



Polycaprolactone Polyols in 2K Polyurethane Coatings

Chuck Jones, PhD

Market Segment Manager - Coatings



AGENDA

1. Introduction to Ingevity
2. Chemistry of Polycaprolactone Polyols
3. Experimental Study
4. Results and Discussion
5. Conclusions



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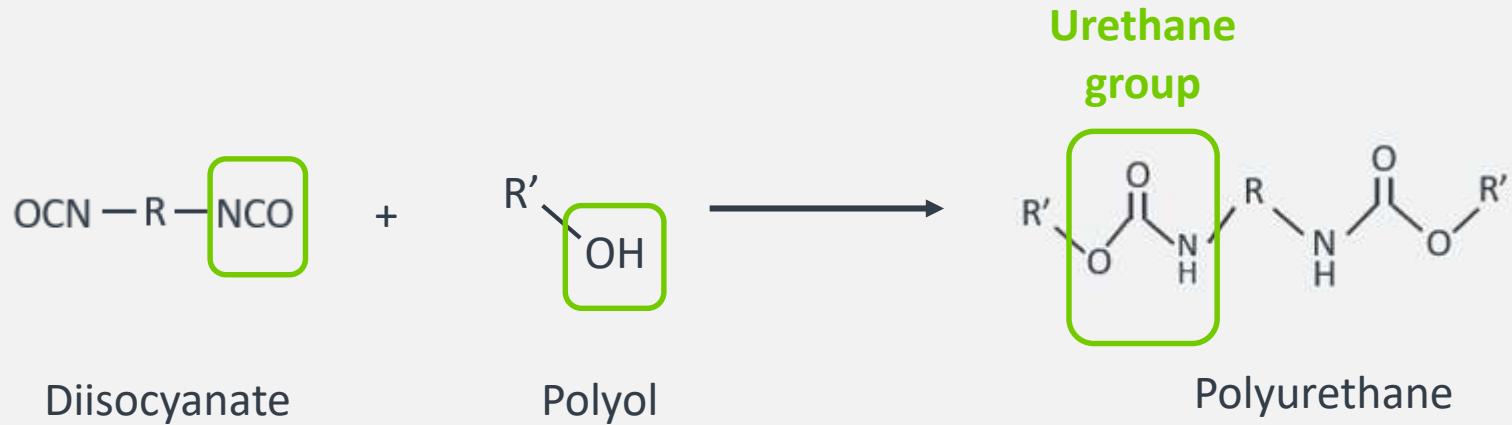
Introduction



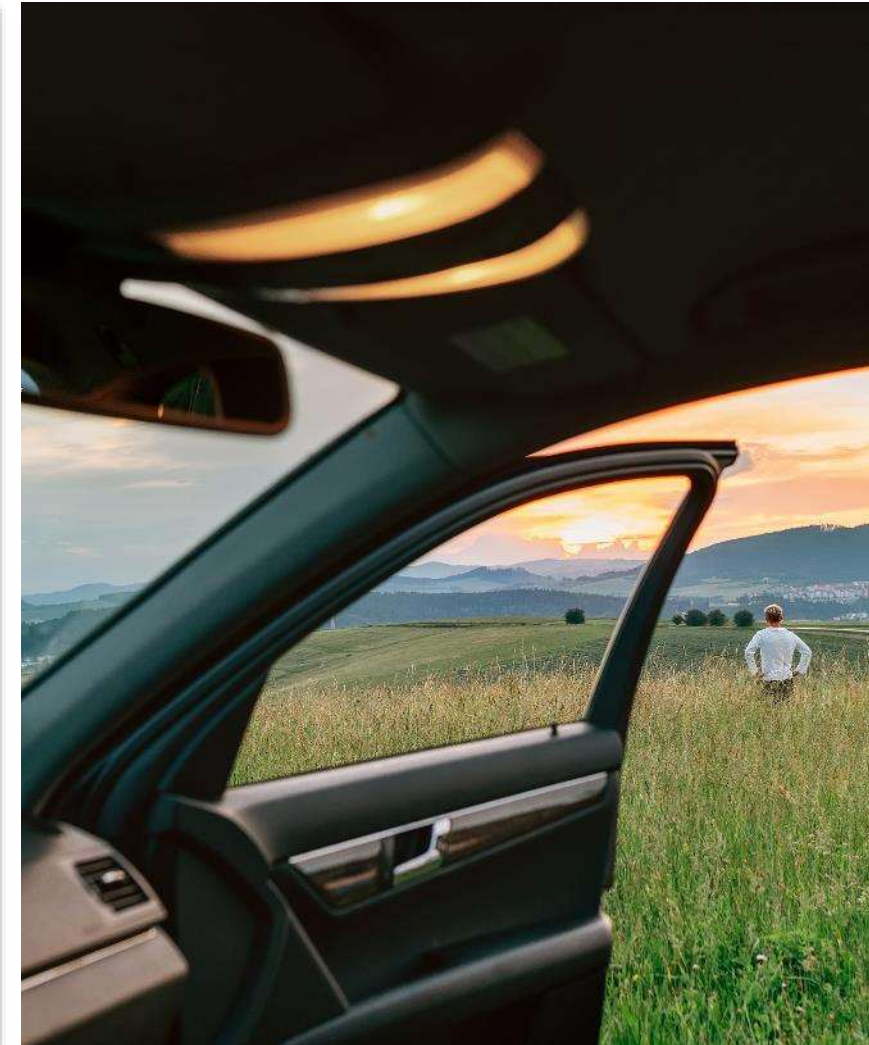
Polyurethane Coatings

High performance, versatile coatings

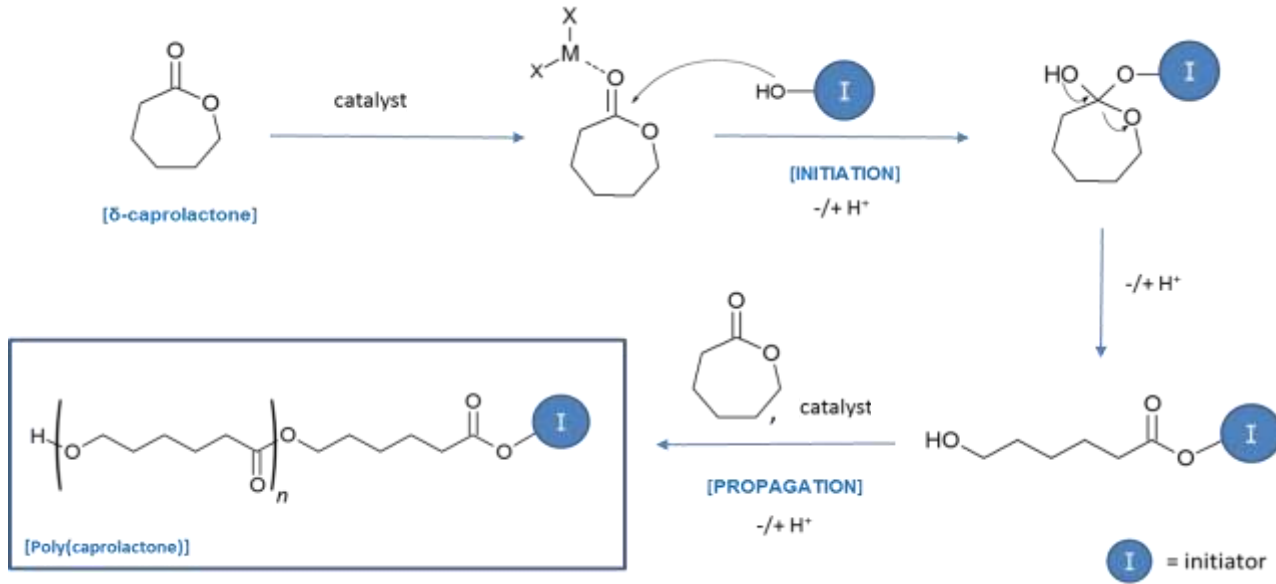
Polyurethane reaction mechanism



A polyurethane is created by reacting an **alcohol (polyol)** with a **diisocyanate**.



