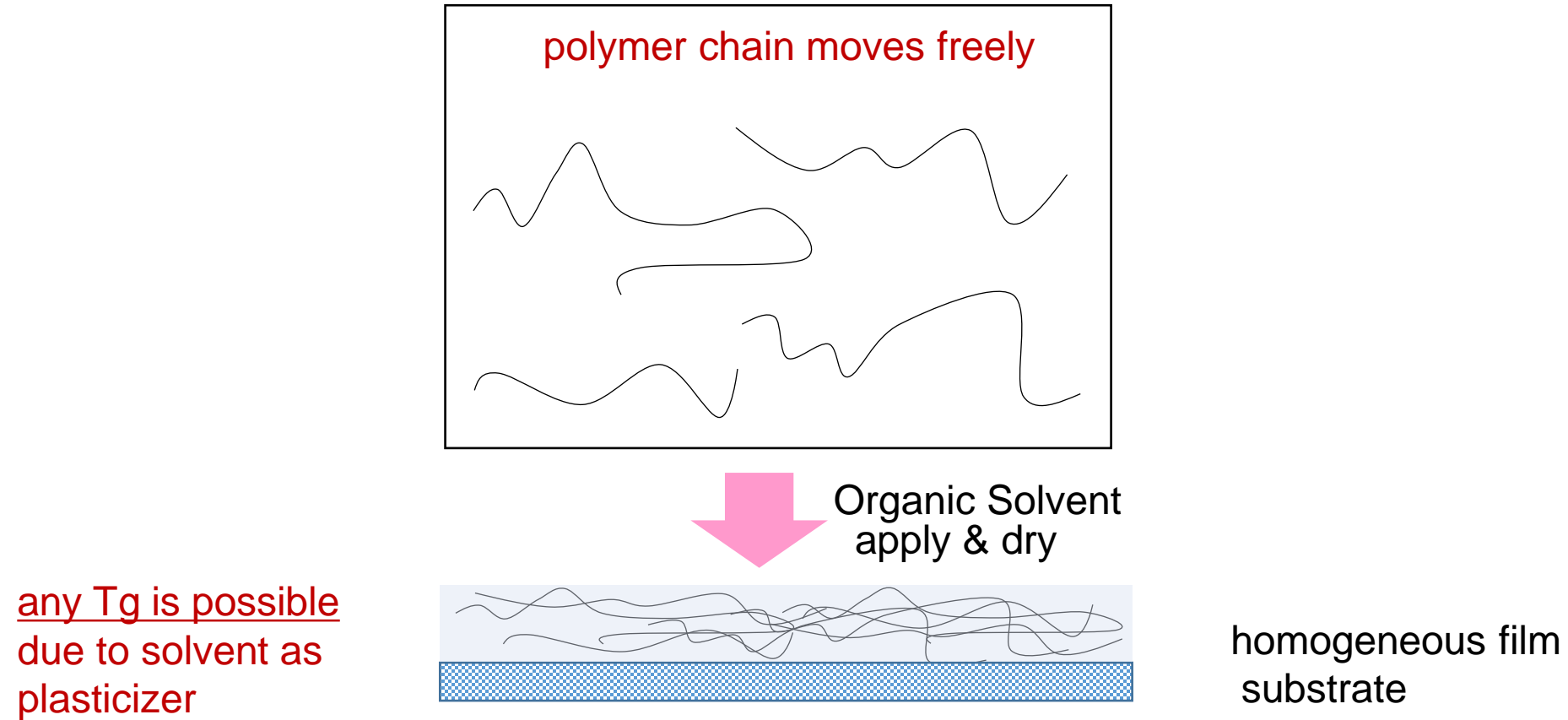


# A Cost-Effective Approach to Formulate Emulsions Leading to Improved Colloidal Stability and Finished Coatings

# Why Improve Coatings

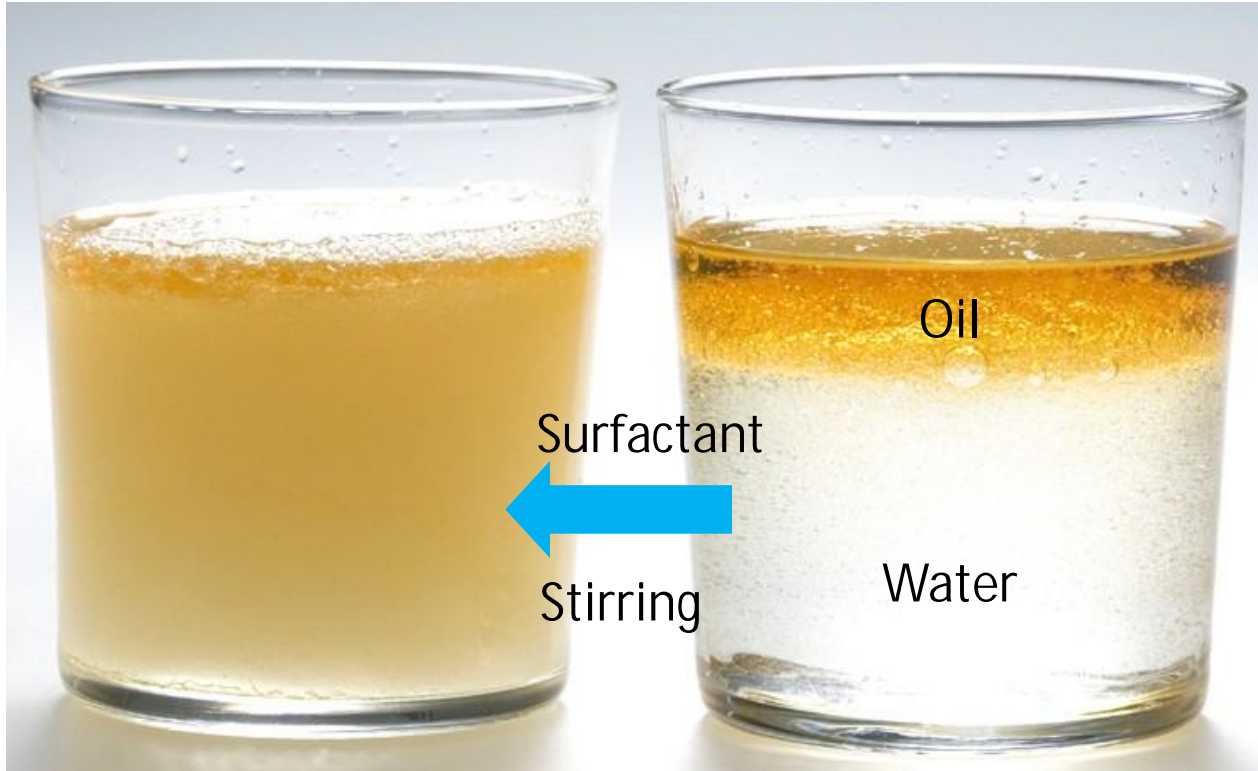
1. Growing demand for higher-quality coatings
2. Need for increased durability
3. Improved adhesion
4. More robust and efficient manufacturing processes
5. Desire for greater transparency

# Solvent-borne Coating (Homogeneous System)



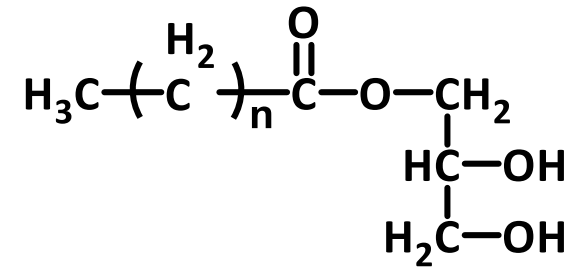
Sufficient entanglement (→ good strength) | Easy to cure by various systems | No hydrophilic impurities

