

Achieving Sustainability and Innovative Targets

with the latest Additive Technologies

CTT 2025



Core Values and Vision



Product Innovation

- Product Innovation – longevity, resource efficiency
- Innovative Business Models
- Closed-loop systems
- Sustainability



Safe Workplace

- Safe working environment
- Health, safety, and environment
- Employee protection



Manufacturing

- Resource Efficiency
- Waste Minimization
- Energy Efficiency
- Social Responsibility



Environment

- Sustainable Technologies
- Emission Reduction
- Environmental Initiatives
- Environmental Stewardship



Contributions to Sustainability



Biobased Raw materials



Cumene/Aromatic-Free



PTFE/ PFAs-Free



Biocide -Free



Label-Free



APEO -Free



Tin-Free



HAPs-Free



Solvent/ VOC -Free



Formaldehyde-Free

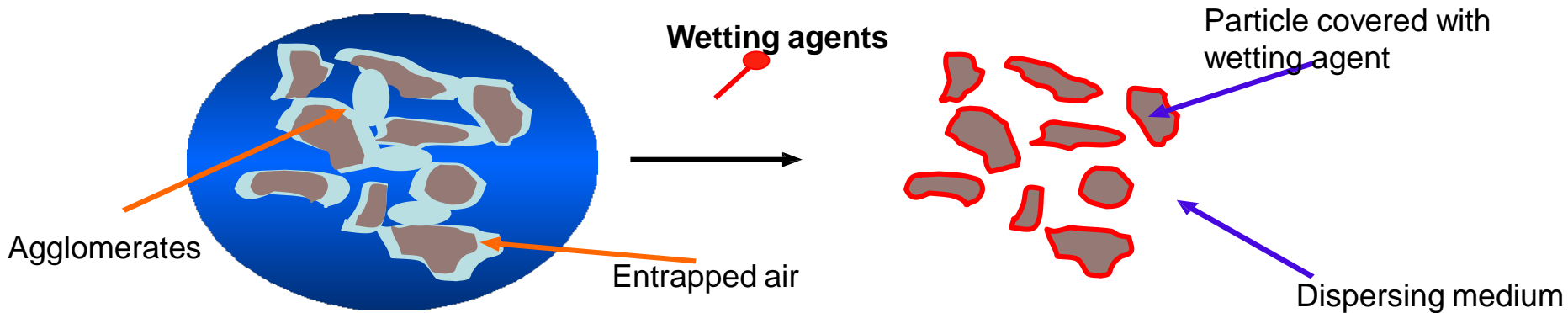




The Science behind Pigment Wetting and Dispersion Stabilizing Additives

Pigment Dispersion & Stabilization

1. Pigment wetting: displacement of air and moisture at the pigment surface by the liquid of the mill-base



Good wetting (adjusting Surface Tension liquid phase, through wetting agent):

- ✓ enabling high pigment solids
- ✓ low mill-base viscosity
- ✓ high milling efficiency