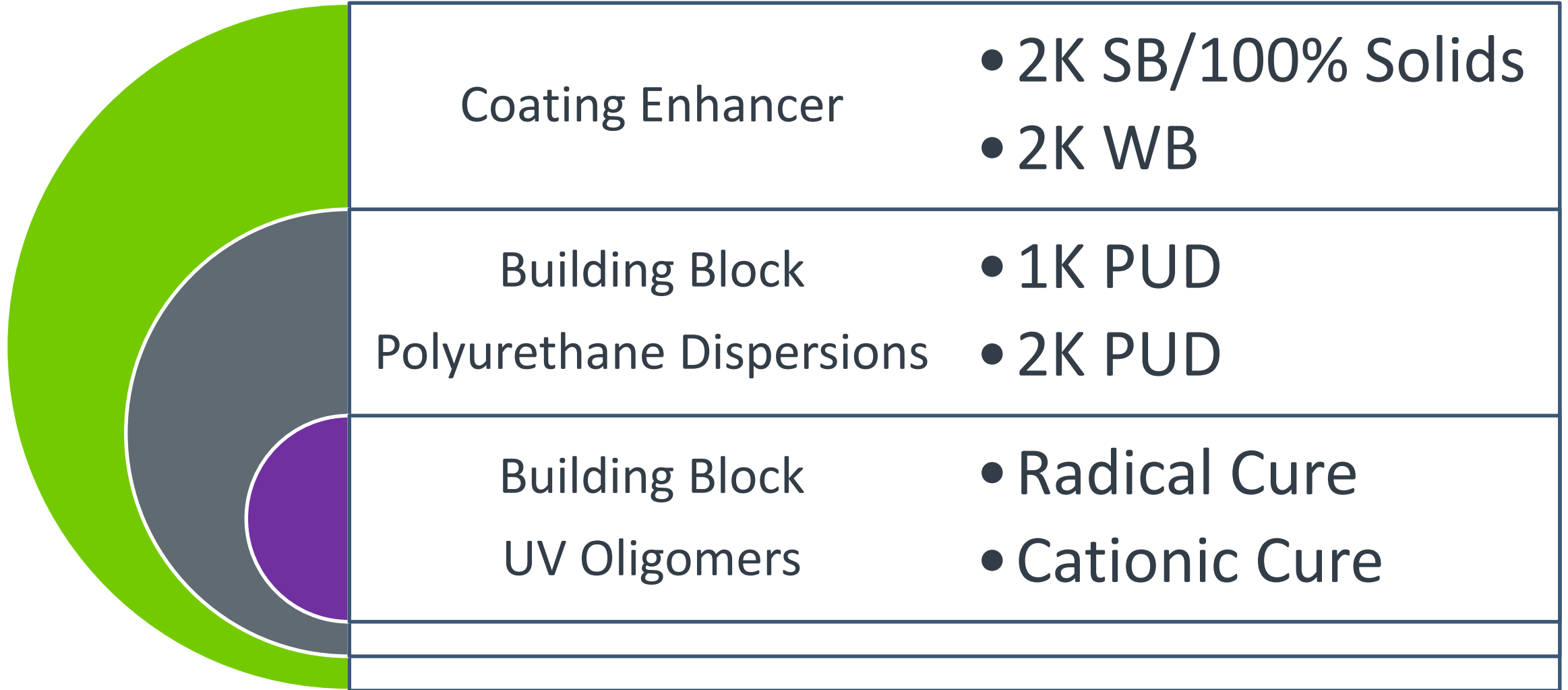
Polycaprolactone (PCL) Polyols



- Made *via* **ring opening polymerization**
- **No by-products**
- Very **low acid values** and water content
- Controlled polymerization – **all hydroxyls are primary**
- Very **narrow polydispersity**



PCL Polyol Uses in Coatings



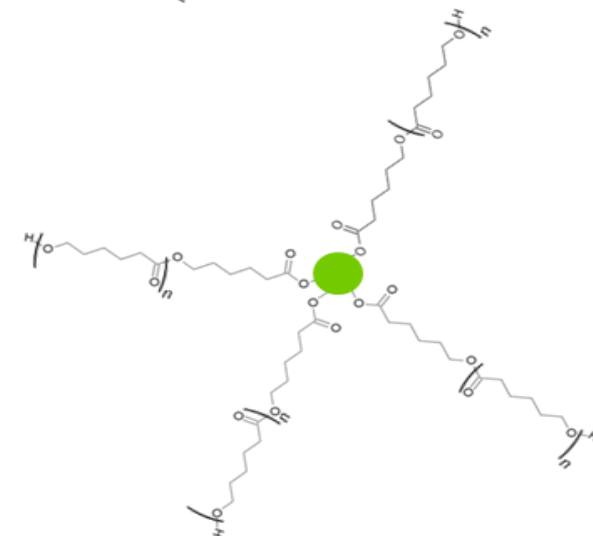
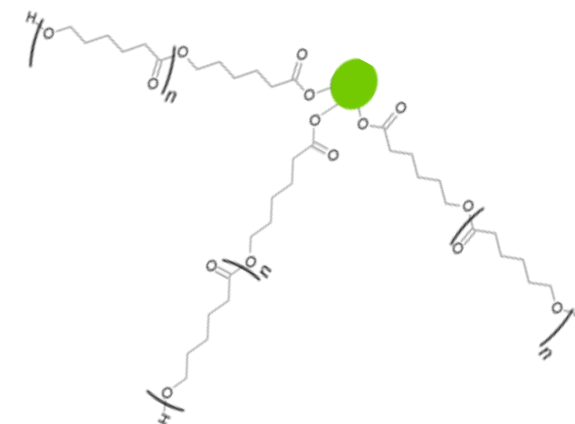
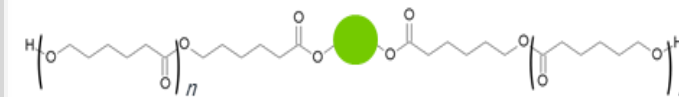
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Experimental

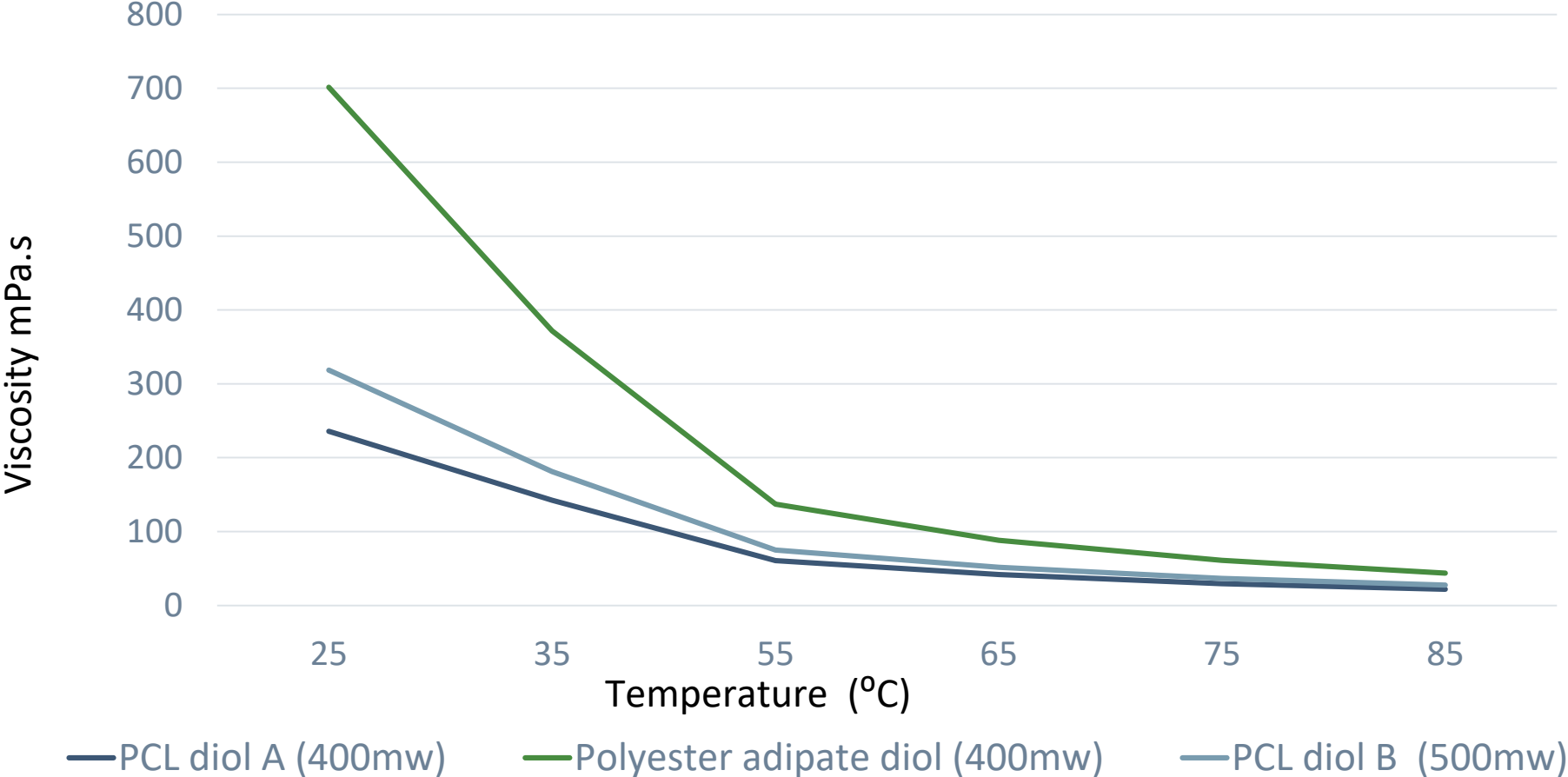


Polyols Used for Testing

Polyol	M _w	Functionality	Solid Content (%)	Viscosity (mPa·s @ 25°C)
Acrylic polyol	1073	2.0	70	3000
PCL diol	400	2.0	100	240
PCL triol A	300	3.0	100	1600
PCL triol B	500	3.0	100	1200
PCL triol C	900	3.0	100	1250
PCL tetrol	1000	4.0	100	1800
Polyester adipate diol	400	2.0	100	705