# The Function of a Surfactant



Absorbs at water/oil interface  
and stabilizes oil droplets in  
water

Hydrophobic part

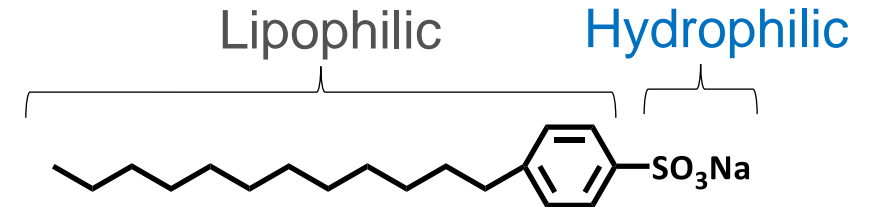


Hydrophilic part

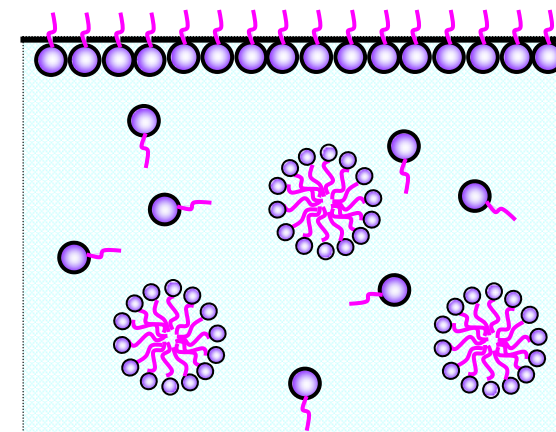
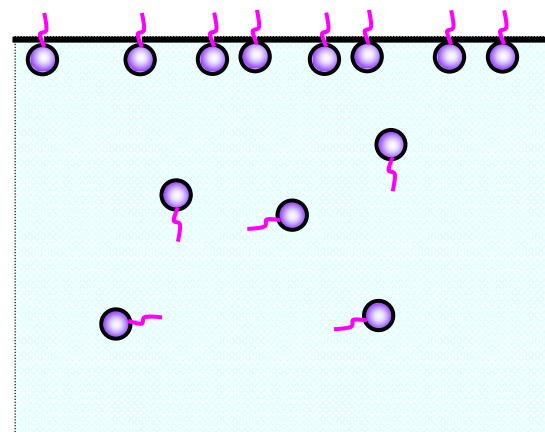
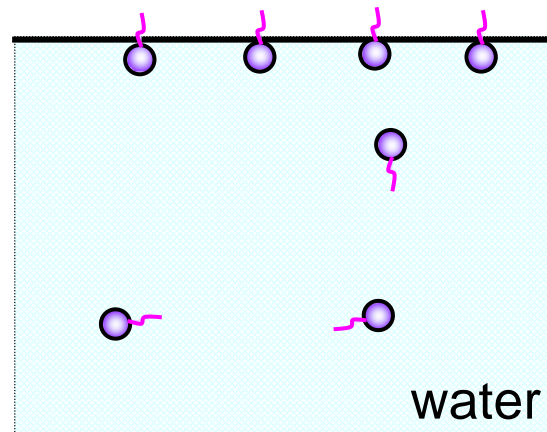
# Typical Surfactant only Emulsion Polymerization

Limited solubility to both water and oil (Styrene, acrylate etc)

→ Localizes at their interface or forms micelle



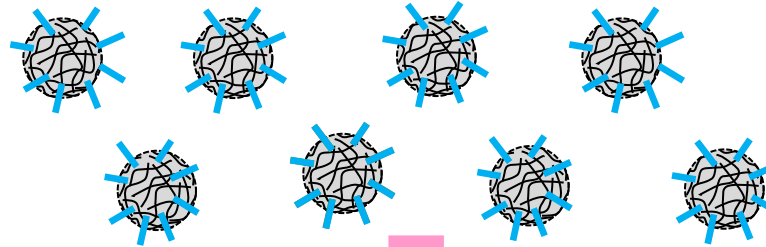
concentration



Surfactant Micelle

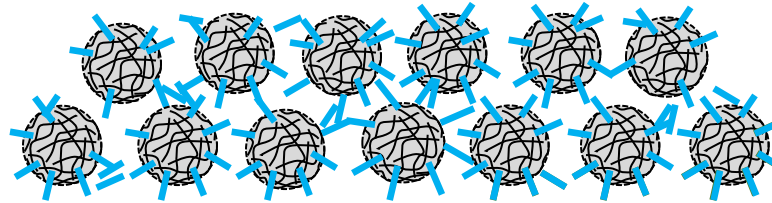
# Water-borne Coating (Heterogeneous System)

polymer particle  
in water (2 phases)



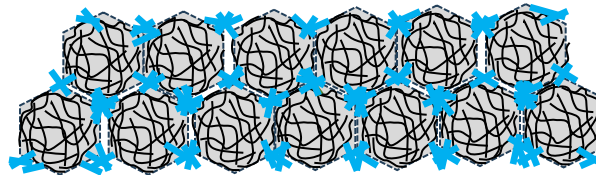
colloidal stability is  
obtained by physically  
adsorbed surfactant

(2 → 1 phase)



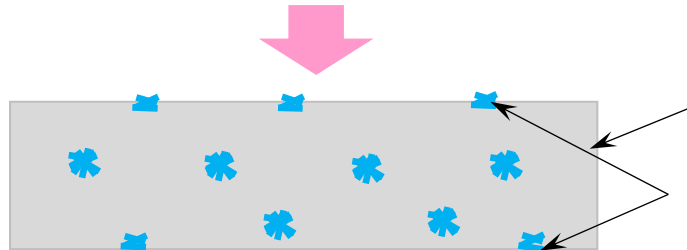
surfactant gradually  
desorb

polymer chain can  
only diffuse over  $T_g$



surfactant is a  
nuisance at this stage!  
(no place to stay)

homogeneous film  
is formed over  $T_g$



absorbing water, causing  
blister & whitening

hindering adhesion

# What is Emulsion Polymerization

## Ingredients

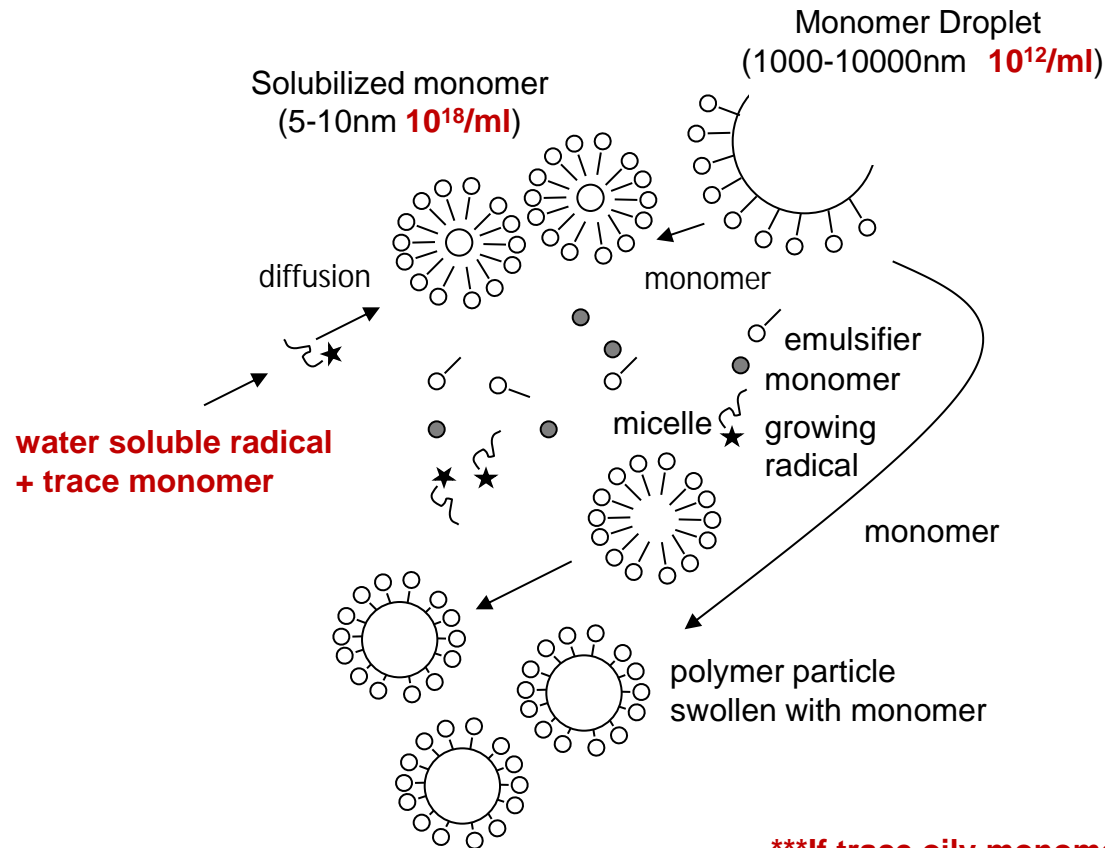
oil-soluble monomer\*  
(\*trace amount dissolves in water)  
surfactant  
**water-soluble initiator\*\***  
water