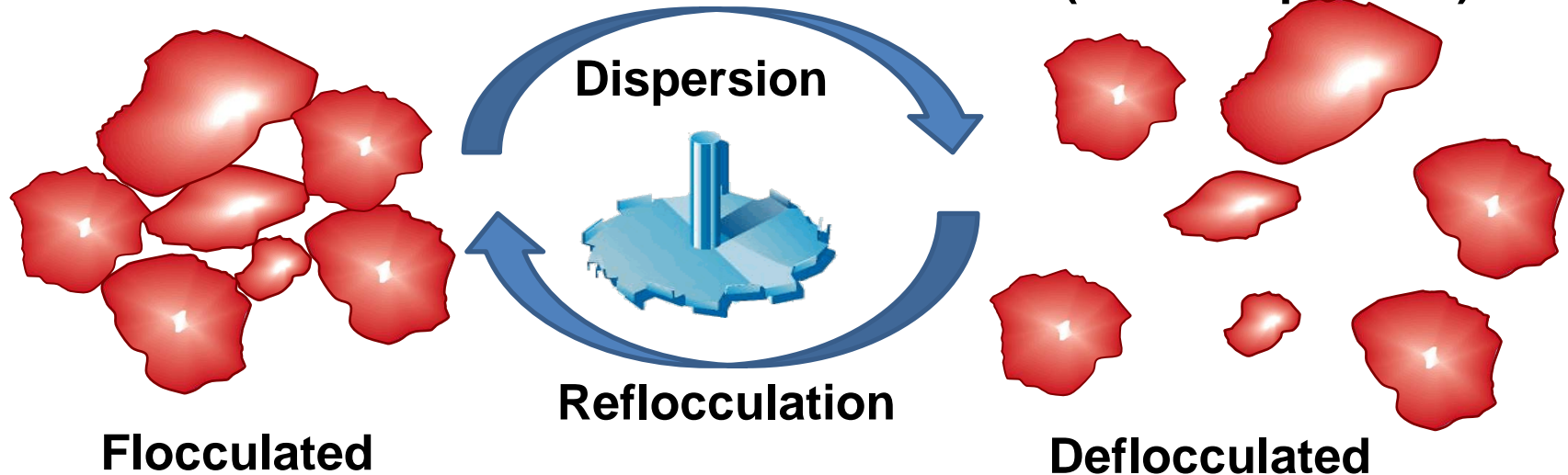
Pigment Dispersion & Stabilization

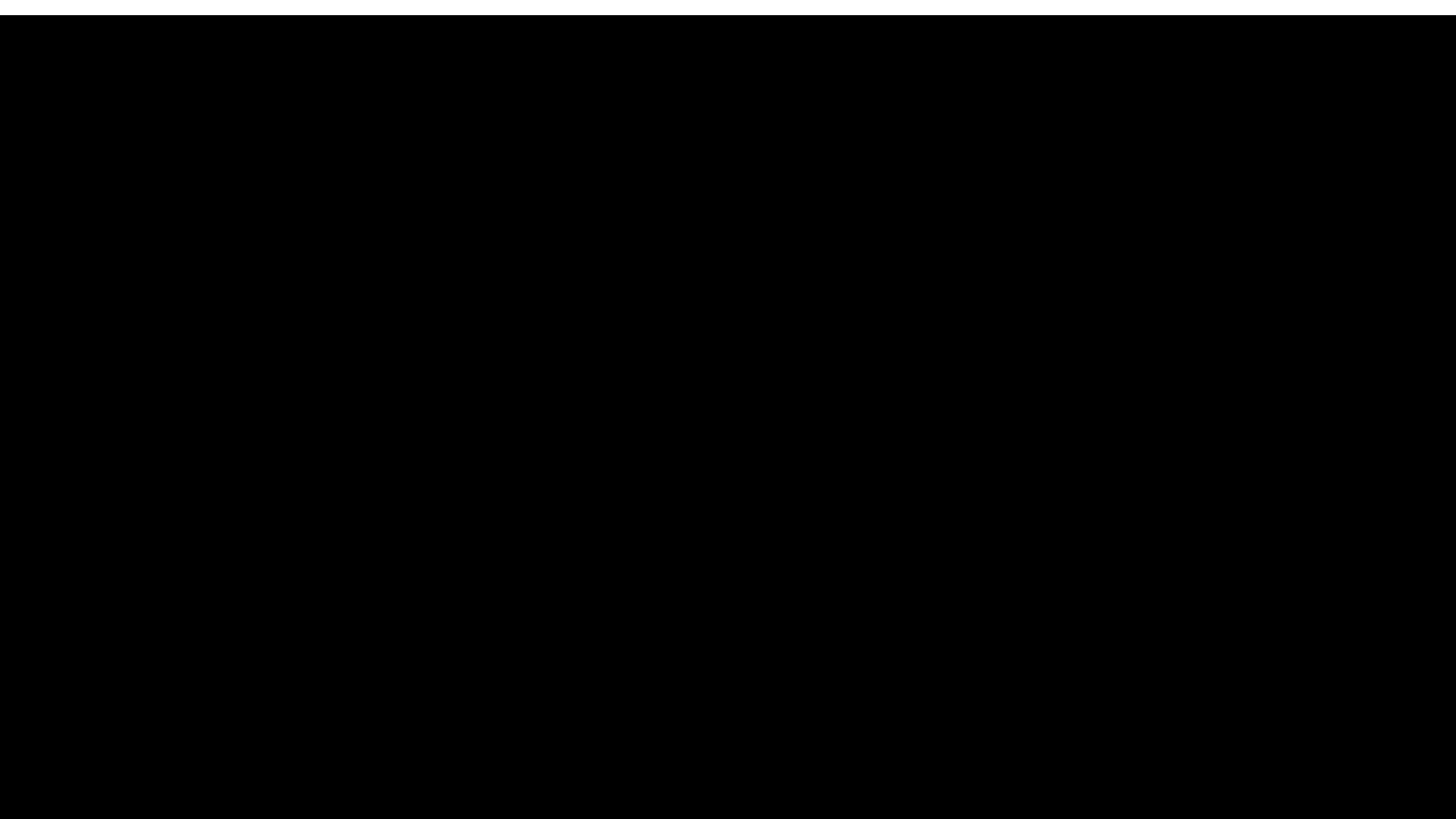
Pigment dispersion: the application of shear of sufficient energy into the millbase to separate pigment agglomerates into their individual unit particles

Wetting and Dispersing Additives

**Agglomerate
Flocculate**

**Primary Particle
(Ideal Dispersion)**







What's the Difference?

- Wetting additives bring the pigment surface into intimate contact with the continuous phase
 - displace air, brings pigment into the liquid phase
- Dispersants coat individual pigment particles to minimize particle to particle interactions providing observed lower viscosity
 - stabilize dispersed pigment, prevents flocculation

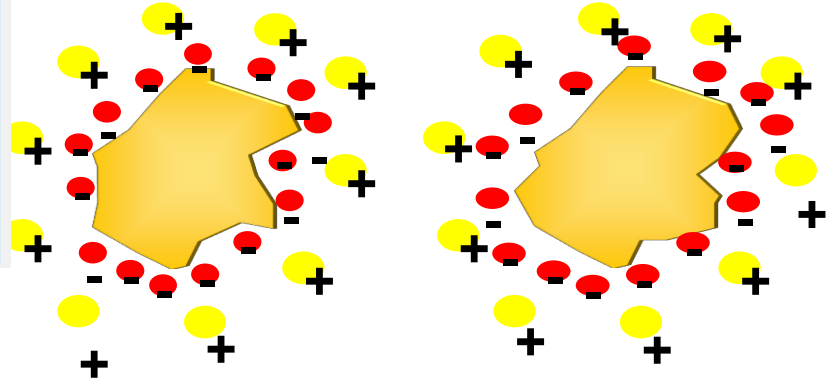
Pigment Stabilization

Pigment dispersion:

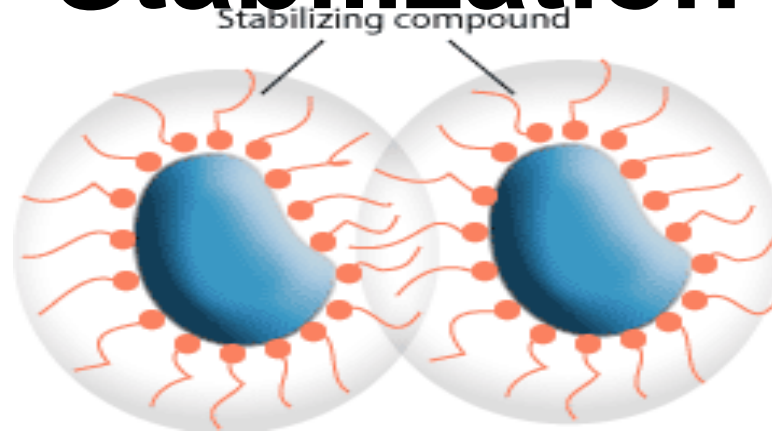
dispersed particles must be stabilized to prevent flocculation, agglomeration. Main stabilization mechanism for waterborne systems:

Electrostatic repulsion

- Adsorbed polyelectrolytes
- Stabilization efficiency increases with density of electrical double layer
- Additives used for dispersion in waterborne systems are high molecular weight products containing charged polymeric side chains



Steric or Entropic Stabilization



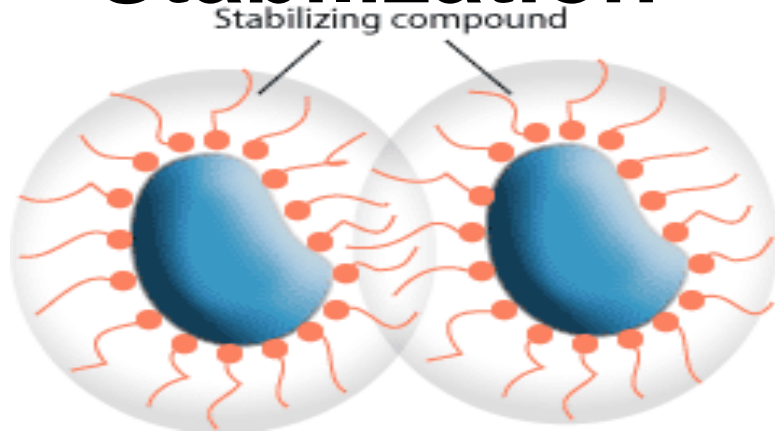
Main stabilization in solvent borne systems

Dispersant Requirements:

- Contains affinic groups providing strong adsorption on pigment surfaces
- Resin/solvent compatible chains directed into the surrounding vehicle.

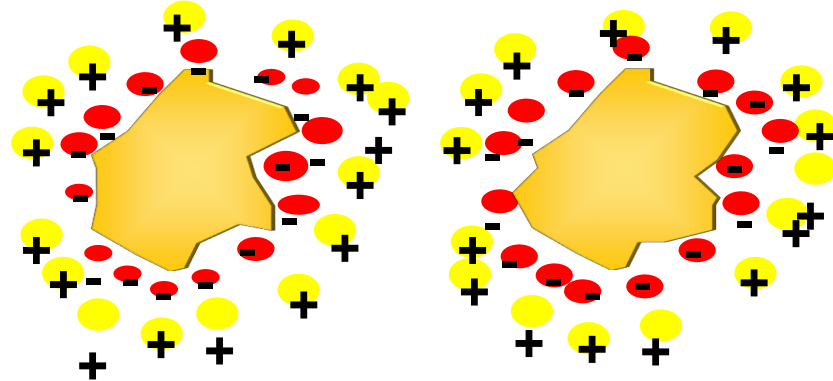


Steric or Entropic Stabilization



Main stabilization in solvent systems

Electrostatic Repulsion



Main stabilization in aqueous systems

Dispersant Requirements:

- Contains affinic groups providing strong adsorption on pigment surfaces
- Resin/solvent compatible chains directed into the surrounding vehicle.

Dispersant Requirements

Resin compatibility

Solubility

Pigment dispersion stability

Pigment affinic/anchoring groups

compatible with pigment surface chemistry

No degradation of important film properties

No reduction in application properties

Technology Challenge

- A. For **waterborne** dispersions
- a) Wetting hydrophobic pigments, extenders
 - b) Electrostatical stabilization, for all pigments
 - c) Additionally: transition to steric stabilization during dry;
the drying film shifts from aqueous to be more organic and hydrophobic due to loss of water and concentration of the resin emulsion particles

Technology Challenge

- A. For **apolar** solvent-borne dispersions
 - a) Wetting hydrophobic and hydrophilic pigments fillers, and extenders
 - b) Steric – for all pigments, as solvents evaporate causing the system to shrink and particulates to compact, stability must not be compromised by the loss of solvents