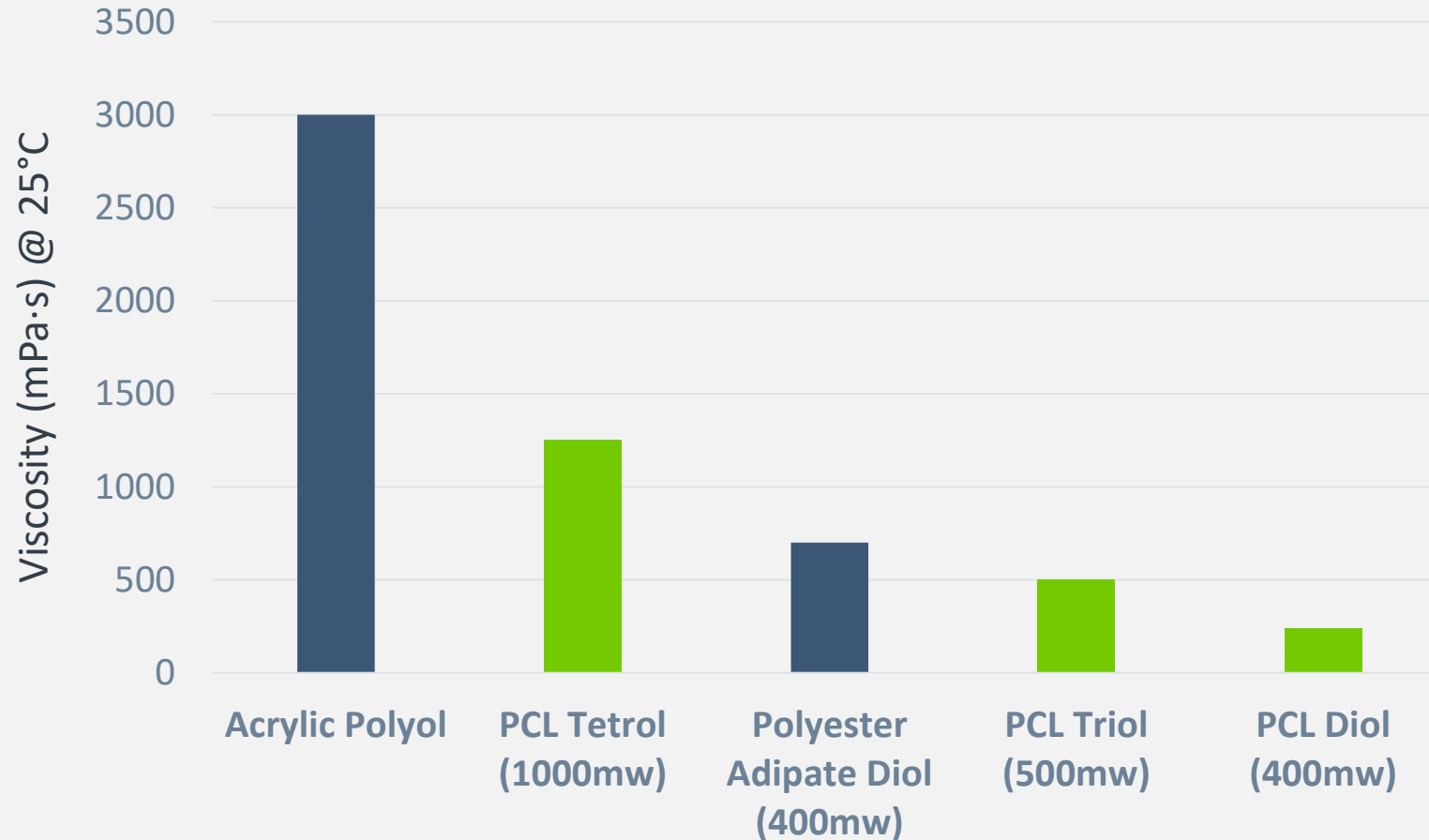
Polyol Properties



Lower viscosity enables higher solids (and lower VOCs)

Polyol Properties

Viscosity differences at room temperature



Experimental Formulation

Raw Material	Function	Weight % (in total formulation)
Hydroxyl functional Acrylate	Solventborne acrylic polyol	57
PCL triol C (900mw)	Polycaprolactone polyol	6
Add above in order under medium shear mixing. Mix for 10 minutes.		
Defoamer additive	Defoamer	0.75
Wetting agent additive	Wetting agent	0.25
DBTL (10% in butyl acetate)	Catalyst	1
Xylene/Butyl Acetate/MPA (1:1:1)	-	9
Add above in order under medium shear mixing. Mix for 10 minutes.		
HDI based Diisocyanate	Crosslinker	21
Butyl Acetate/Xylene (1:1)	-	5

Weight solids: 67%

Viscosity: 45-55 cps

Isocyanate index: 1.00

Application Parameters

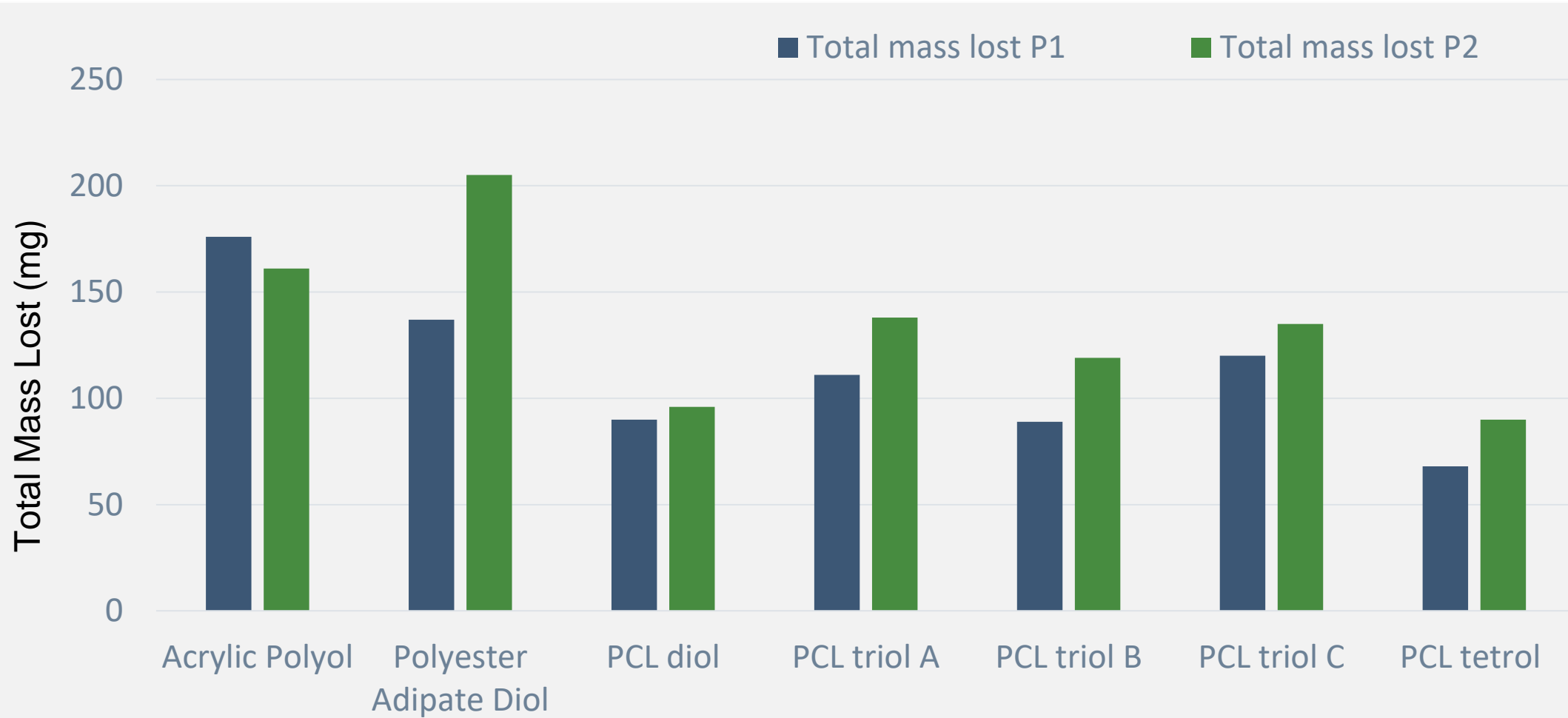
Method of Application	D.F.T (Dry film thickness - μm)	Substrates	Environmental
K- Bar – 100 μm	50-55 μm	Degreased Aluminum, Degreased Mild Steel	23°C \pm 3°C 50% RH \pm 5%