**Primary radical**  
generates  
in water phase

## Where primary radical go ?

to monomer micelle or  
**trace monomer in water phase\*\*\***

Because total surface area  
micelle >>> monomer droplet



**\*\*If oil-soluble initiator is used,**  
polymerization proceeds in large  
droplet and then these aggregate

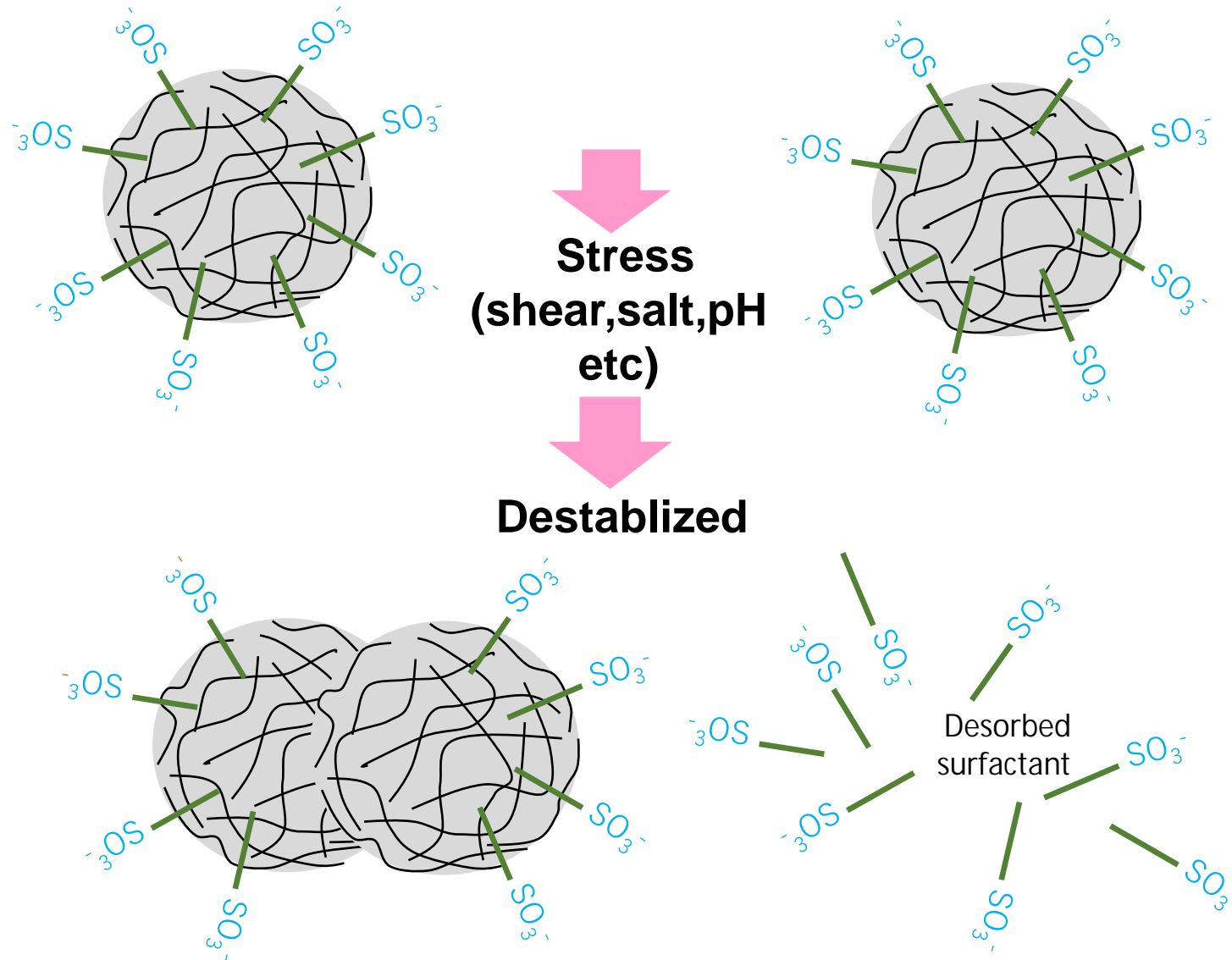
**\*\*\*If trace oily monomer exists in water ,**  
oily oligomer generates and  
enter into micelle. No problem!

**\*\*\*If water-soluble monomer exists in water,**  
water-soluble polymer generates.\*\*\*\*  
These polymer gives bad affect to emulsion  
and film property.

**\*\*\*\*How to prevent this is critical !**



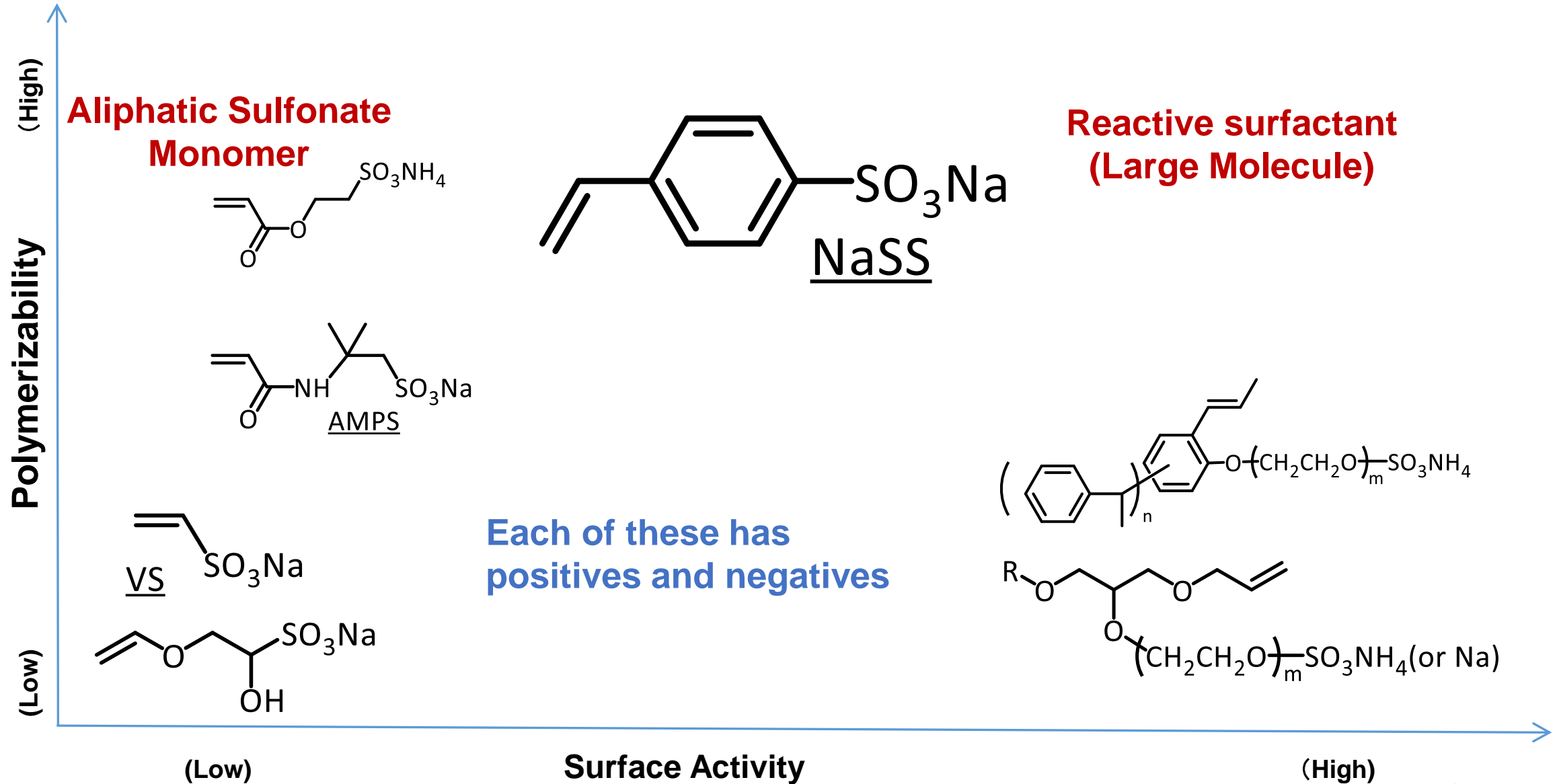
# Emulsion Stabilized by Conventional Surfactant