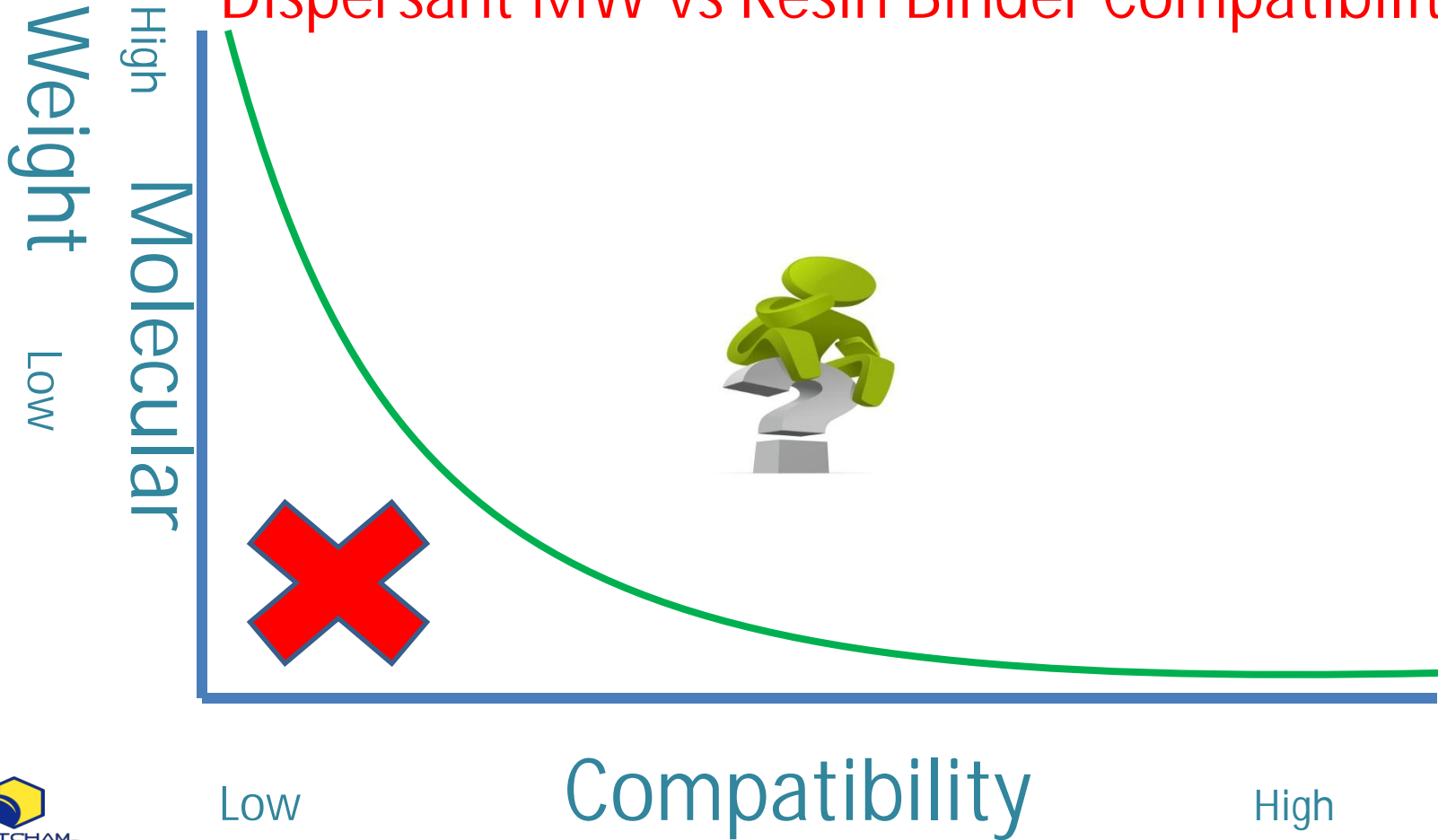
Remarks

Pigment Wetting is a relatively slow process
(Washburn Equation)

Rule 1) the higher the MW of the surfactant, the slower the wetting

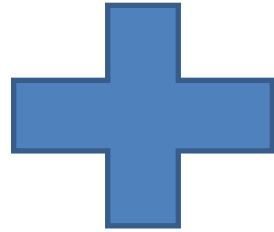
Rule 2) the higher the MW of the dispersant, the stronger the steric stabilization

Dispersant MW vs Resin Binder Compatibility



Patcham New Technology

Dispersion
Stability of
Higher
Molecular
Weight
Additives



Broad
Compatibility
of Lower
Molecular
Weight
Additives

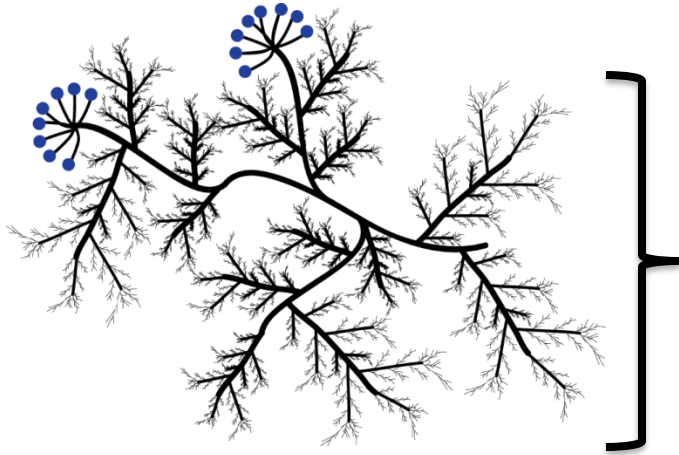


H
M
V

High Molar Volume



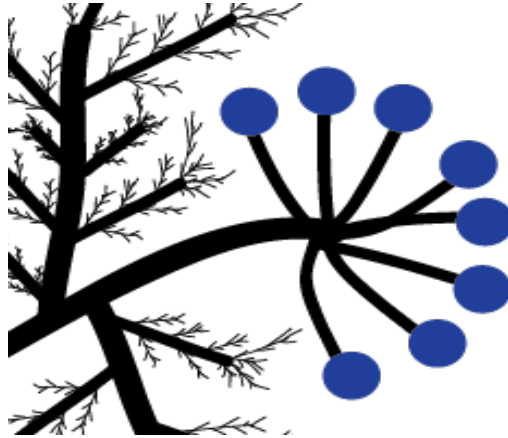
High Molar Volume Technology



Higher volume for a
given mass of
polymer

Branched polymeric
structures between
the pigment particles
creates greater steric
repulsion