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Results & Discussion



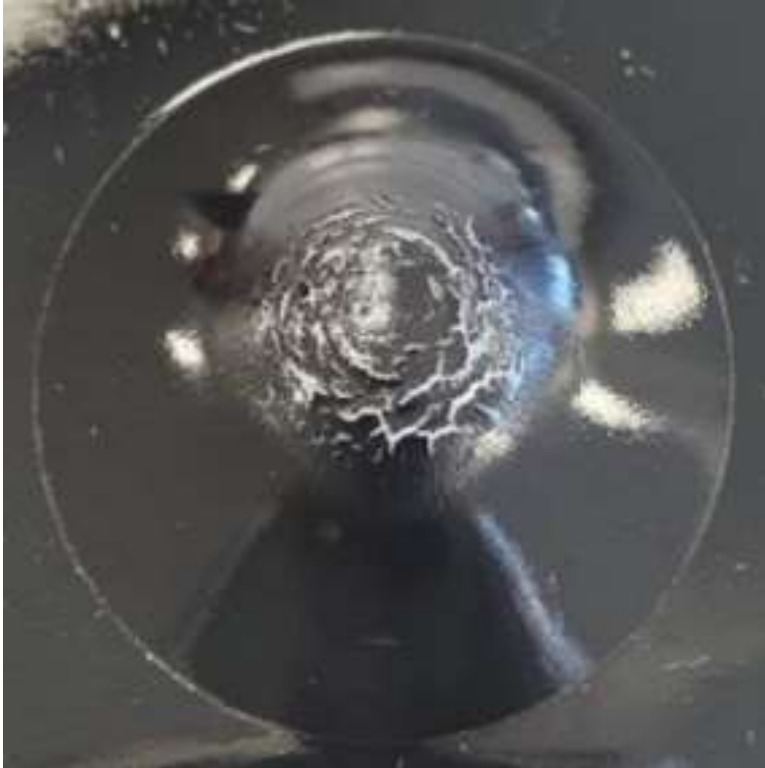
Taber® Abrasion Testing



- Acrylic-based 2K coatings modified with PCL polyols show **improved abrasion resistance**
- **Increased cross-linking**, provided by the PCL tetrol, shows the best results

Rapid Deformation Impact Resistance Test (ISO 6272-1)