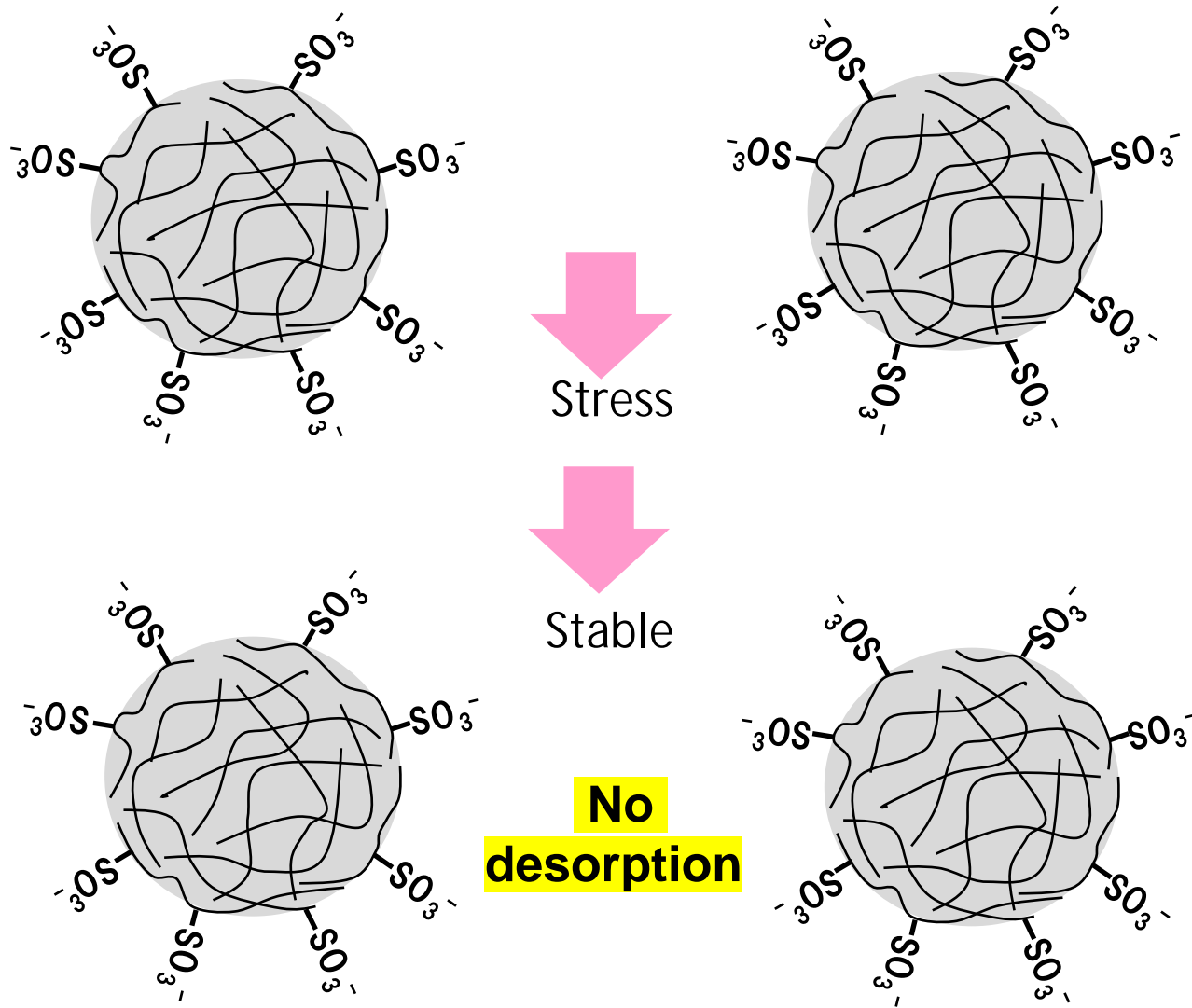
Aggregation



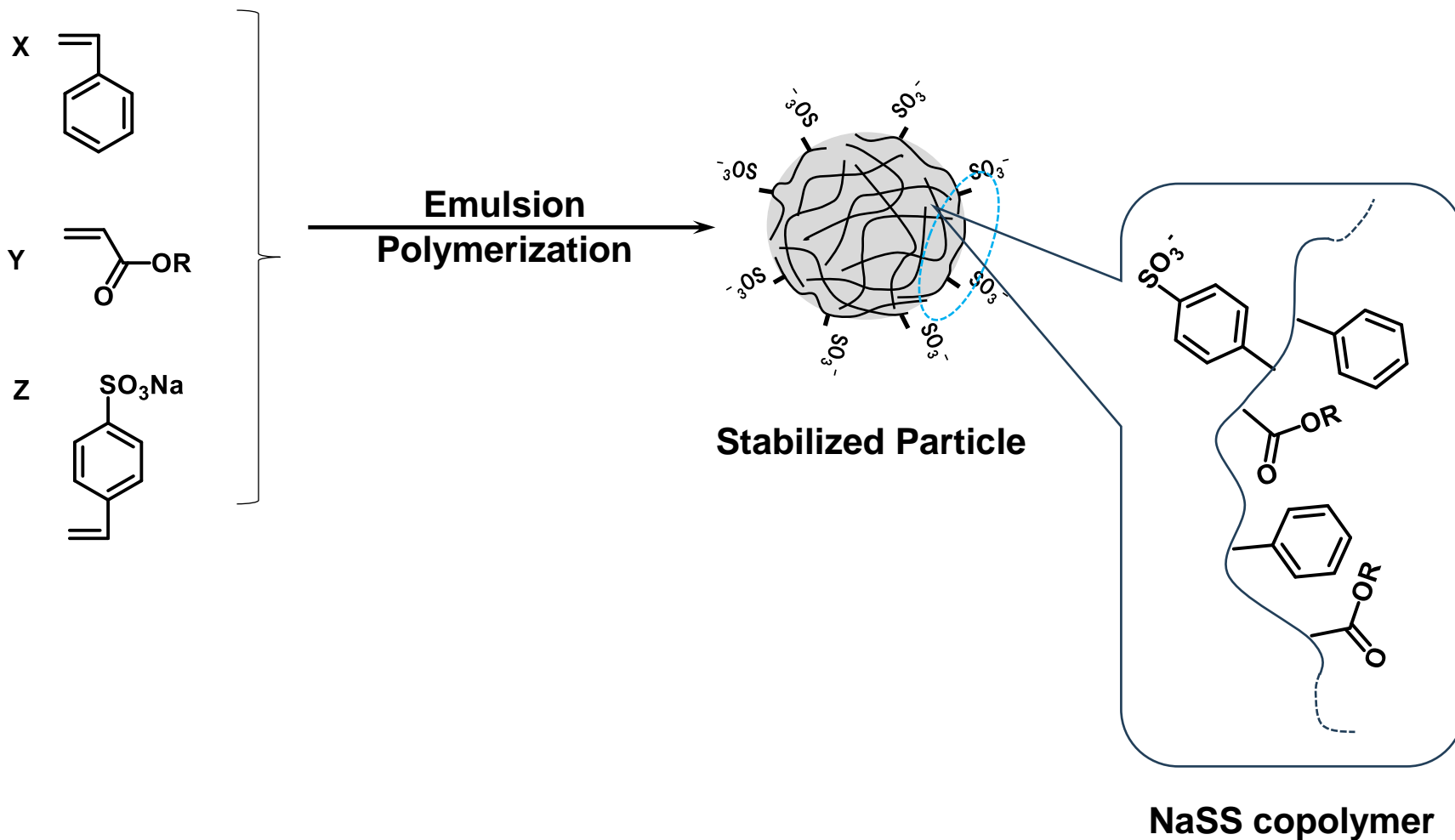
# Reactive Emulsifiers



# Emulsion Stabilized by NaSS (ideal image)



# Emulsion Stabilized by NaSS (closer look)



# Major Advantages/Disadvantages

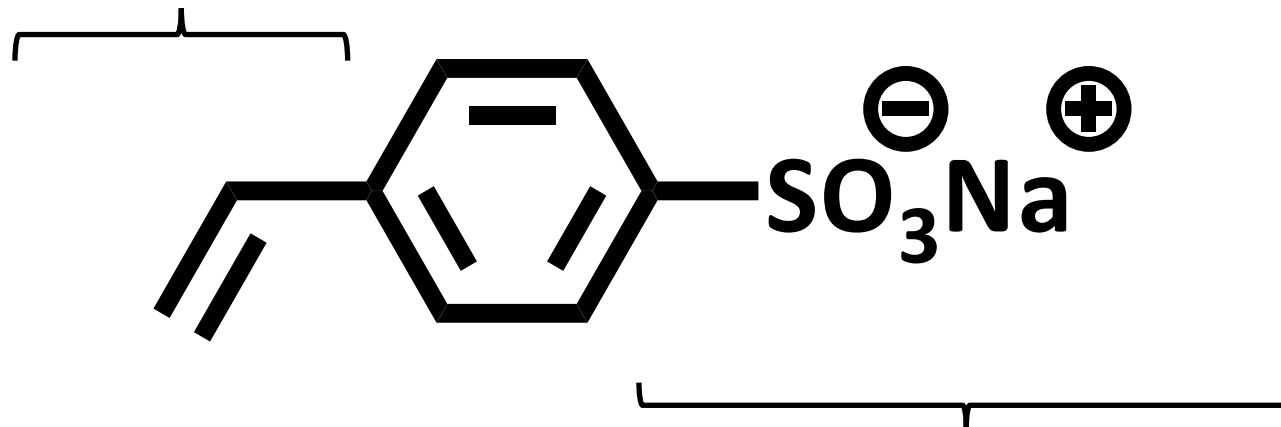
1. Small RE molecules (AMPS, SVS) lack emulsification
2. Large RE molecules have a higher molecular weight and usage and cost/kg
3. Large RE molecules can also act as a defect similar to emulsifiers
4. While NaSS isn't a 'Superman' solution, it does offer significant advantages

# Sodium Styrene Sulfonate

**(1) Good Surface Activity**



**(2) High Reactivity**



**(3) High Thermal