High Molar Volume Technology

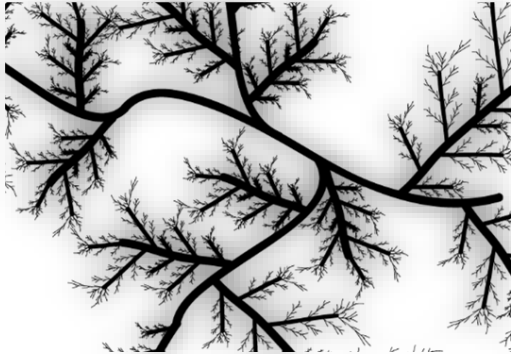


Multiple anchoring
groups

Faster wetting rate
and lower mill base
viscosity

Stronger Stabilization

High Molar Volume Technology

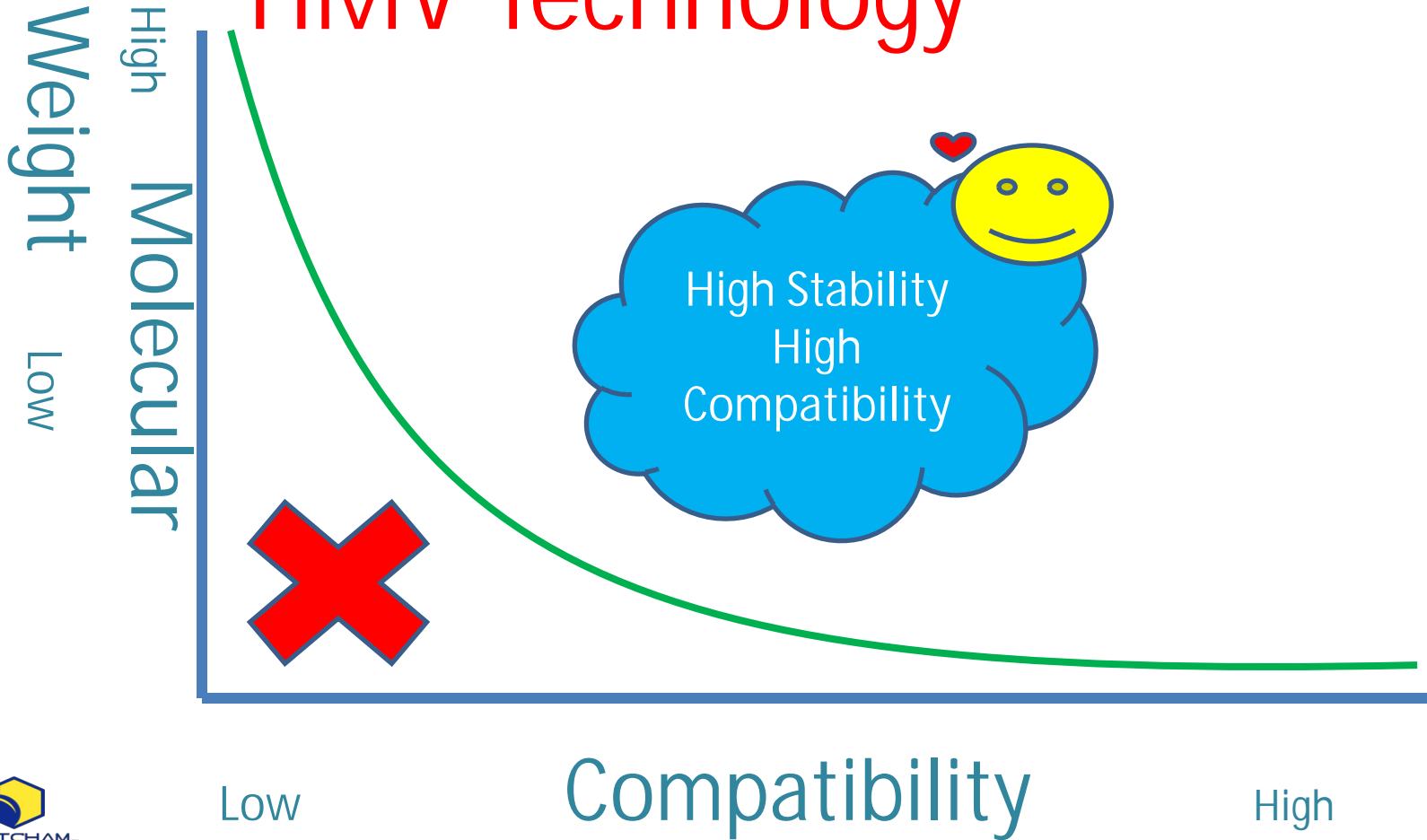


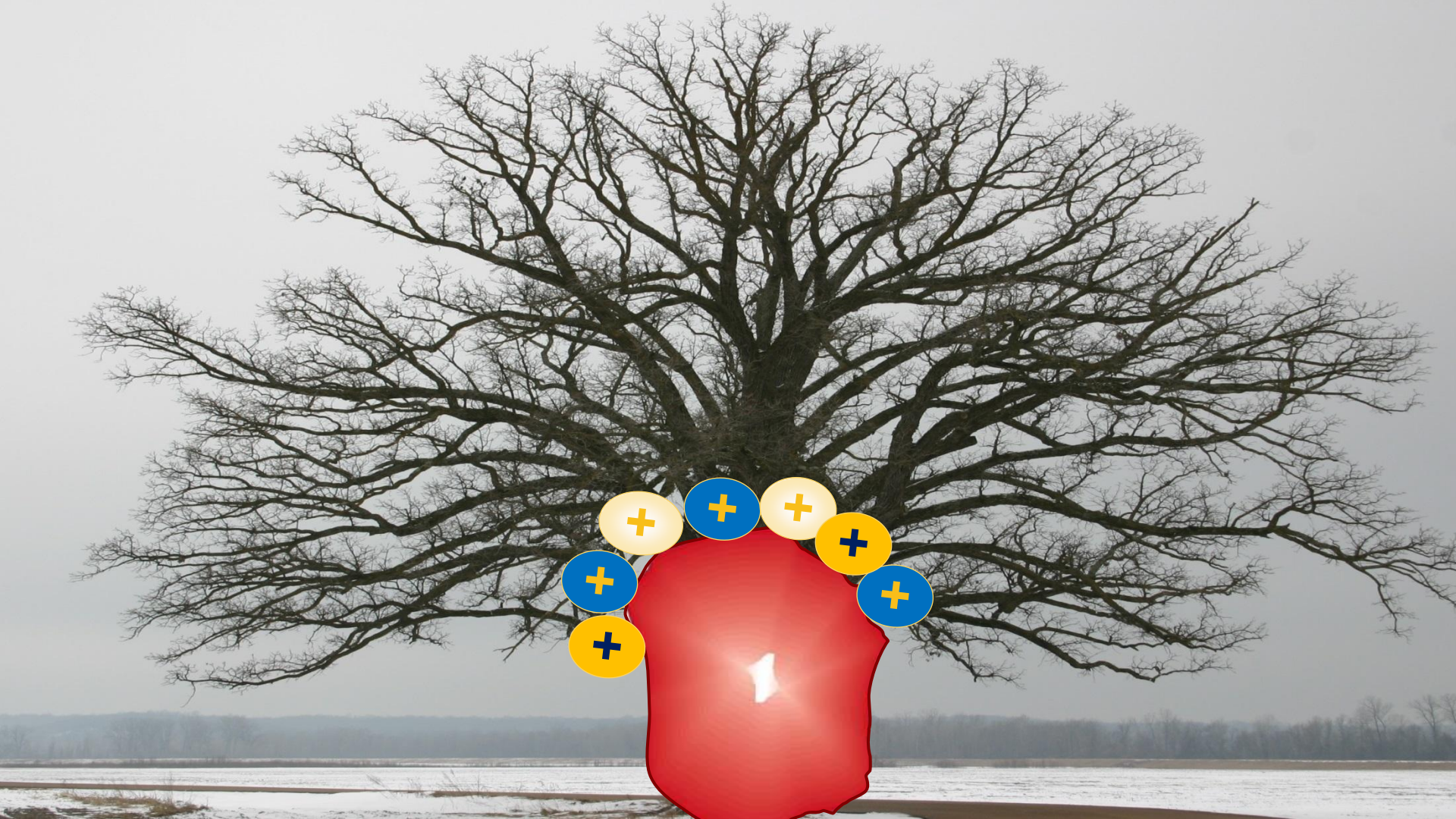
Lower MW Branched Polymeric Segments

Allows it to be used in a wider range of resin systems without incompatibility issues of similar MW linear or unbranched structures