Acrylic Polyol



Acrylic Polyol + PCL Diol

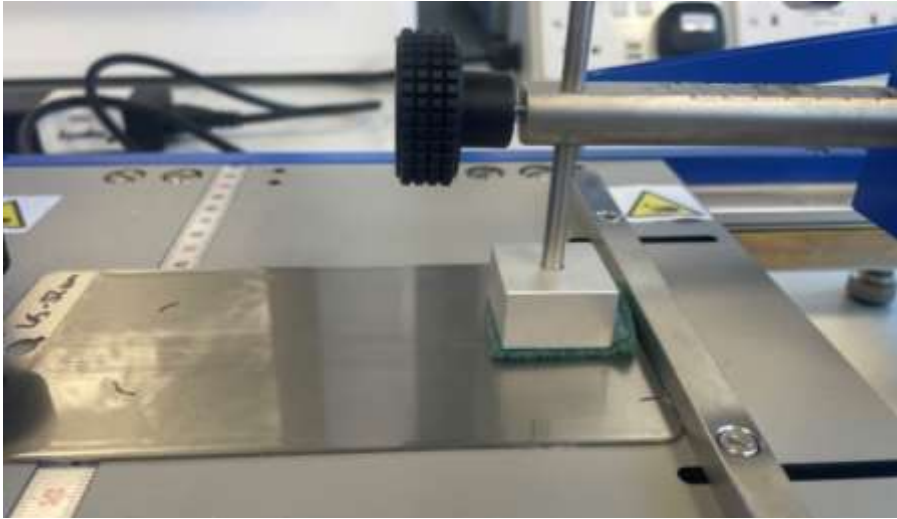


Acrylic Polyol + Polyester Adipate Diol

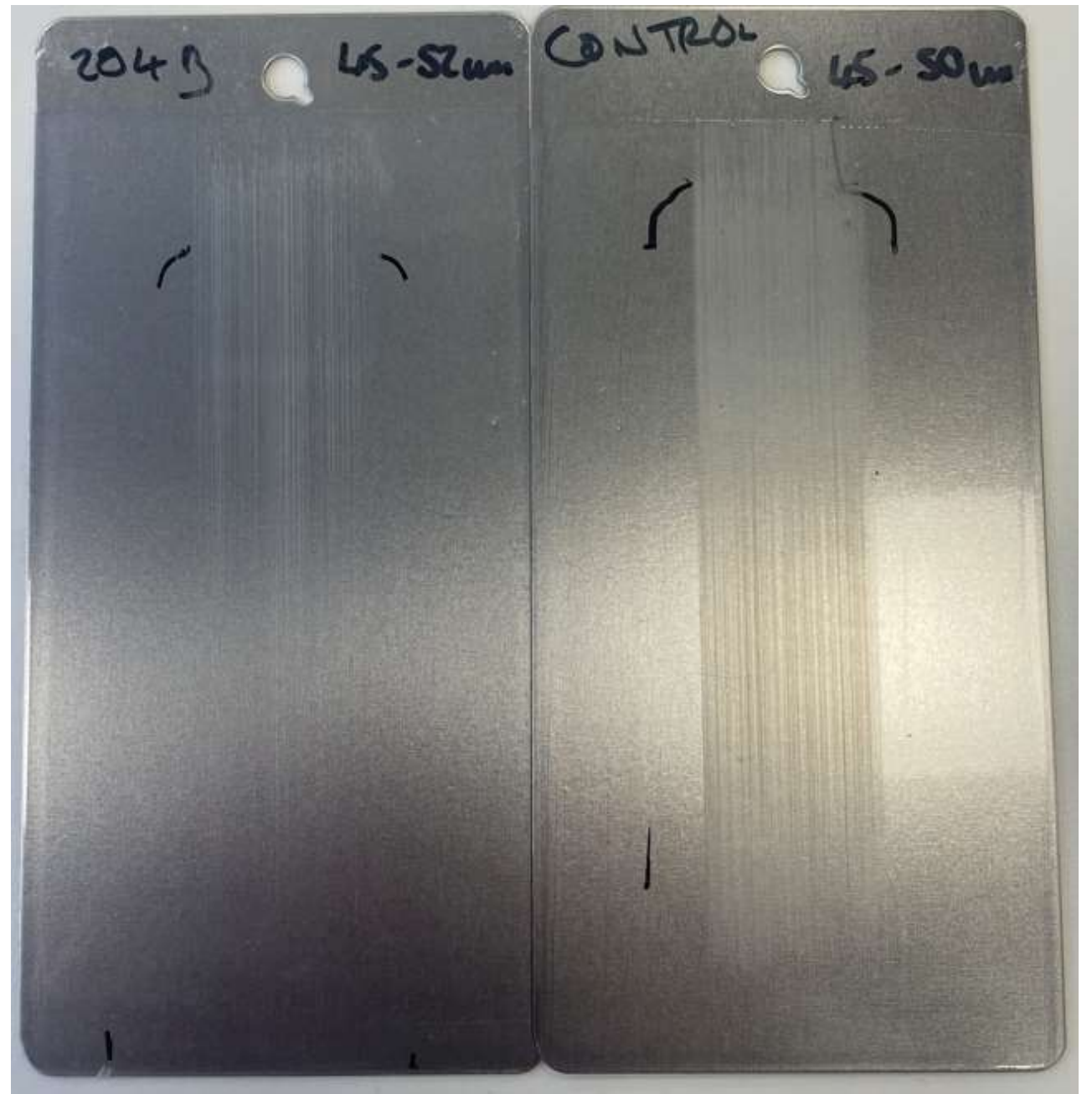


Enhancement with PCL polyols give improved indirect impact resistance

Mar Resistance



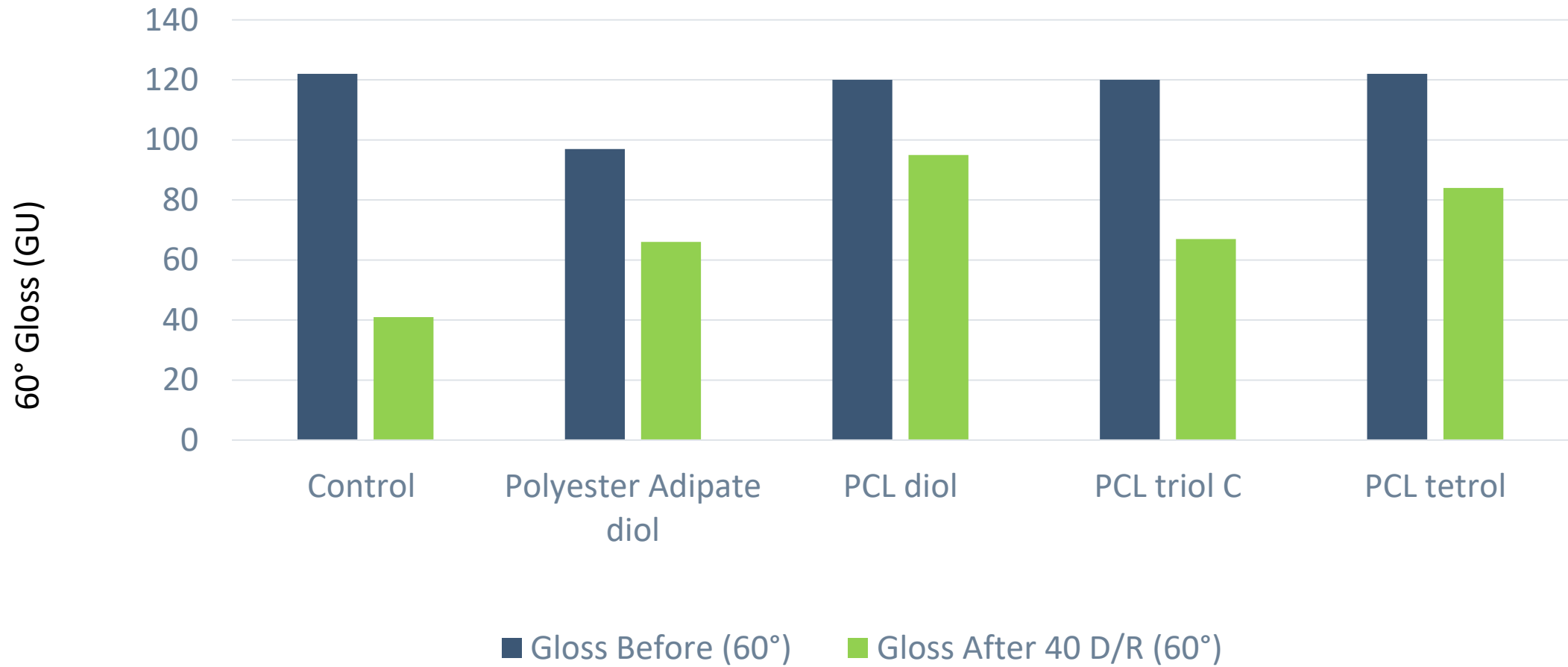
40 Double Rubs with a
Scrub Pad



Acrylic Polyol + PCL diol

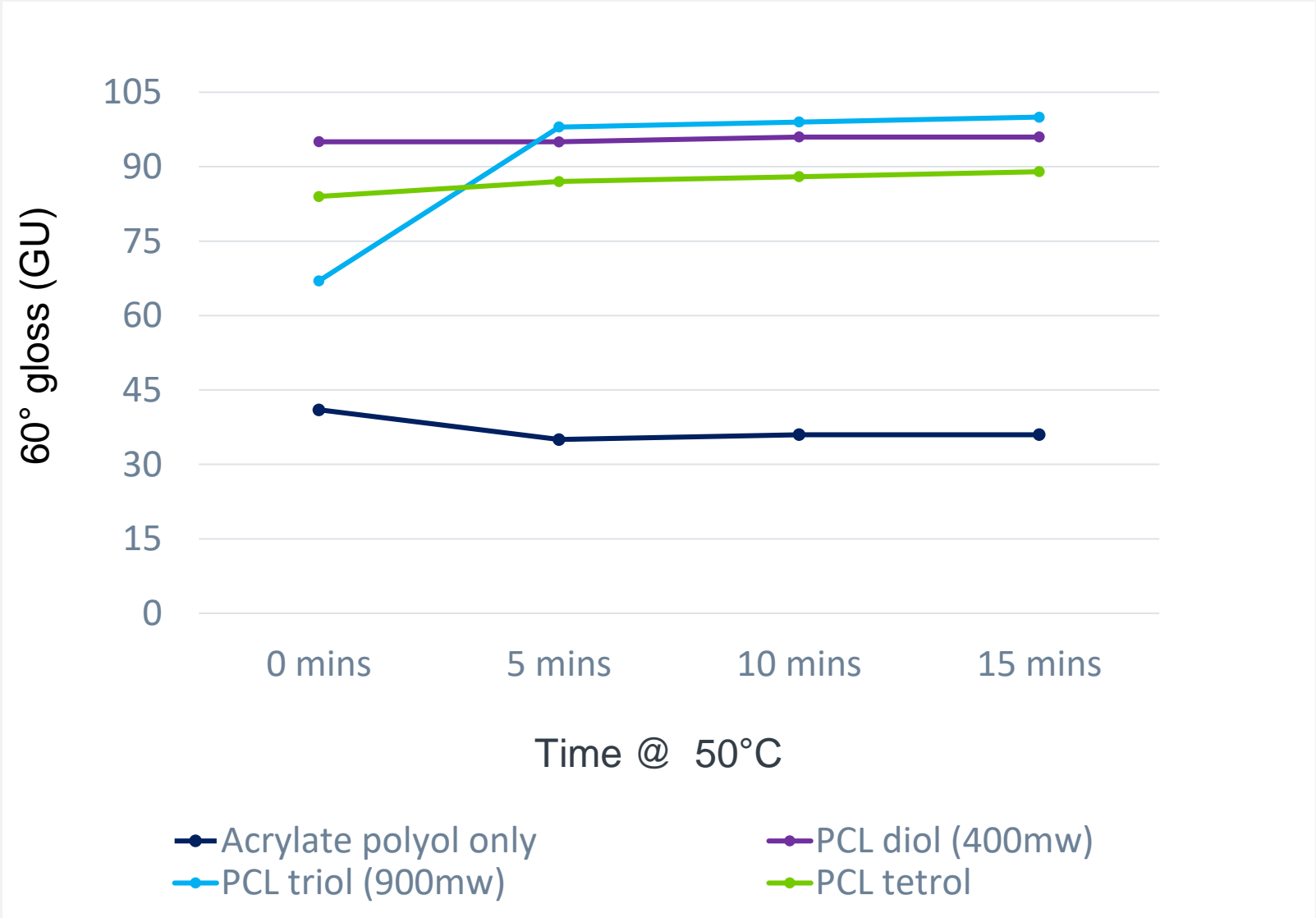
Acrylic Polyol

Mar Resistance – Pre & Post Rub




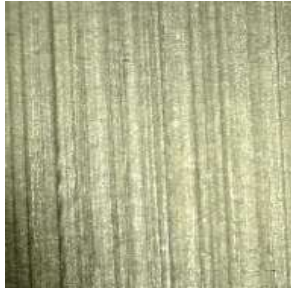
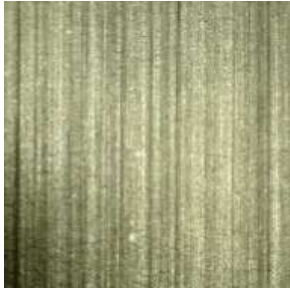