Stability**



# Surfactant-free Emulsion Polymerization by NaSS

Polymerization proceeds by small dosage of NaSS

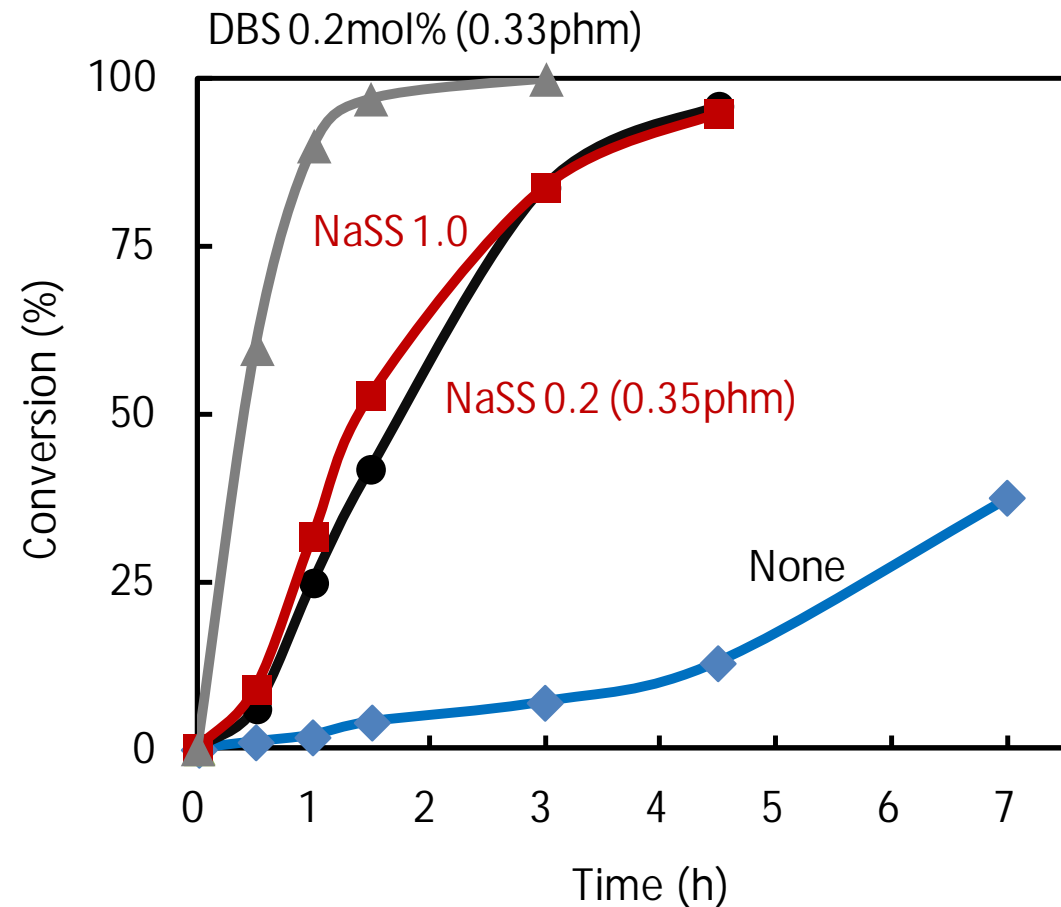
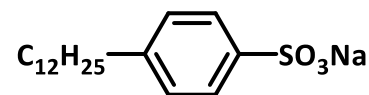
Ingredients	
Styrene/n-Butylacrylate	= 1/1 wt.r
NaSS or DBS*	= variable
Potassium Persulfate	= 0.2 mol%/MM
Water	(Total MM=20wt%)



Dosed all at once

70°C×5h

\*DBS



Polymerization time vs. conversion



# Surfactant-free Emulsion Polymerization by NaSS

Optimal NaSS dosage  $\sim 0.5$  mol%

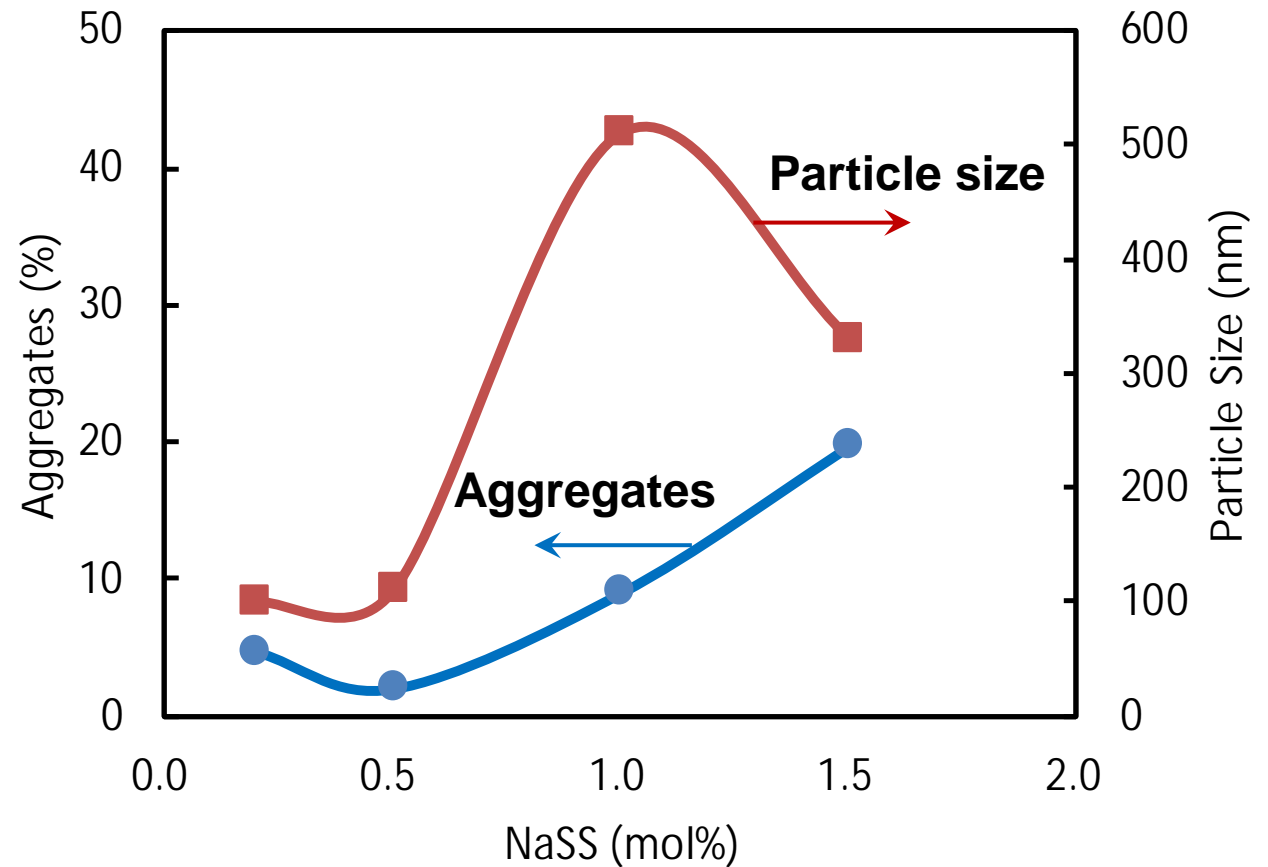
## Ingredients

Styrene/n-Butylacrylate	= 1/1 wt.r
<b>NaSS</b>	= <b>variable</b>
Potassium Persulfate	= 0.2 mol%/MM
Water	(Total MM=20wt%)