HMV Technology

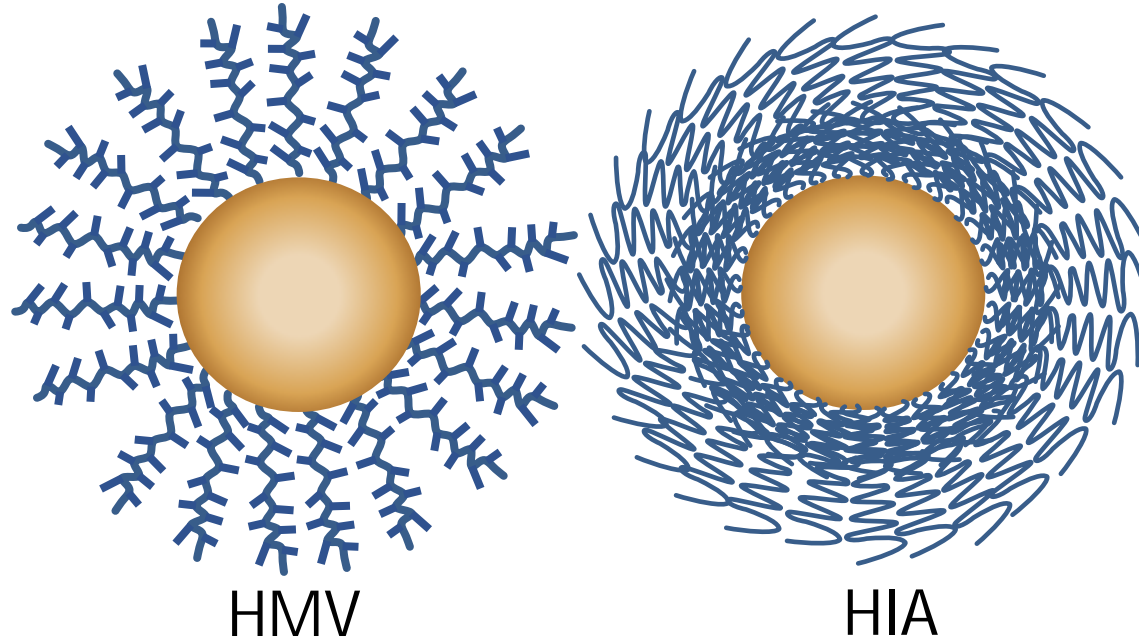




Advantages of HMTV Technology

- Faster Wetting/Dispersion > Productivity
- Optimize grinding times, better performance
- Increased stabilization improves stability
- Compatible in wide range of resins and solvents
- Development of resin minimal and resin free pigment concentrates easily achieved