PCL diol showed the best resistance to gloss reduction

Gloss Recovery – Post Rub

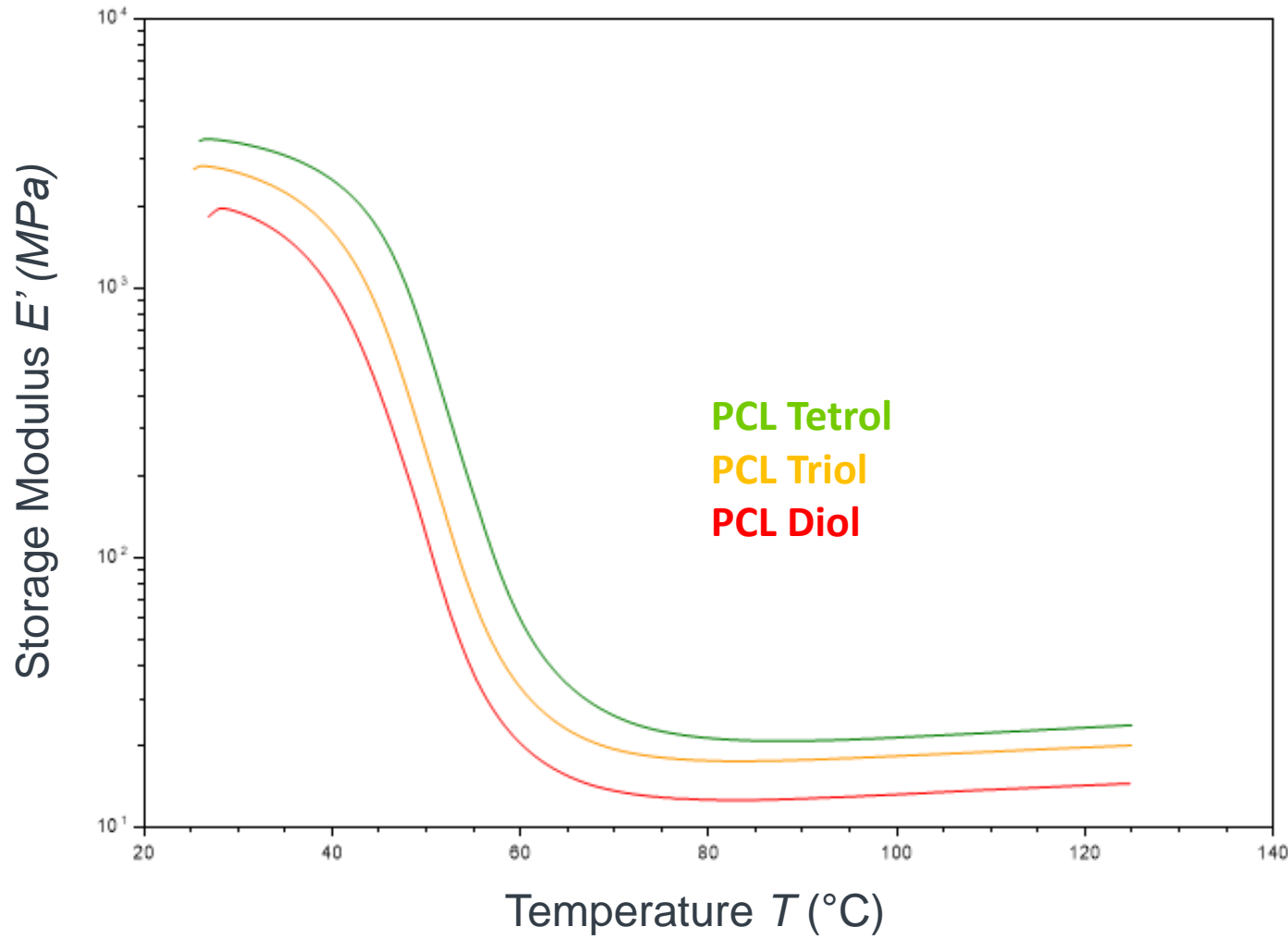


- PCL triol showed the largest gloss recovery
- PCL diol and PCL tetrol showed the best abrasion resistance
- Longer chains present in PCL triol help chain mobility

Intrinsic Self-Healing Properties

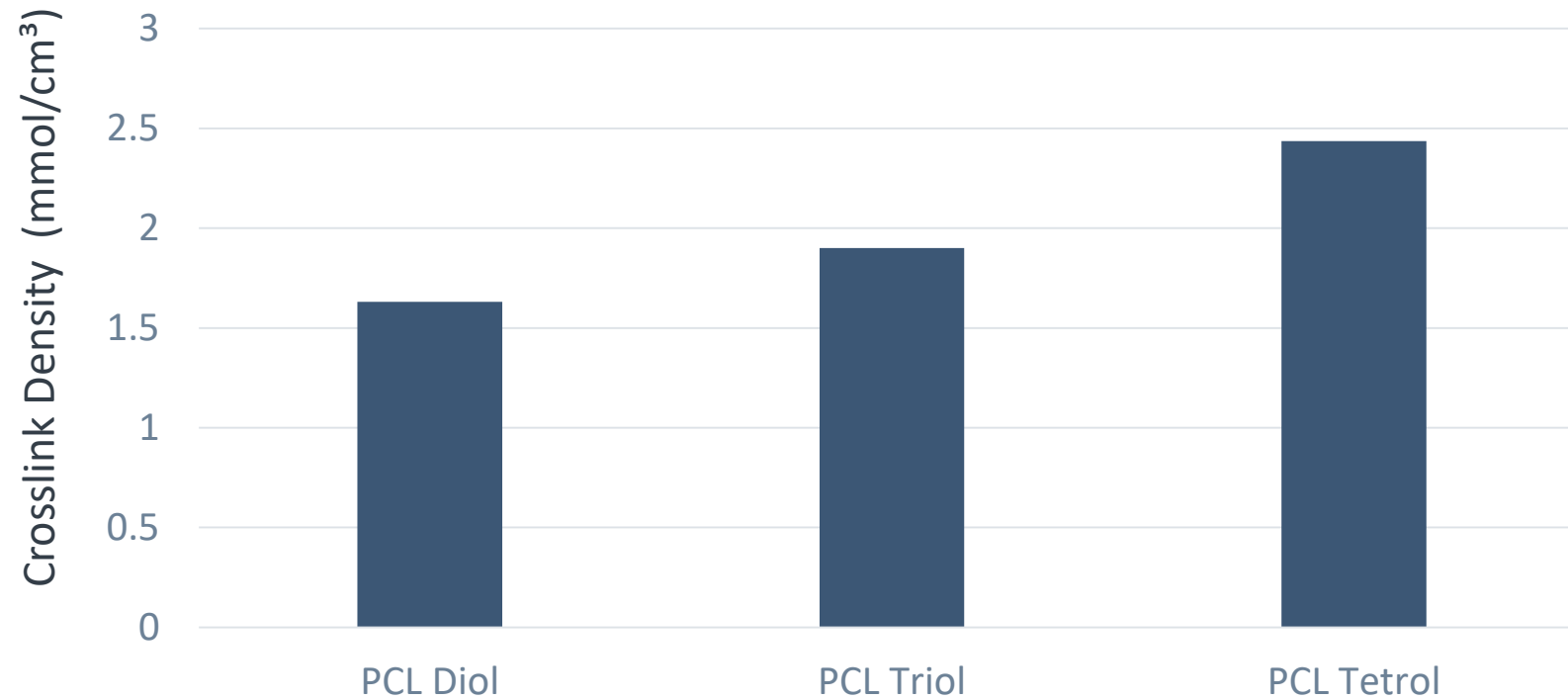
Polyol Used	40 Scotch-Brite® Double Rubs	5 mins @ 50°C	10 mins @ 50°C
Acrylic Polyol			
Acrylic polyol + PCL triol			

DMA (Dynamic Mechanical Analysis) – Crosslinking Density



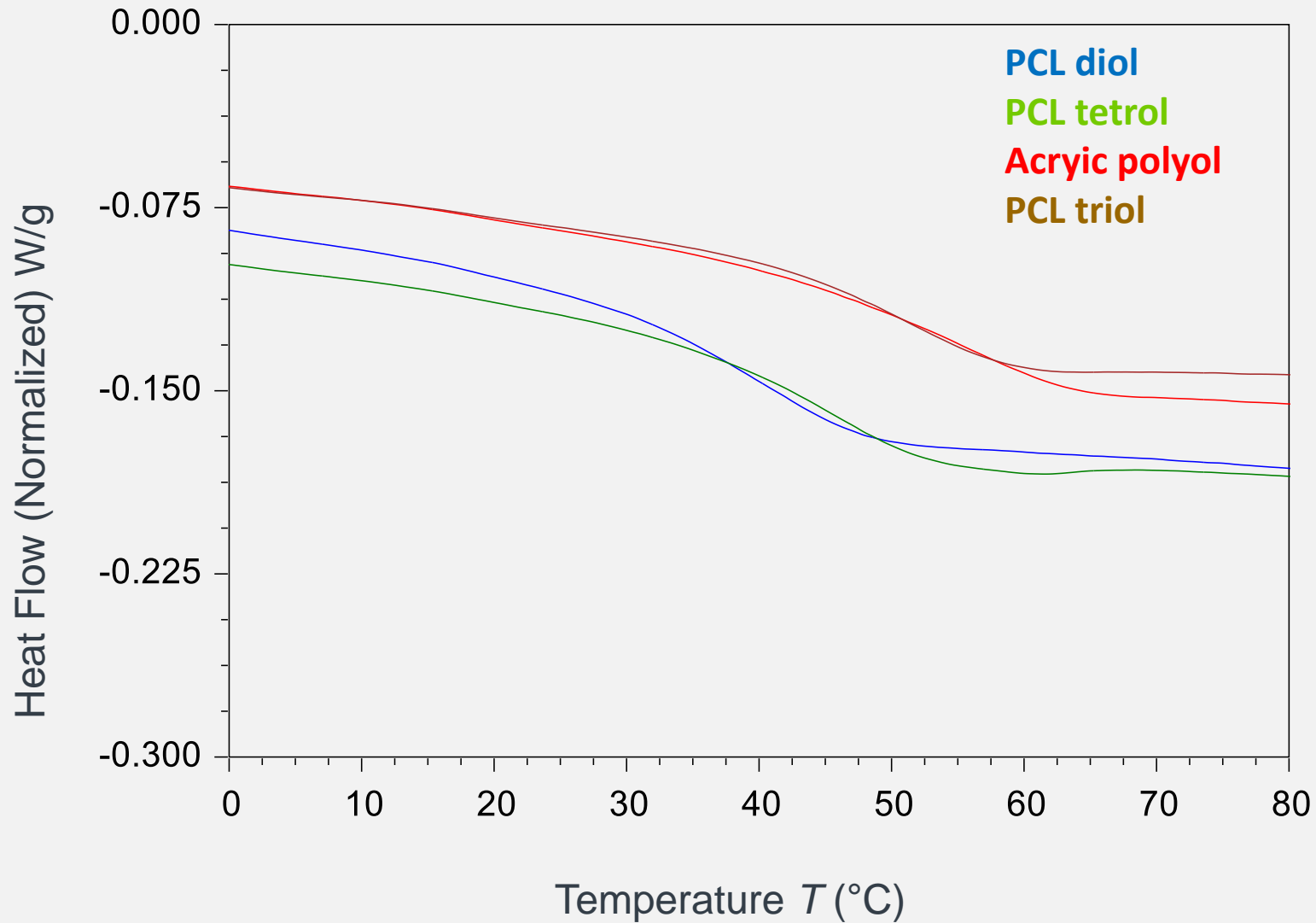
- The degree of crosslinking is related to the storage modulus (E') at the rubbery plateau region.
- The value of the rubbery plateau region can be closely related to the chain entanglement.