**Dosed all at once**

**70°Cx5h**



NaSS dosage vs. aggregates and particle size

# Practical Example of Emulsion Polymerization

## Condition

**Pre-emulsified monomer  
was continuously dosed**

**Nonionic surfactants were  
used to emulsify monomer**

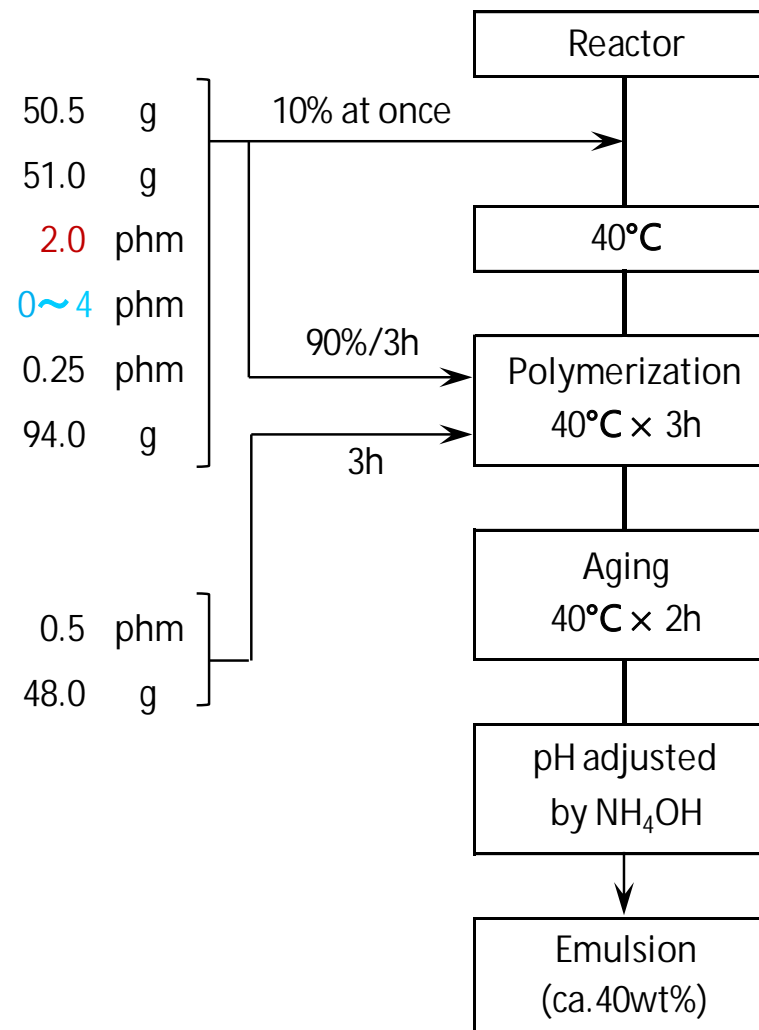
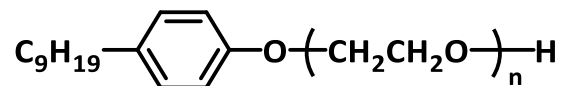
### Monomer Emulsion

Styrene	50.5	g
n-Butylacrylate	51.0	g
Sulfonate monomer*	2.0	phm
Surfactant**	0~4	phm
Sodium bisulfite	0.25	phm
Water	94.0	g

### Initiator

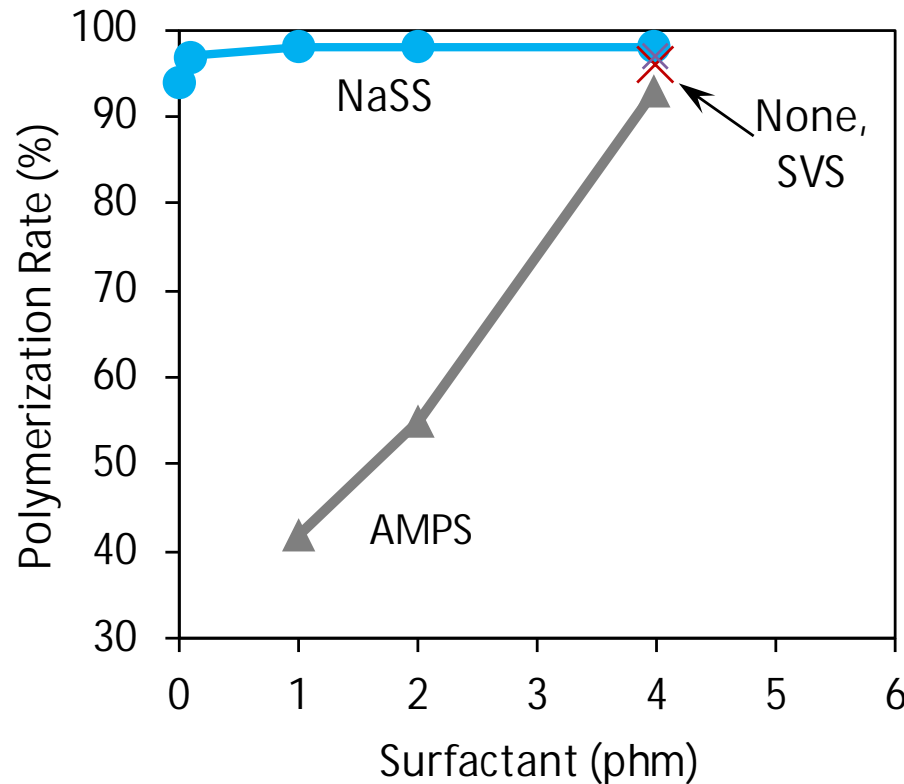
Ammonium persulfate	0.5	phm
Water	48.0	g

\*\*Surfactant



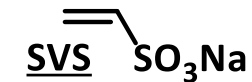
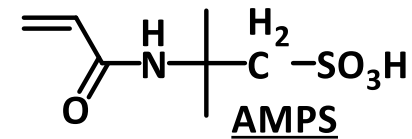
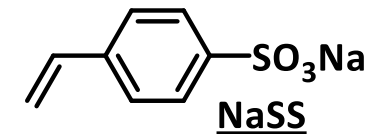
# Benefits of NaSS : Polymerization Rate

Emulsion polymerization proceeds up to high conversion with less surfactant



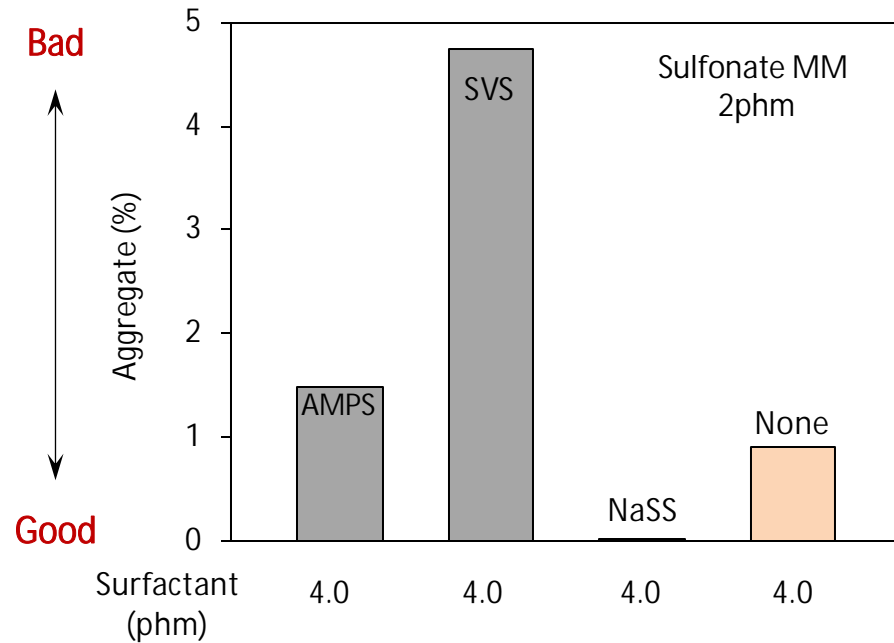
Surfactant dosage vs. conversion

\*Sulfonate monomer  
(2phm  $\approx$  0.9~1.7mol%)