HMV compared to HIA





Hyper Intermolecular Association

- Concept describing the potential of dispersant polymers for forming multi-layer molecular structures with neighboring components: non-adsorbed dispersant molecules, binders, including solvents
- HIA refers to the capacity of adsorbed dispersant to immobilize neighboring components, thus increasing the thickness of the adsorbed protective layer
- HIA is essential in increasing stabilization efficacy of low molar mass dispersants, as the individual molecule is too small for the formation of a stable protective layer



Novel HIA Polymeric Dispersant for Solvent borne Systems

Pat-Add DA 3054

| Compatible Resin Systems | |
|--------------------------|--|
| Resin Systems | Description |
| 2K PU | Acrylic Polyol |
| Epoxy | Bisphenol A type cured with Amine or Amide |
| TPA | Acrylic Resins |
| Acid Cure | Short Oil Alkyd and UF cured with PTSA |
| Alkyd-Melamine | Short Oil Alkyd with MMF |
| Polyester-Melamine | Saturated Polyester with MMF |
| Universal Grinding Resin | Aldehyde based |
| Long Oil Alkyd | Vegetable oil and fatty acid based |
| Medium Oil Alkyd | Vegetable oil and fatty acid based |
| Short Oil Alkyd | COFA-based |
| CAB Solution | CAB 381 |
| | CAB 531 |
| | CAB 551 |

| Solubility | |
|----------------------|---|
| Solvents | |
| MPA | ● |
| Xylene | ● |
| Toluene | ● |
| MIBK | ● |
| EEP | ● |
| Acetone | ● |
| Di-Acetone alcohol | ● |
| n-Butyl Acetate | ● |
| Ethyl glycol acetate | ● |
| IPA | ○ |
| N-Butanol | ○ |
| Solvesso 100 | ● |
| Solvesso 150 | ● |
| Oxsol 100 | ● |
| Propylene carbonate | ● |
| MT0 | ○ |

Soluble ● Insoluble ○ Partially soluble ◐



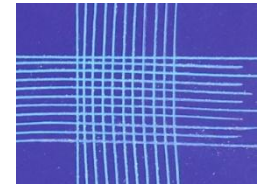
Novel HIA Polymeric Dispersant for Solvent borne Systems

Pat-Add DA 3054

| Direct Grinding Systems using PB 15:3 | | | |
|---------------------------------------|----------|----------|--------------------|
| | SB 2K PU | SB Epoxy | Polyester Melamine |
| Pigment Loading | 4.5% | | |
| %SOP of New Dispersing Agent | 35% | | |
| Paint Properties | | | |
| Fineness of Gauge | 7+ | 7+ | 7+ |
| Visosity, KU | 68 | 88 | 63 |
| Gloss @20° | 93 | 99 | 68 |
| Gloss @60° | 95 | 100 | 87 |
| Adhesion test | Pass | Pass | Pass |
| Conical Mandrel Bend test | Pass | Pass | Pass |
| MIBK Rub (100 rub) | Pass | Pass | Pass |
| MEK Rub (100 rub) | Pass | Pass | Pass |