DMA (Dynamic Mechanical Analysis) – Crosslinking Density



PCL tetrol promotes better chemical resistance

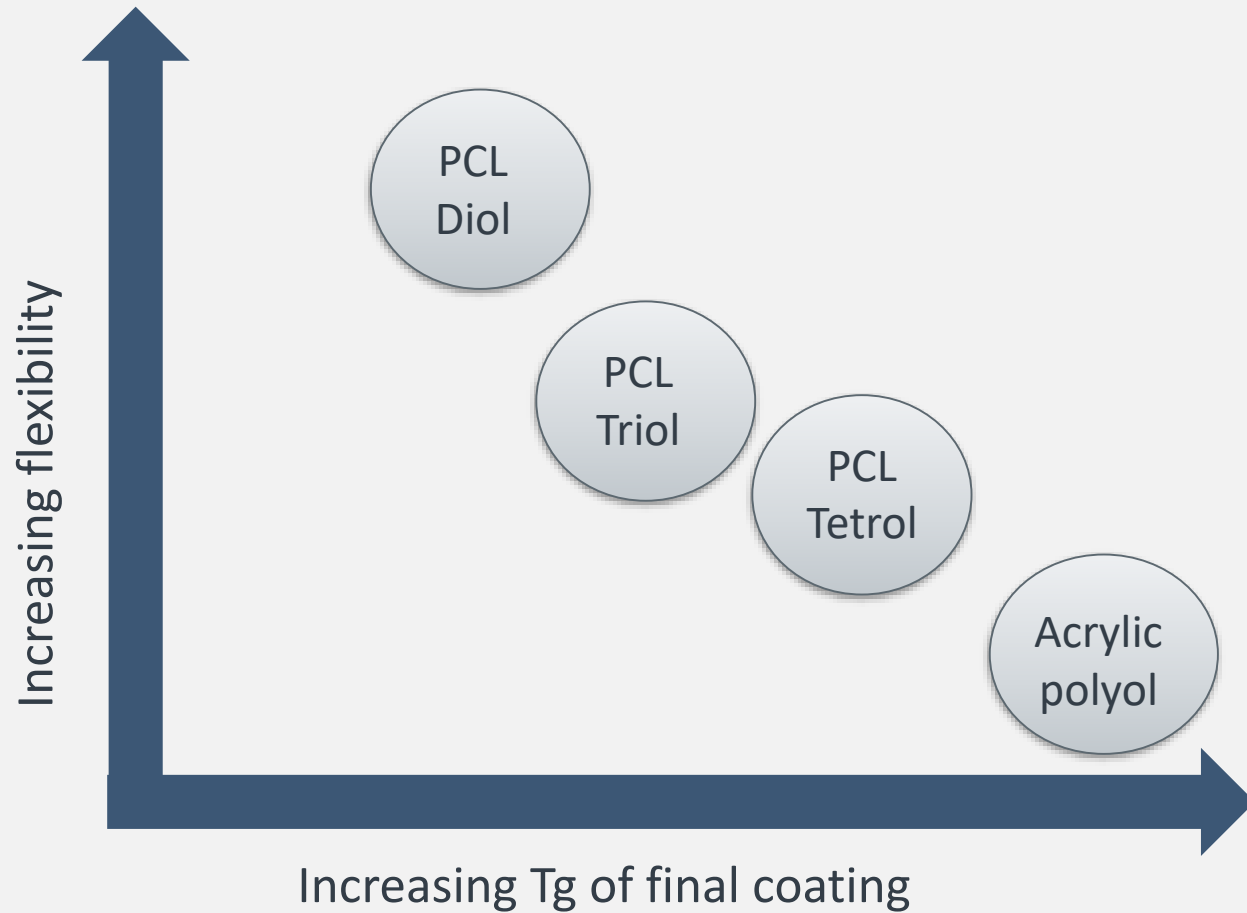
DSC Analysis (Differential Scanning Calorimetry)



Test Method:

1. Heat 10°C/min to 100°C
2. Cool 10°C/min to -50°C
3. Heat 10°C/min to 100°C
4. Cool 10°C/min to -50°C

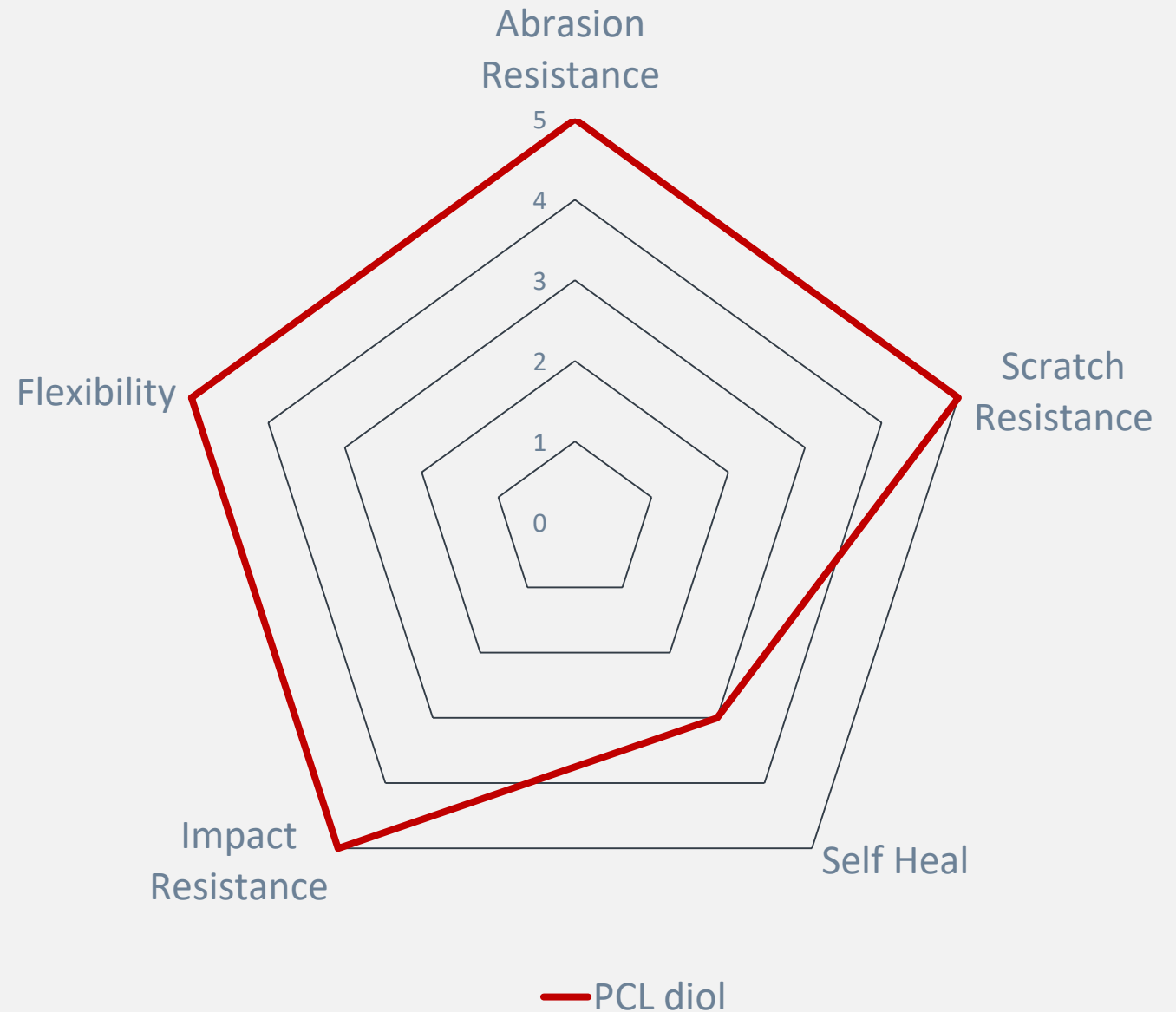
Tg Impact - Improved Low Temperature Performance