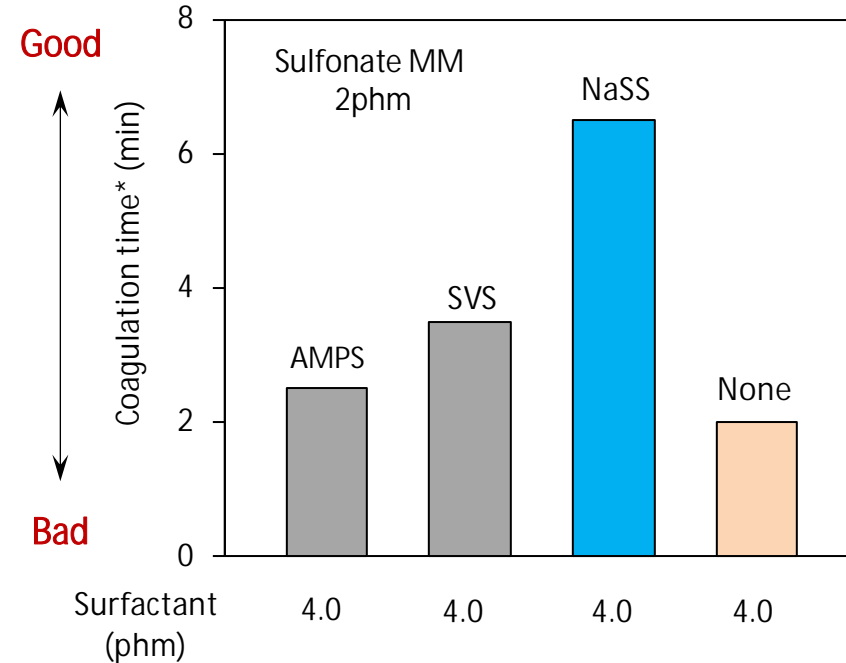


# Benefits of NaSS : Emulsion Stability

NaSS gives excellent colloidal stability



**Aggregation during polymerization**



**Mechanical stability of emulsion**

\*by agitation at 5,000rpm (Maron test)

# Benefits of NaSS : Emulsion Properties

NaSS gives good film-forming property

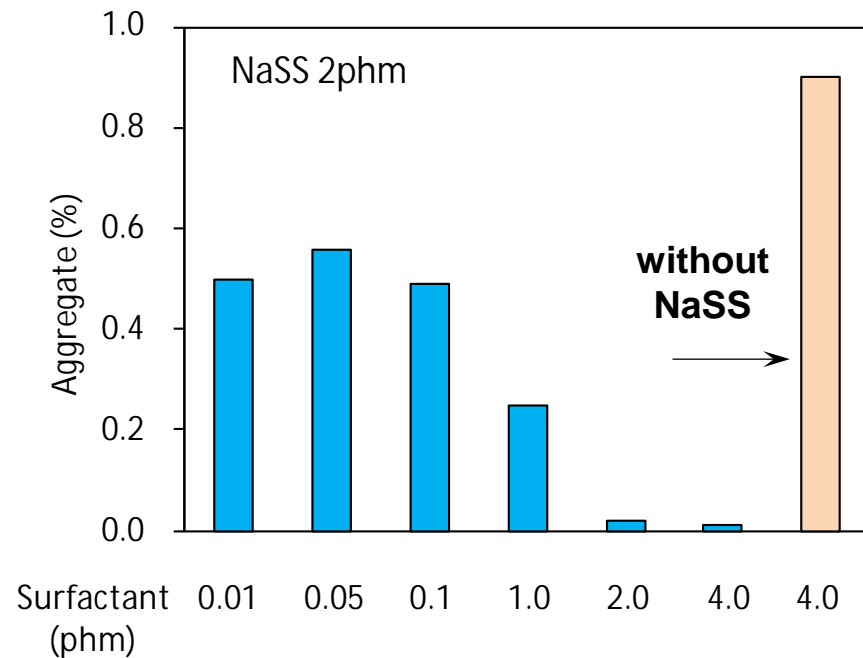
Sulfonate monomer (phm)		Emulsifier (phm)	Aggregate (%)	Particle diameter (μm)	Mechanical* Stability (min)	Appearance of dry film	Water absorption of film (%)**
None	0.0	4.0	0.90	0.19	2.0	translucent	12.7
NaSS	2.0	4.0	0.01	0.09	6.5	transparent	12.0
AMPS	2.0	4.0	1.48	0.19	2.5	opaque	11.8
SVS	2.0	4.0	4.74	0.20	3.5	translucent	18.3

\* Coagulation time by agitation at 5,000rpm (Maron test)

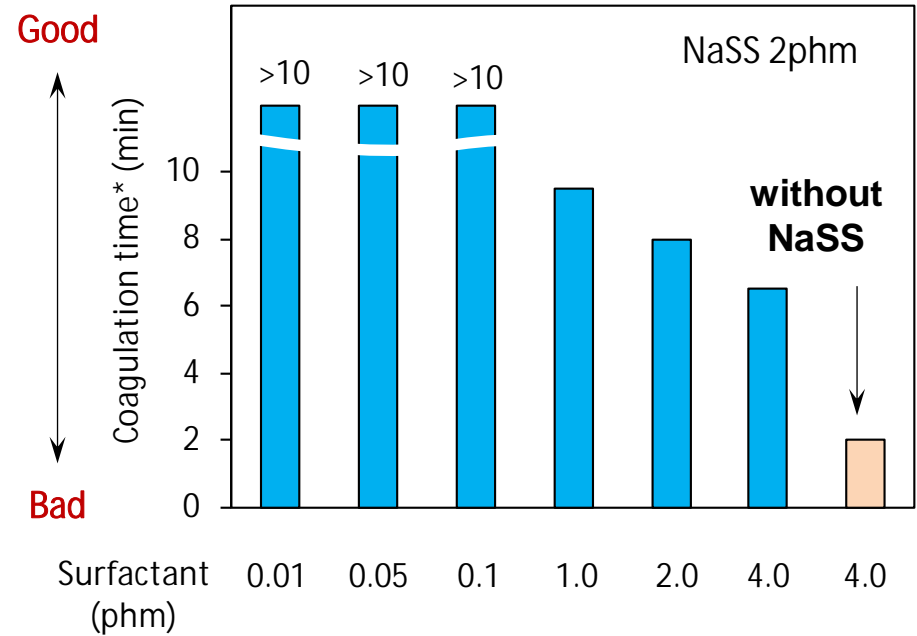
\*\* Immersion of dried-film in water for 48h at RT, film thickness ~0.5mm

# Benefits of NaSS : Reduction of Surfactant

Excellent colloidal stability is obtained with less surfactant.



**Aggregation during polymerization**



**Mechanical stability of emulsion**

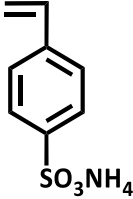
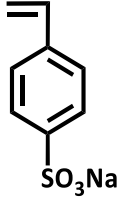
\*by agitation at 5,000rpm (Maron test)



# AmSS

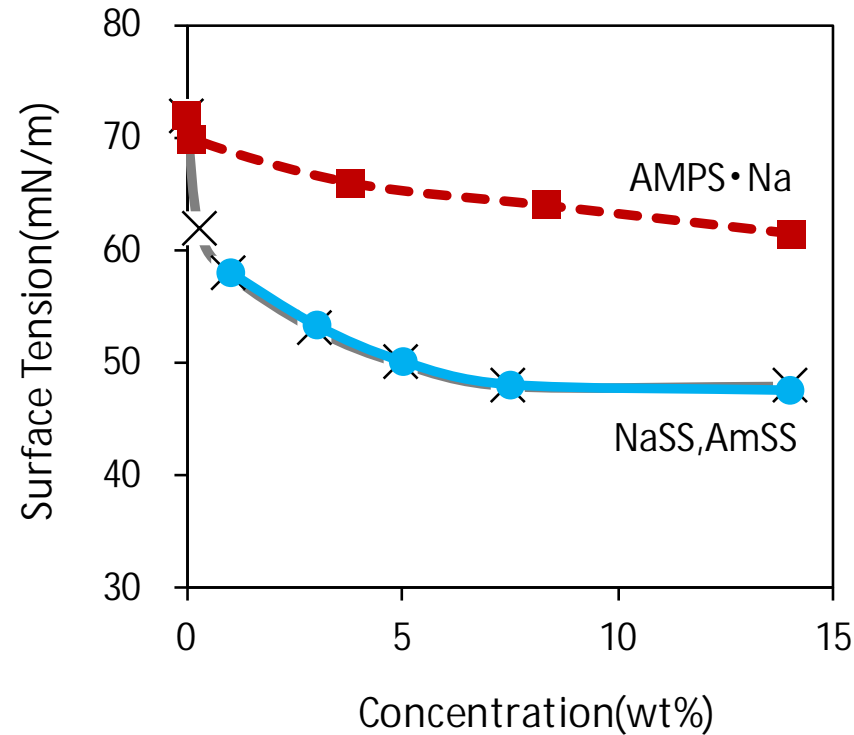
## Ammonium Styrene Sulfonate

1. Metal-free
2. High surface activity
3. High reactivity
4. Organic solvent soluble
5. Higher corrosion resistance compared to NaSS