Adhesion, 2mm crosshatch



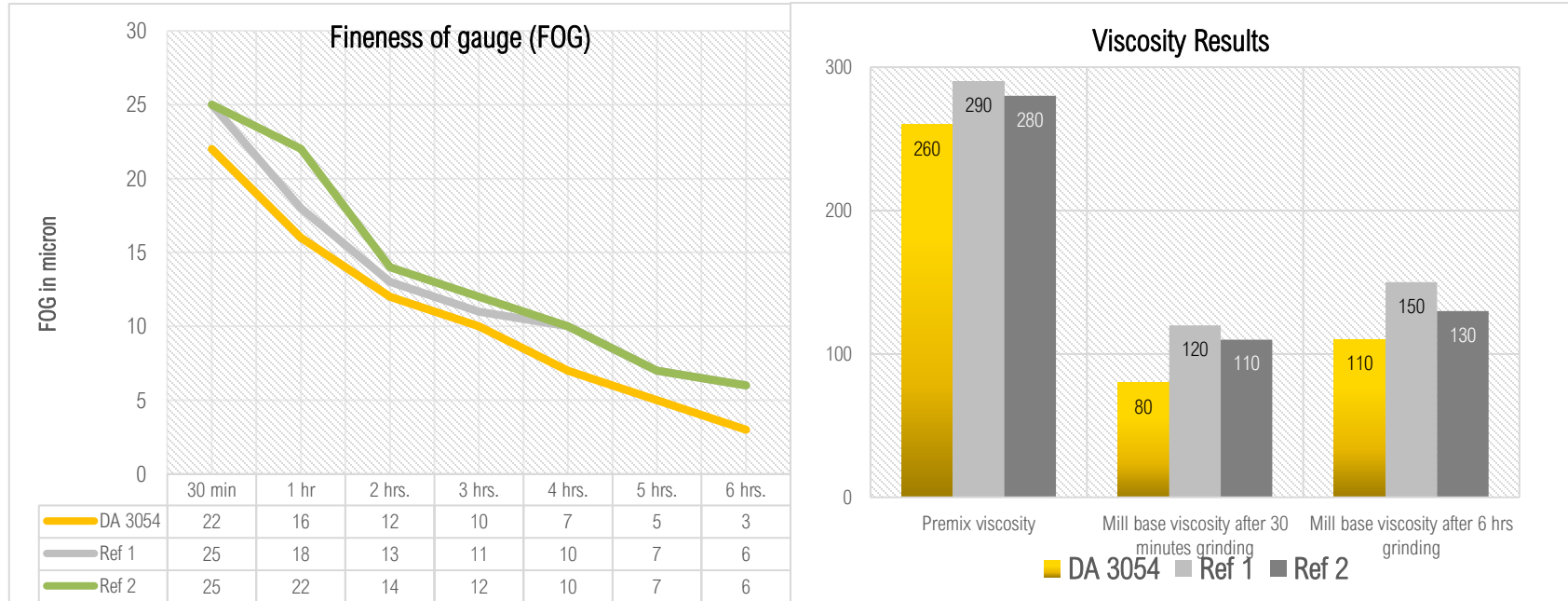
Conical Mandrel Bend test



Novel HIA Polymeric Dispersant for Solvent borne Systems

Pat-Add DA 3054

2K PU Topcoat Black

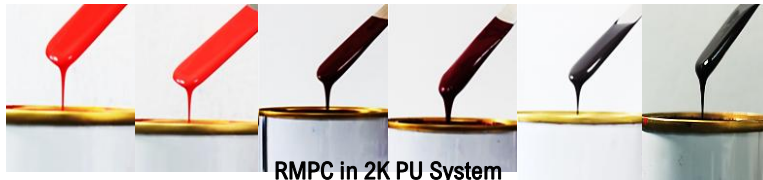


Novel HIA Polymeric Dispersant for Solvent borne Systems

Pat-Add DA 3054

Pigment Concentrates with Organic Pigments

| Sudaperm Red 2957 | | Hostaperm Red E5B02 | | Hostaperm RL spec | |
|-------------------|-----------------|---------------------|-----------------|-------------------|-----------------|
| PR 254 | | PV 19 | | PV 23 | |
| Initial | After Stability | Initial | After Stability | Initial | After Stability |



| | Sudaperm Red 2957 | | Hostaperm Red E5B02 | | Hostaperm RL spec | |
|-----------------------|-------------------|-----------------|---------------------|-----------------|-------------------|-----------------|
| | PR 254 | | PV 19 | | PV 23 | |
| Properties | Initial | After Stability | Initial | After Stability | Initial | After Stability |
| Viscosity, KU | 65 | 66 | 66 | 83 | 75 | 77 |
| Masstone | | | | | | |
| L* | 41.34 | 41.50 | 35.78 | 36.02 | 27.03 | 27.30 |
| a* | 45.55 | 45.96 | 34.45 | 34.80 | 2.00 | -3.09 |
| b* | 25.00 | 25.68 | 12.56 | 12.66 | -1.94 | -2.60 |
| Gloss @ 20 | 98 | 91 | 92 | 91 | 89 | 84 |
| Gloss @ 60 | 101 | 95 | 95 | 95 | 94 | 92 |
| dE, Rub out Tint Tone | 0.54 | 0.29 | 0.68 | 0.43 | 0.80 | 0.90 |





Latest Dispersant Technology

Opportunities and Applications

Solventborne Systems:

Automotive Coatings

Wood Coatings

Industrial Coatings

Protective and Marine Coatings

Coil Coatings

Inks

SB Pigment Dispersions





Now,
Commercially
Available
Dispersant
Technologies
for

- Solvent Borne Systems
- Water Borne Systems
- Solvent Free Systems



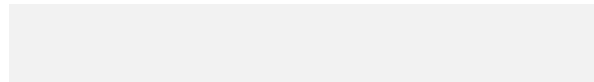
- These technologies can each be used to make a complete Colorant System
- Can be used to make co grinds in traditional paint manufacturing methods
- Can be used with traditional grinding resins

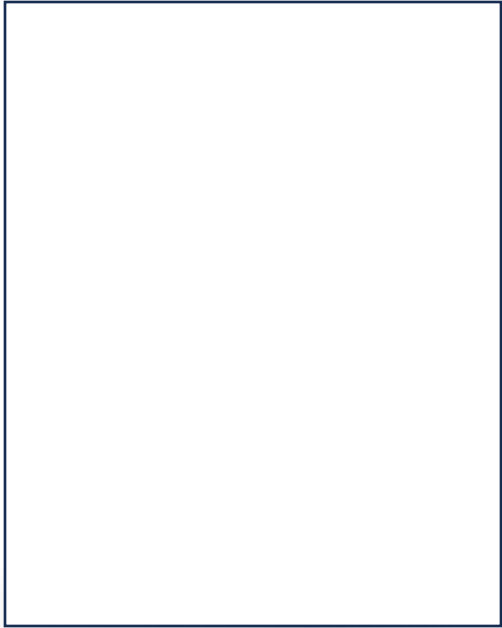
Dispersant Selection

- Determine solubility
- Determine compatibility
- Establish surface chemistry suitability
- Optimize order of addition (laboratory vs production)
- Confirm pigment dispersion stability and compatibility
 - From liquid to solidification following application
- Optimize use level
- Evaluate color stability
- Test film properties

Acknowledgement of Headquarters Development Team

- Bhavesh Patel
- Johan Bieleman





Thank You!





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