- **PCL diol** shows the largest reduction in T_g of the coating 2K coating system
- **A lower T_g** of the coating helps **durability** at **lower temperatures** and influences **crack bridging properties**
- **PCL tetrol** has a good balance of reducing the T_g while **incorporating cross-linking**

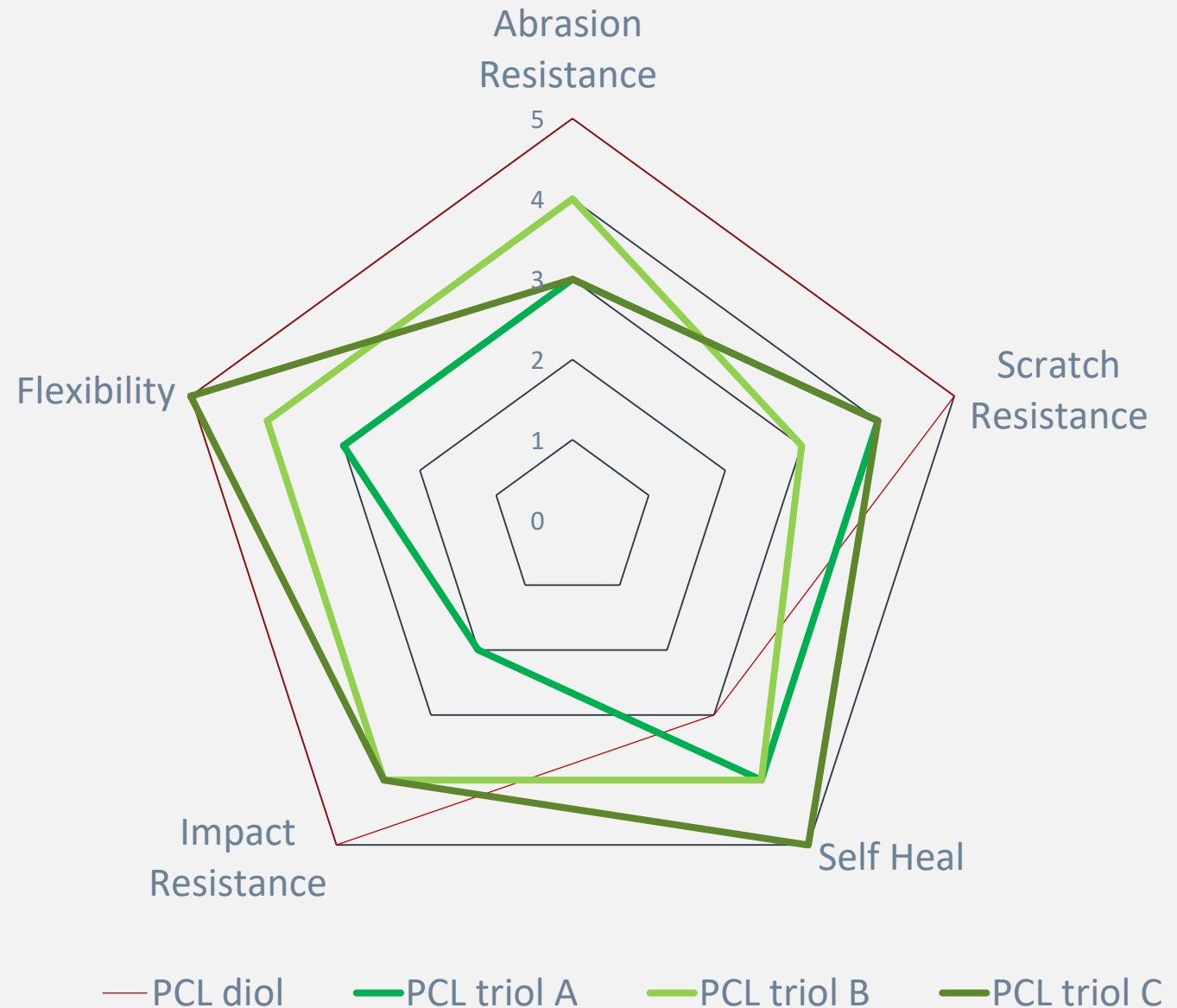
Polycaprolactone Polyol Properties – Summary

- **PCL diols** improve **impact resistance**, and can help improve **resistance to wear** mechanisms



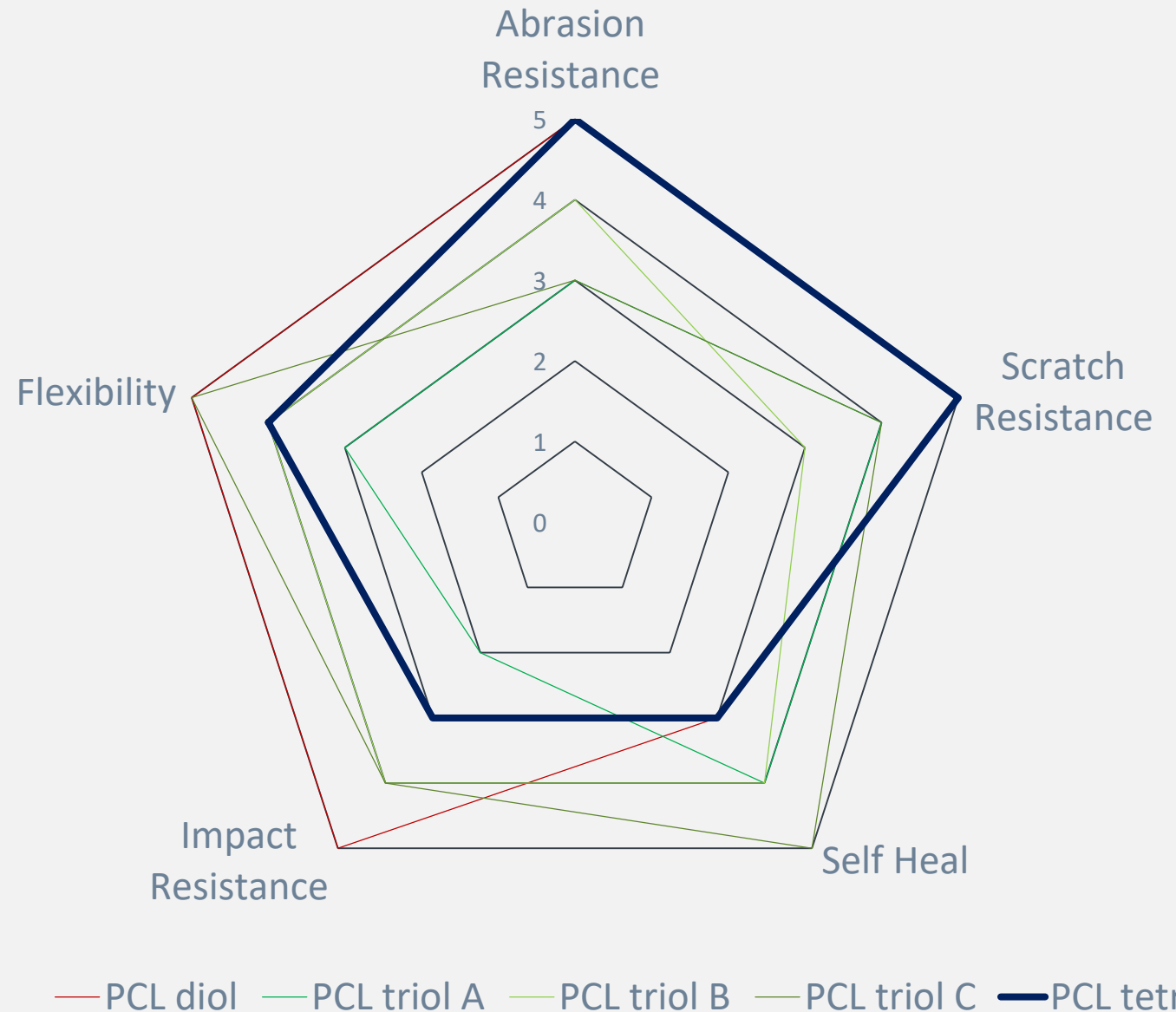
Polycaprolactone Polyol Properties – Summary

- **PCL diols** improve **impact resistance**, and can help improve **resistance to wear** mechanisms
- **PCL triols** will **improve crosslinking**, along with adding **flexibility** – this can help **adhesion**



Polycaprolactone Polyol Properties – Summary

- **PCL diols** improve **impact resistance**, and can help improve resistance to **wear** mechanisms
- **PCL triols** will **improve crosslinking**, along with adding **flexibility** – this can help **adhesion**
- **PCL tetrol** can help improve with its additional cross-linking but greatly improve **chemical resistance**



Acknowledgements

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Chuck Jones

Chuck.Jones@Ingevity.com

(215) 499-2356

Amanda Tosh

Amanda.Tosh@Ingevity.com

+44 (0) 1925987801

THANK YOU