	 <b>AmSS</b>	 <b>NaSS</b>
<b>Registrations</b>		
CAS No.	19922-72-6	2695-37-6
MITI(JAPAN)	3-1948	3-1903
TSCA(USA)	-	Listed
REACH(EU)	-	Registered
ECL(KR)	-	KE-13273
<b>Specification</b>		
Purity wt%	≥ 95	≥ 84
Metal wt%	< 0.5	11~14
pH of 10wt%aq.	5~6	10~12
<b>Solubility at 25°C wt%</b>		
H <sub>2</sub> O	26.0	21.0
Methanol	13.2	4.4
DMSO	43.9	27.3
NMP	31.0	8.6

# Surface Activity

Like NaSS, AmSS demonstrates good compatibility with emulsion polymerization processes



**Aqueous solution conc. vs Surface tension**

(Wilhelmy method, Pt plate at 25°C)

# Surfactant-Less Emulsion Polymerization by AmSS

Polymerizes efficiently to high conversion levels.

## Ingredients

Styrene/n-Butylacrylate = 1/1 wt.r  
**AmSS or AMPS·NH<sub>4</sub>** = 1 mol%  
DBS·NH<sub>4</sub><sup>\*</sup> = 0.02 mol%/MM  
Ammonium Persulfate = 0.10 mol%/MM  
Water (Total MM=34wt%)

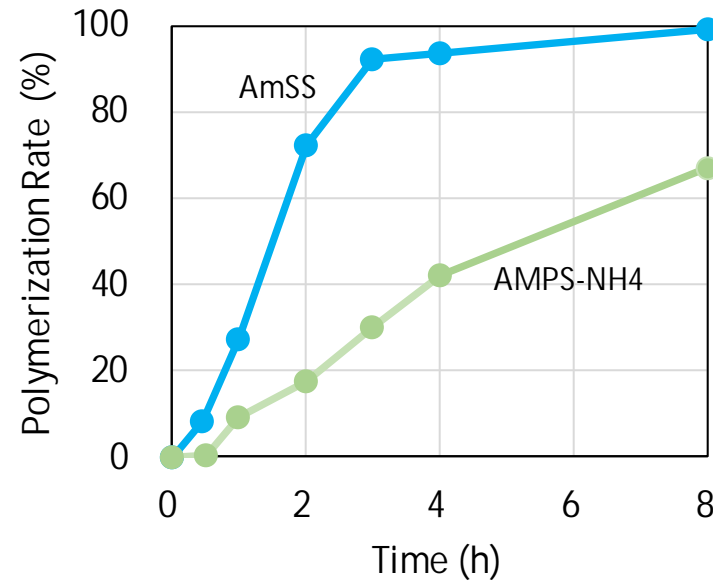
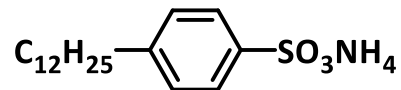


Dosed all at once

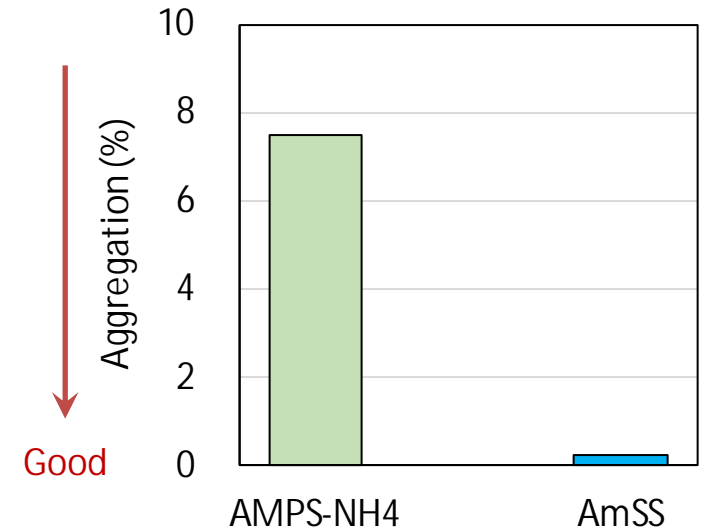
60°C



\*DBS·NH<sub>4</sub>



Time vs. conversion



Aggregation during polymerization

# Advantages of Being Metal-free

AmSS enhances the water resistance properties of the coating

## < Model Polymer Synthesis >

- Monomer  
St/n-BA : 50/50wt.r  
AmSS(NaSS) : 0.5 or 2 mol%  
Overall : 48wt%
- Initiator V-40 : 1mol%
- Solvent : N,N-Dimethylacetamide



Temp. : 90°C×46h  
Overall conv. : >99%

## < Film Preparation >

- Substrate : Slide Glass
- Dry : 100°C×24h,full vacuum
- Thickness : ca.0.15mm

## < Water Resistance >

- Immersed in water at 40°C for 25h

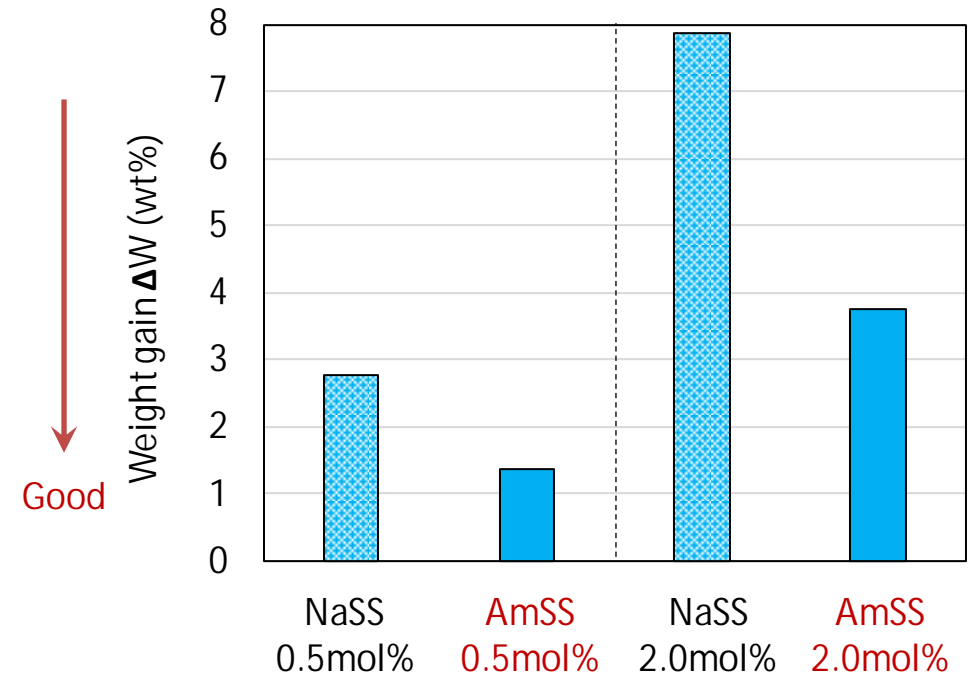


Fig.14 Water absorption of dry film

# Contact

## **BOOTH NO. 26** **TOSOH USA Inc.**

**Tosoh USA, Inc.**  
**<https://www.tosohusa.com>**

3600 Gantz Road  
Grove City, OH 43123  
Tel: +1-614-277-4348  
Fax: +1-614-875-8066  
E-mail: [michael.east@tosoh.com](mailto:michael.east@tosoh.com)